

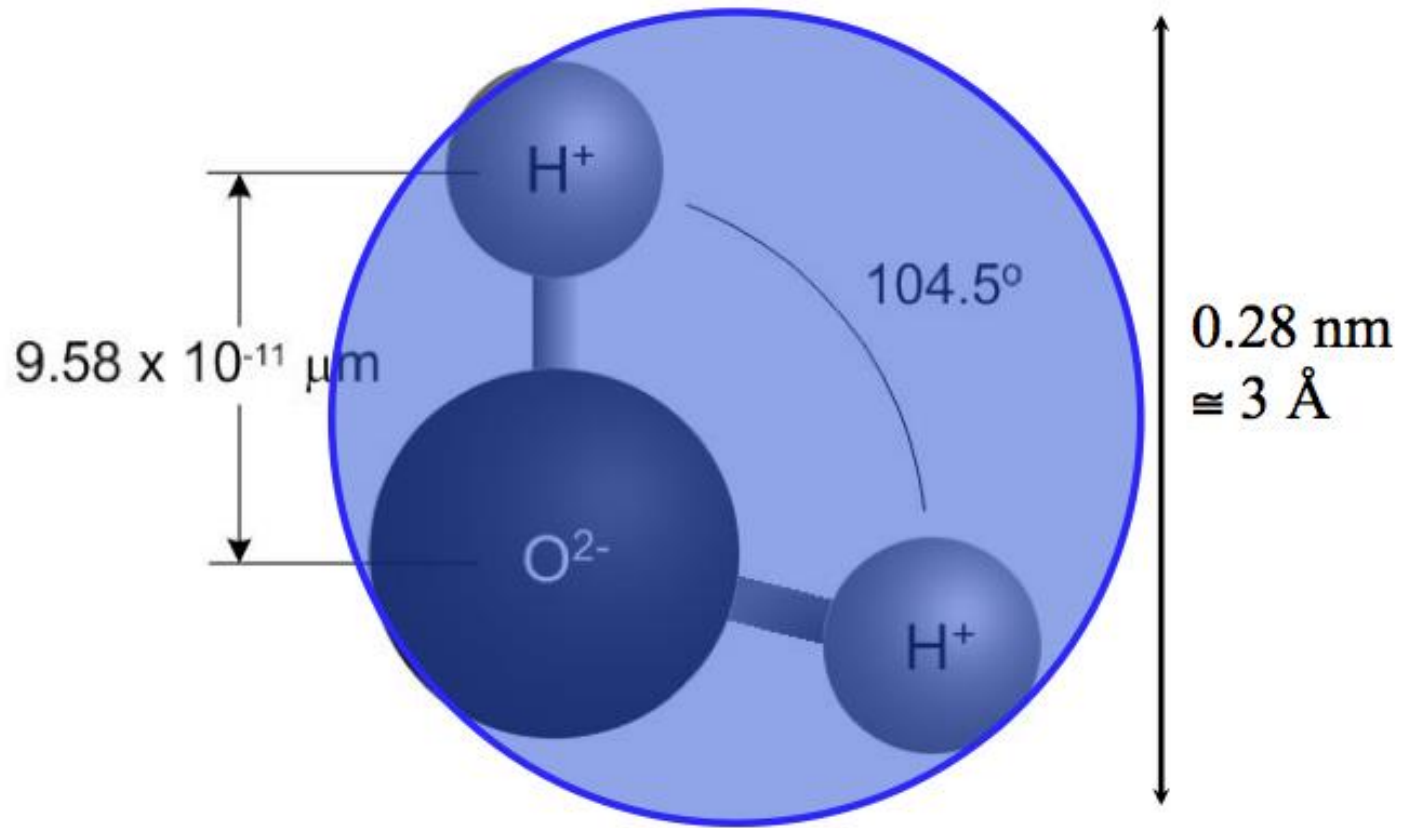
Joseph Lstiburek, Ph.D., P.Eng, ASHRAE Fellow

# Building Science

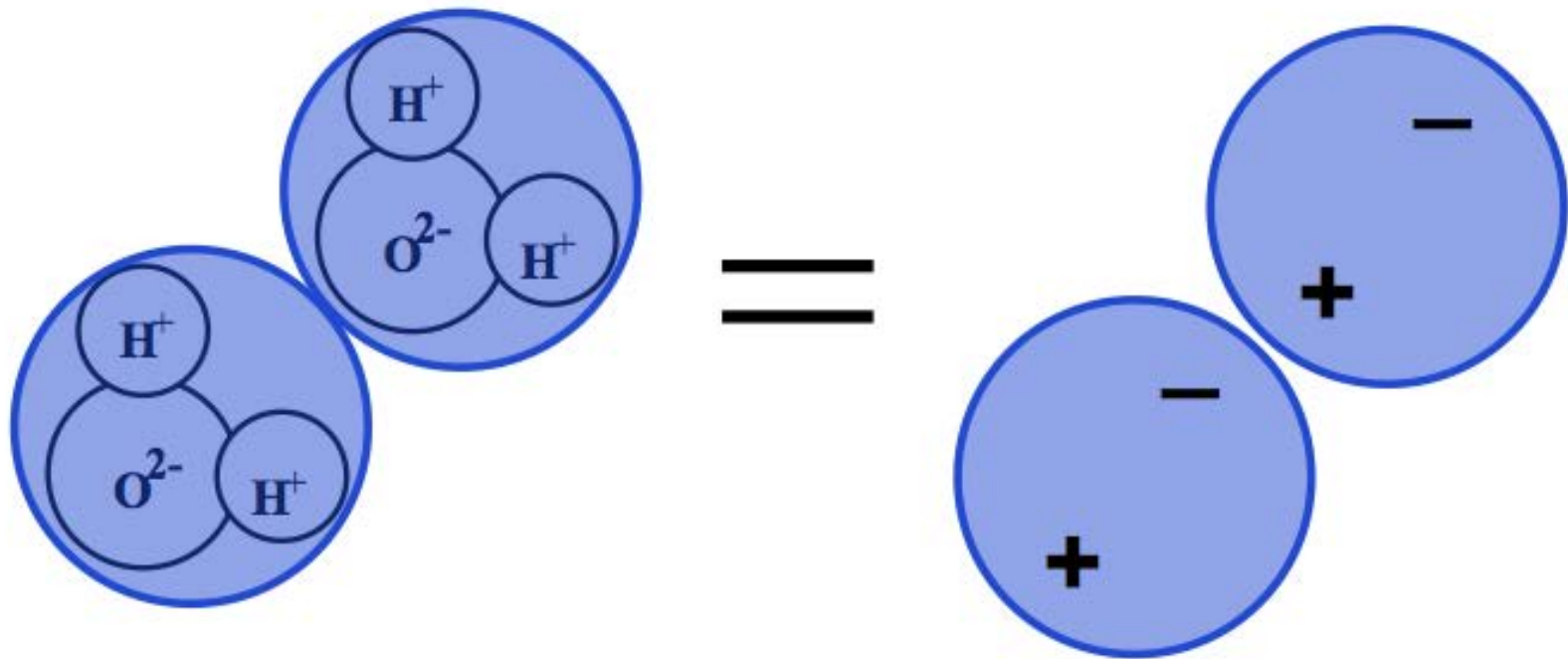
---

## The Water Molecule

# Water Molecules



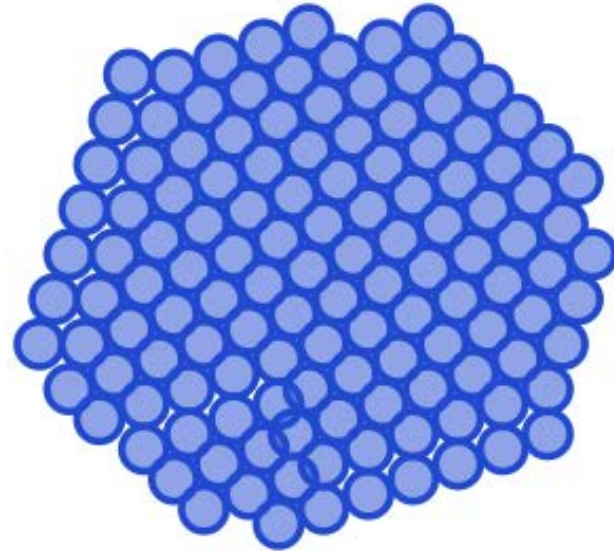
# Polar Molecule



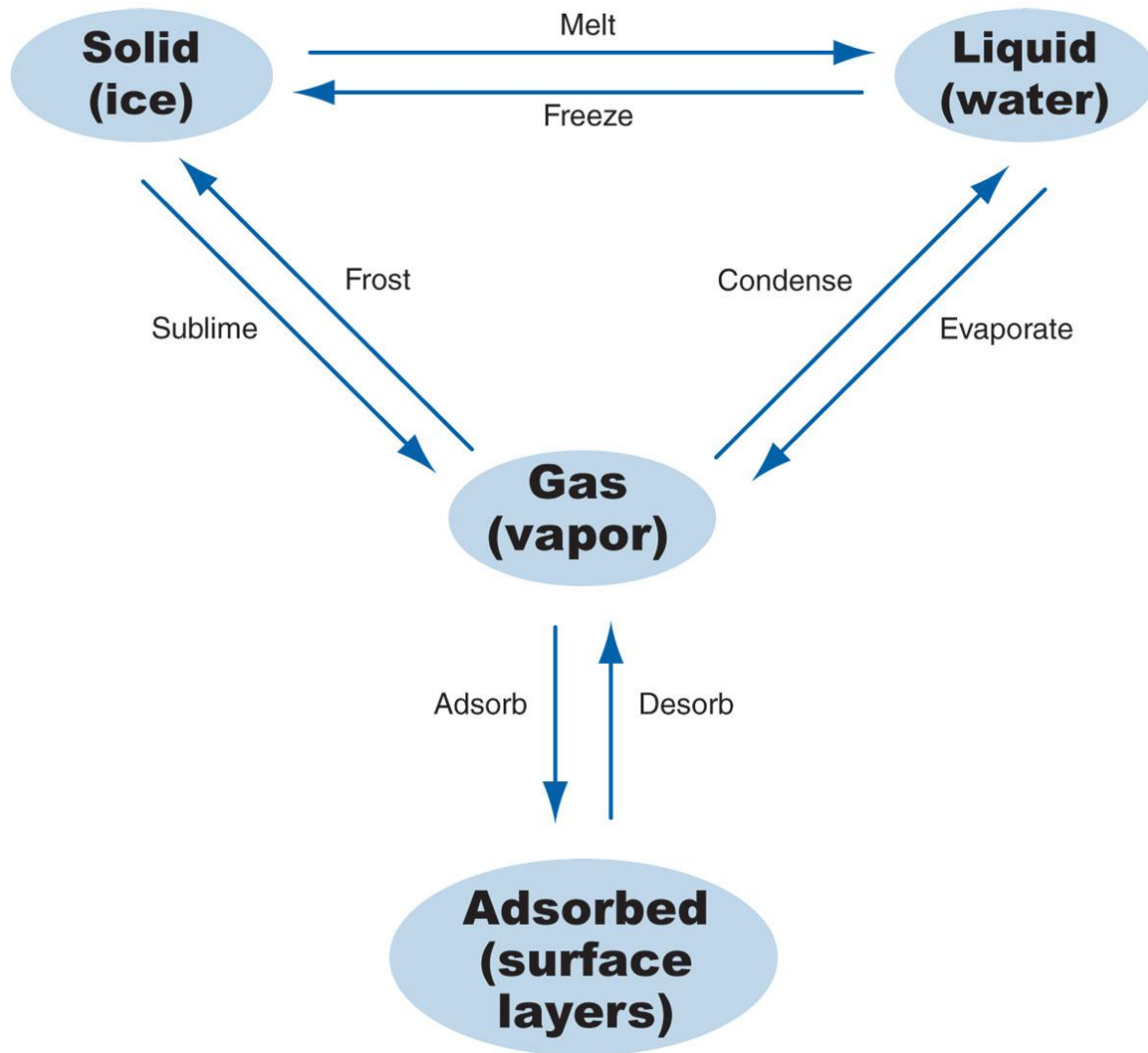
# Size Matters



**Vapor**



**Liquid**





## Moisture Transport in Porous Media

Phase	Transport Process	Driving Potential
Vapor	Diffusion	Vapor Concentration
Adsorbate	Surface Diffusion	Concentration
Liquid	Capillary Flow	Suction Pressure
	Osmosis	Solute Concentration

## Moisture Transport in Assemblies

Phase	Transport Process	Driving Potential
<b>Vapor</b>	Diffusion	Vapor Concentration
	Convective Flow	Air Pressure
-----		
<b>Adsorbate</b>	Surface Diffusion	Concentration
-----		
<b>Liquid</b>	Capillary Flow	Suction Pressure
	Osmosis	Solute Concentration
	Gravitational Flow	Height
	Surface Tension	Surface Energy
	Momentum	Kinetic Energy
	Convective Flow	Air Pressure

Vapor

Diffusion

Convective Flow

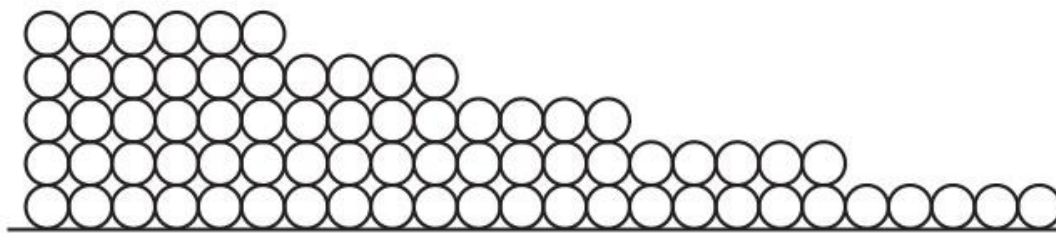
Vapor Concentration

Air Pressure

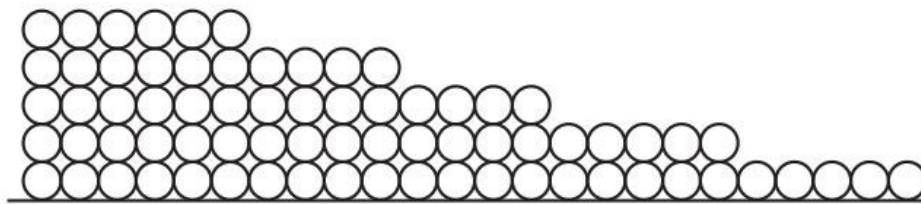
Adsorbate

Surface Diffusion

Concentration



Monolayers of adsorbed water increase with increasing RH



Monolayers  
flow along surface  
following concentration gradient



Vapor

Diffusion

Convective Flow

Vapor Concentration

Air Pressure

Adsorbate

Surface Diffusion

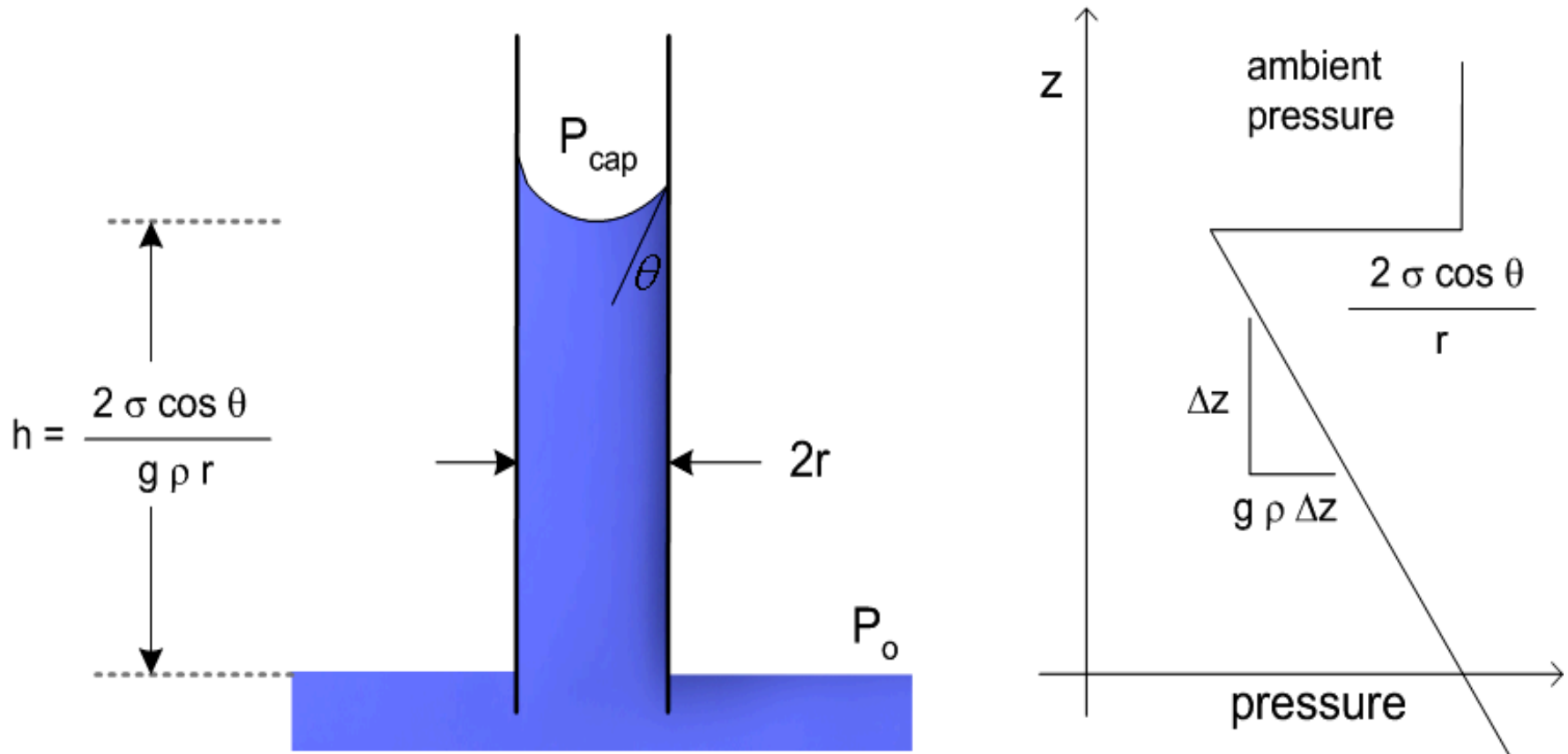
Concentration

Liquid

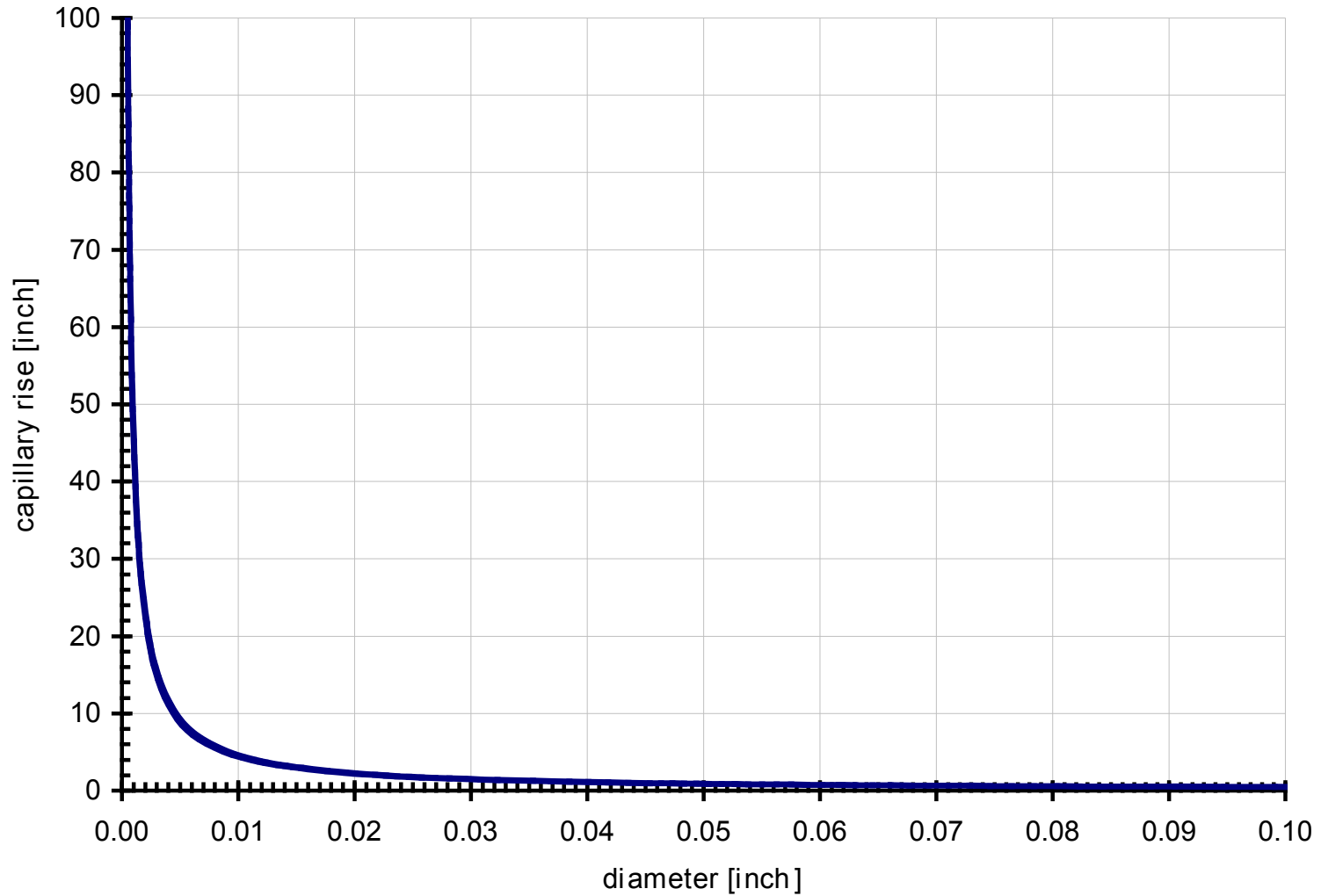
Capillary Flow

Suction Pressure

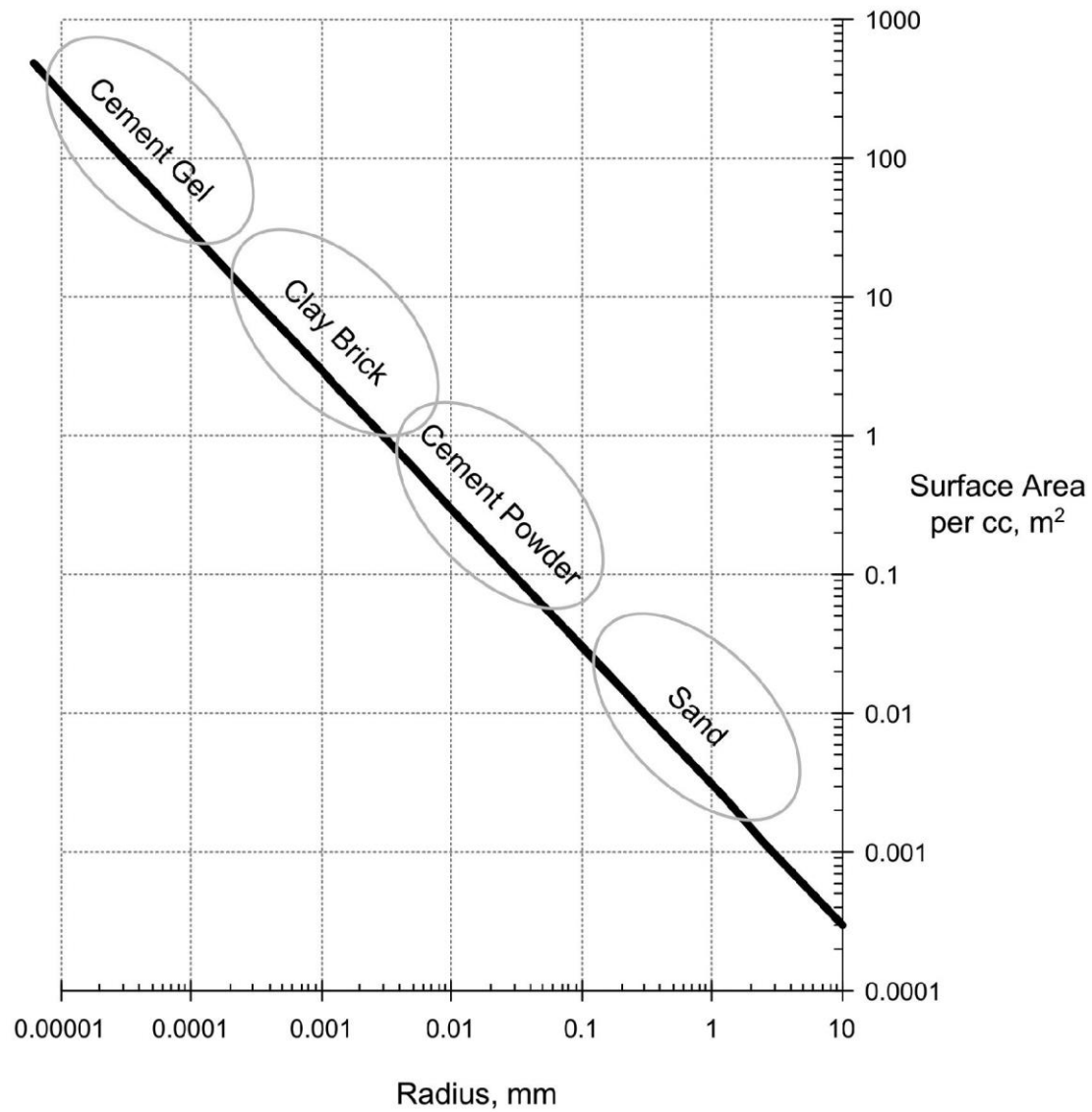
# Calculating capillary rise



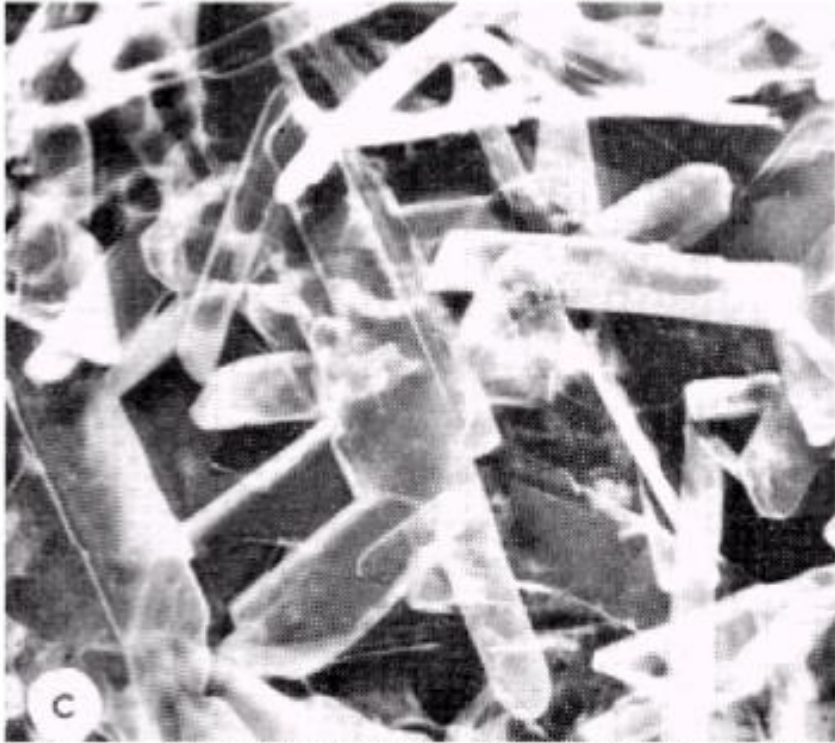
# Capillary rise versus diameter



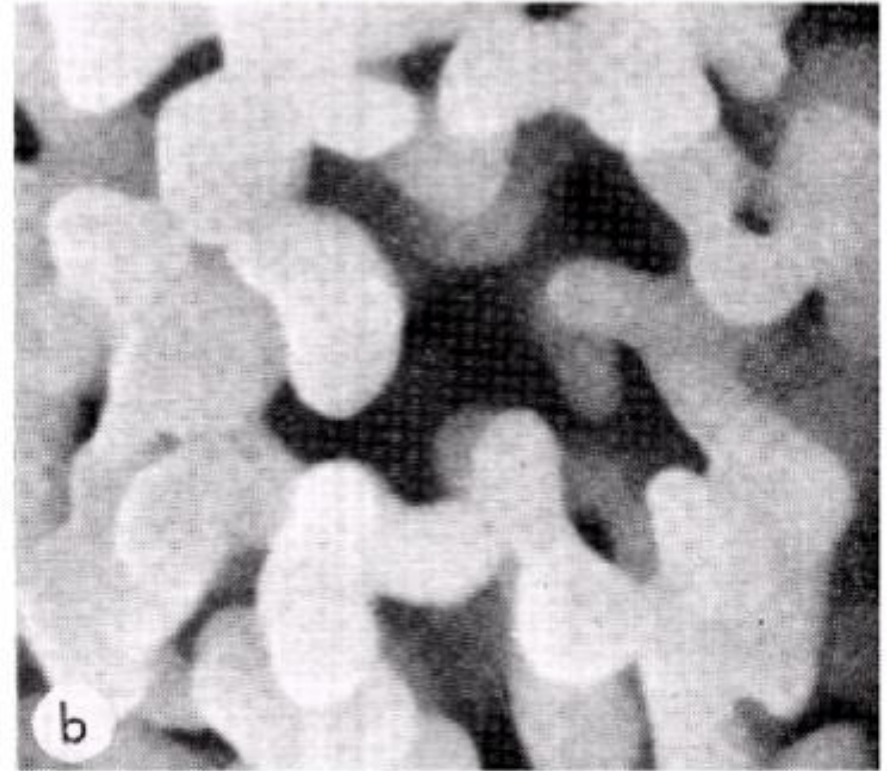




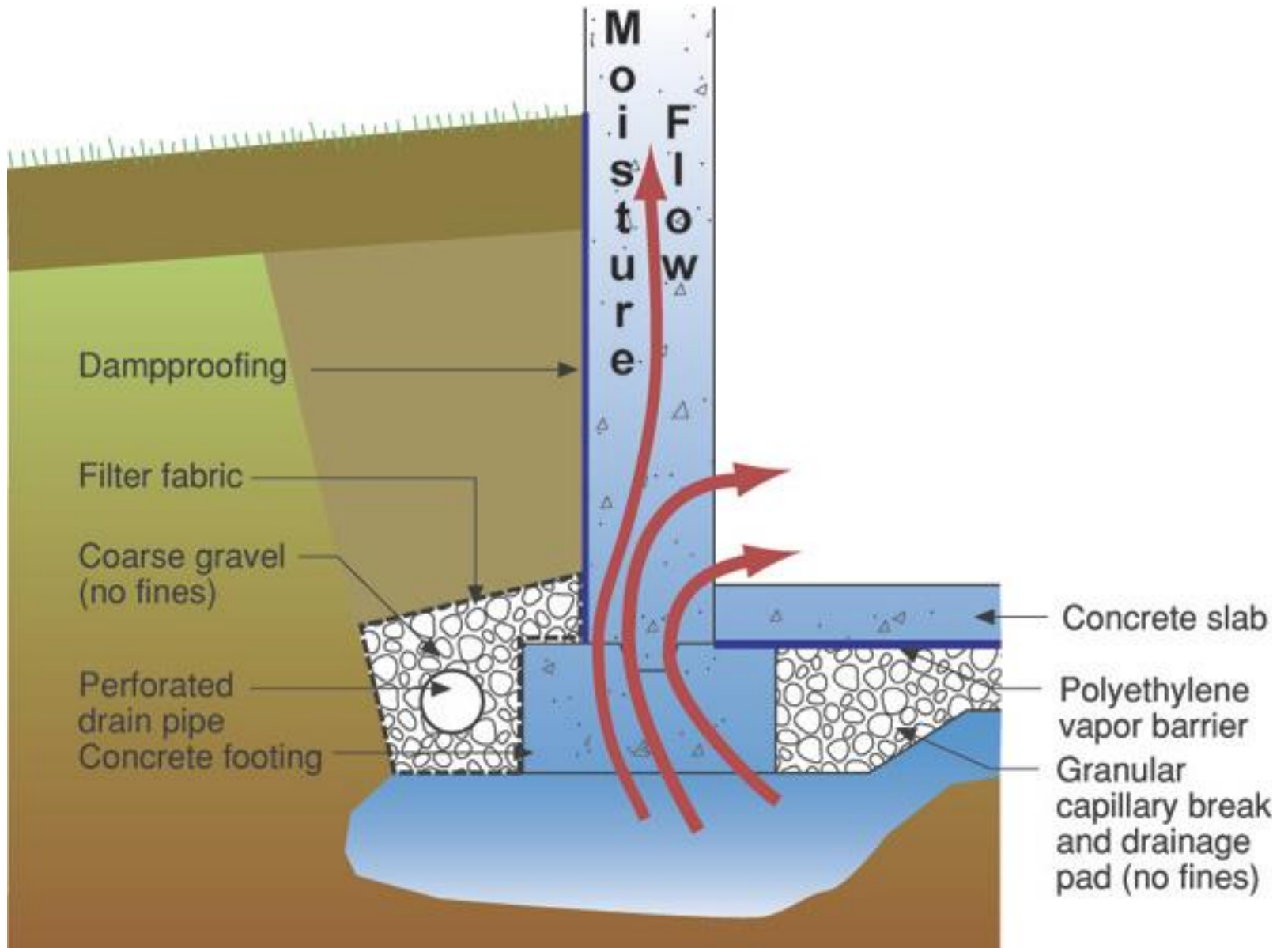
**Surface area vs. particle size**  
From Straube & Burnett, 2005

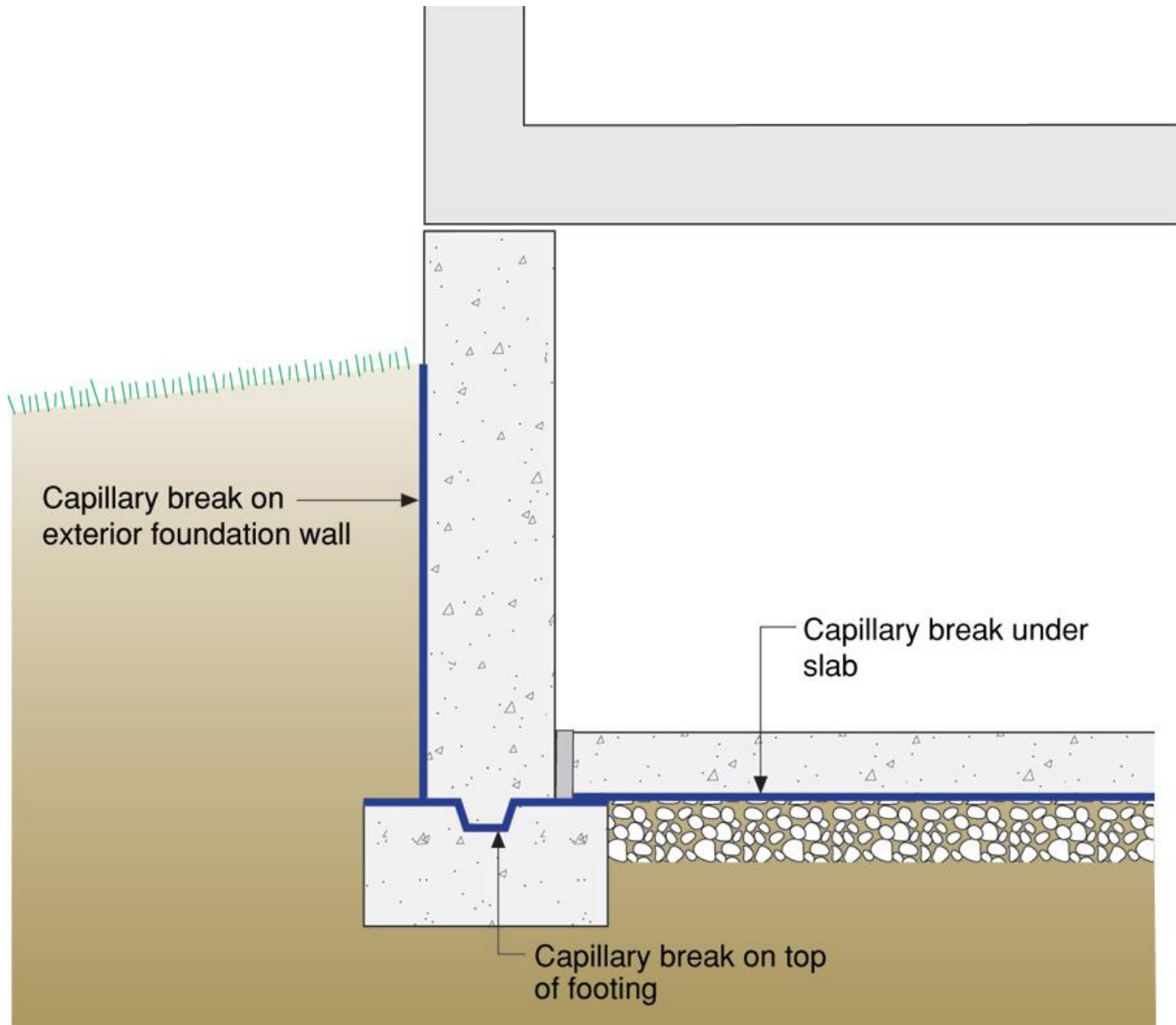


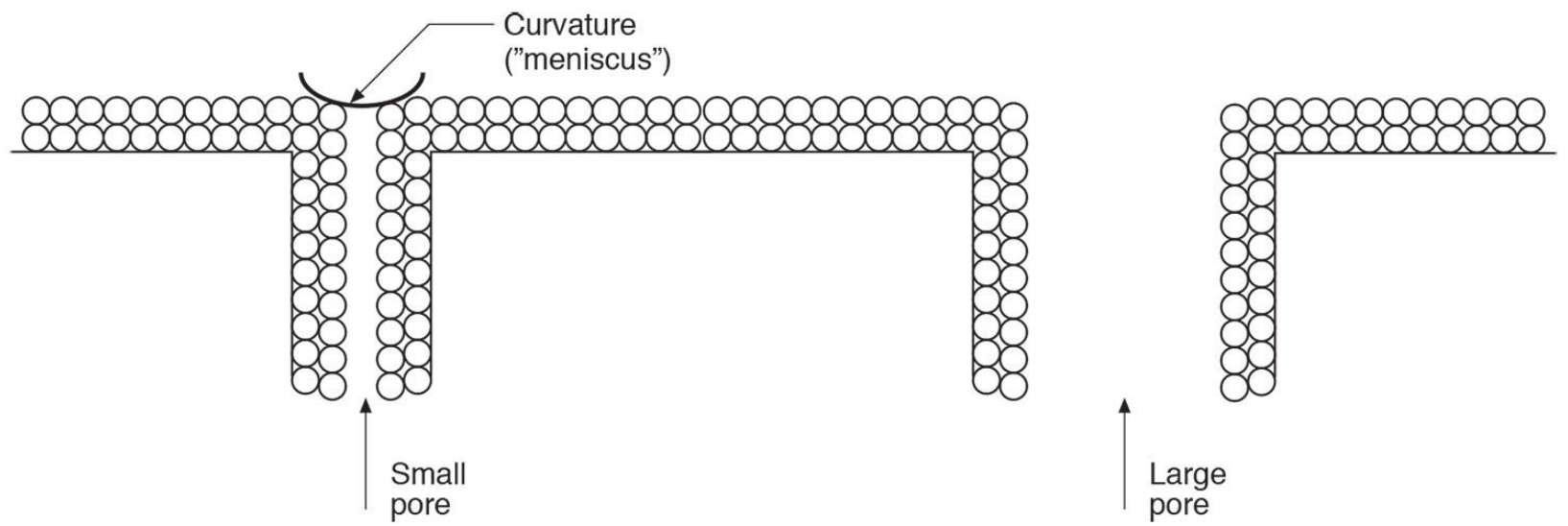
*Figure 1c. Gypsum, hydrated from plaster of paris and water, porosity 30 per cent.*



*Figure 1b. Brick, sintered clay, porosity 40 per cent.*









Vapor

Diffusion

Convective Flow

Vapor Concentration

Air Pressure

Adsorbate

Surface Diffusion

Concentration

Liquid

Capillary Flow

Osmosis

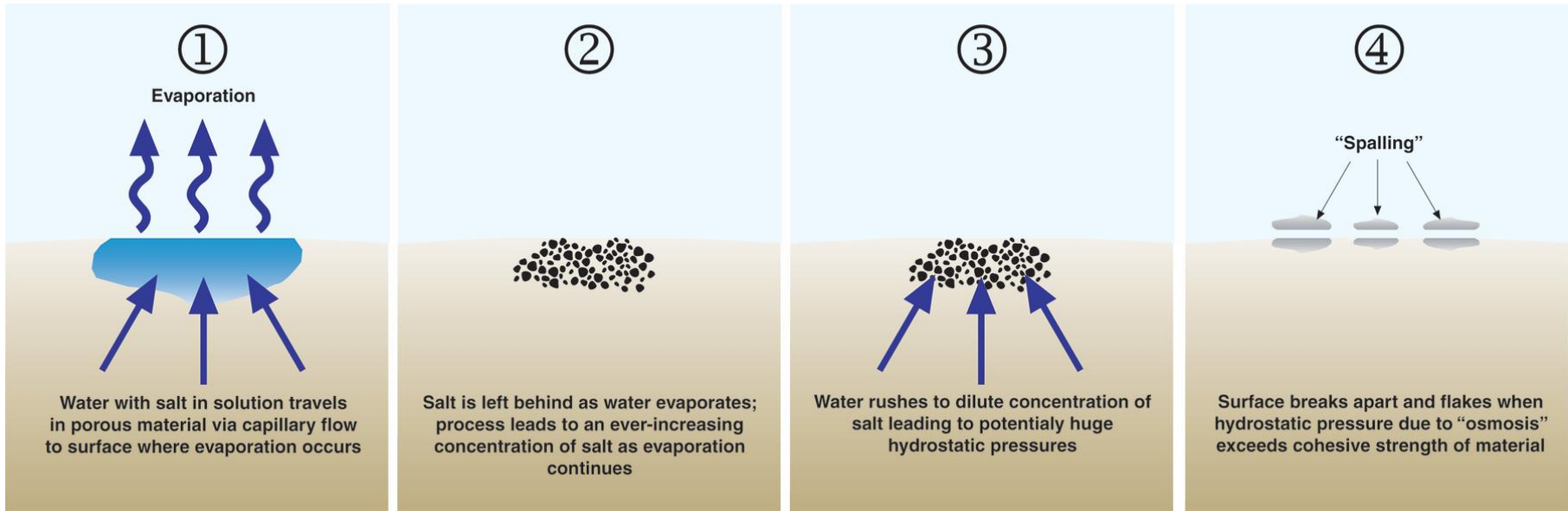
Suction Pressure

Solute Concentration

# Capillarity + Salt = Osmosis

- Mineral salts carried in solution by capillary water
- When water evaporates from a surface the salts left behind form crystals in process called efflorescence
- When water evaporated beneath a surface the salts crystallize within the pore structure of the material in called sub-efflorescence
- The salt crystallization causes expansive forces that can exceed the cohesive strength of the material leading to spalling

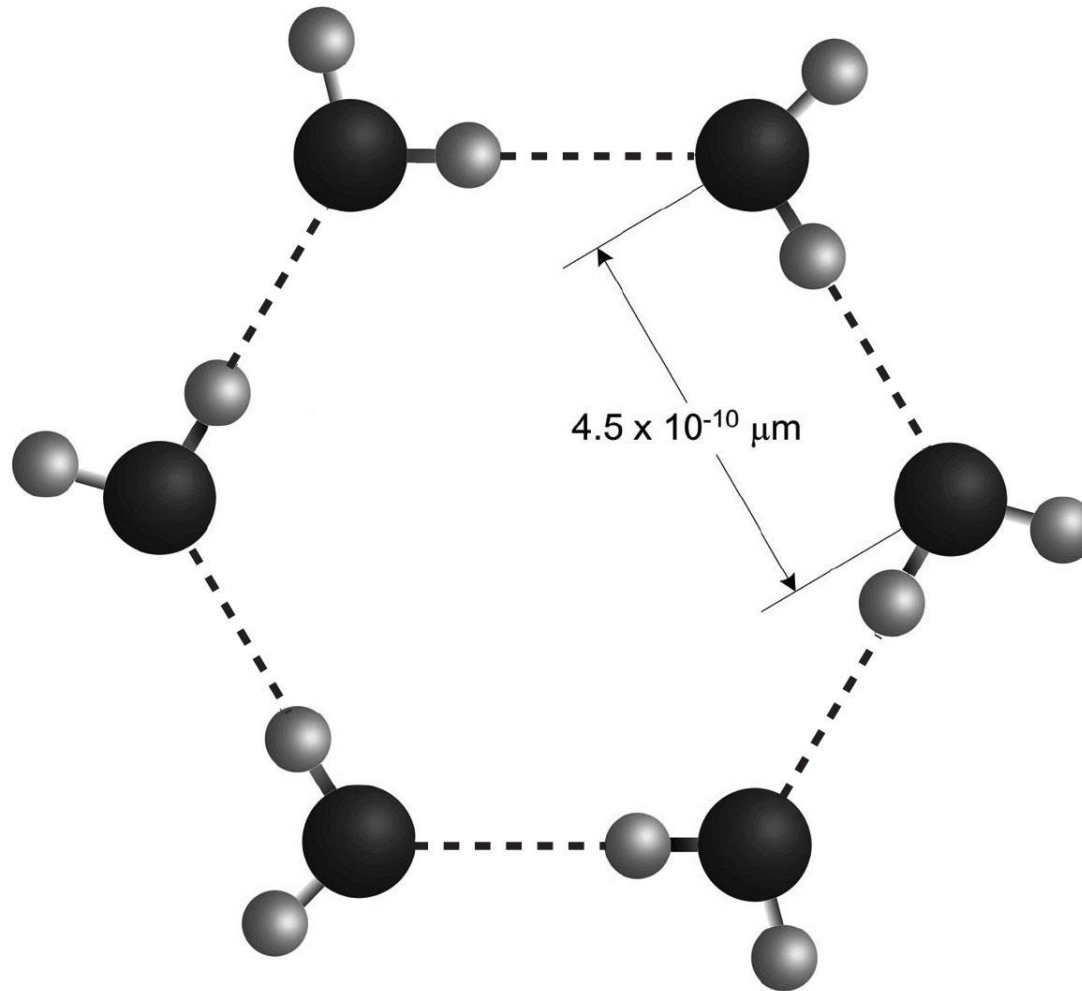




# Pressures

- Diffusion Vapor Pressure 3 to 5 psi
- Capillary Pressure 300 to 500 psi
- Osmosis Pressure 3,000 to 5,000 psi







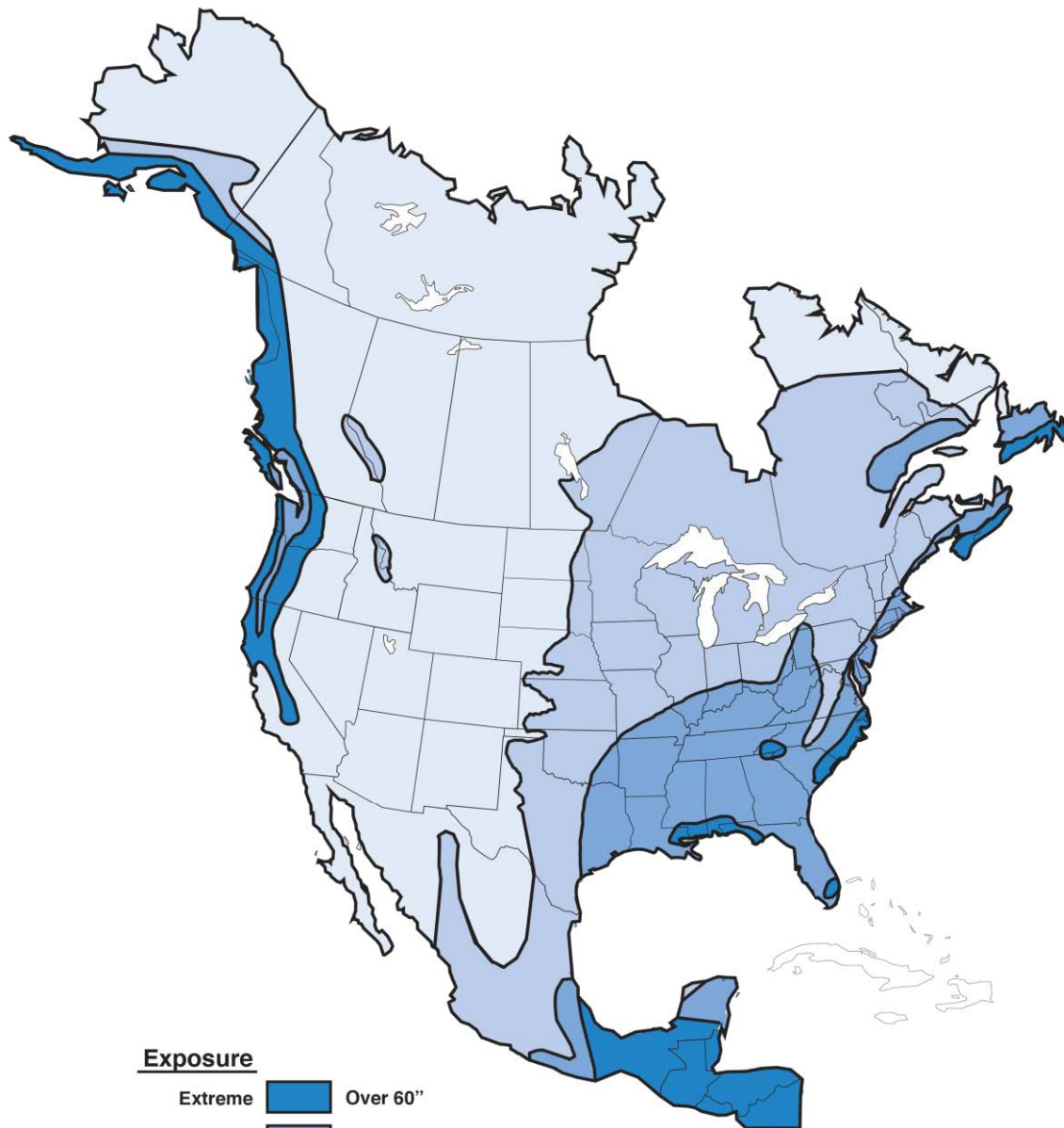
# Freeze-Thaw Damage

Freeze-Thaw Damage  
Freezing Temperatures  
Water  
Susceptible Brick



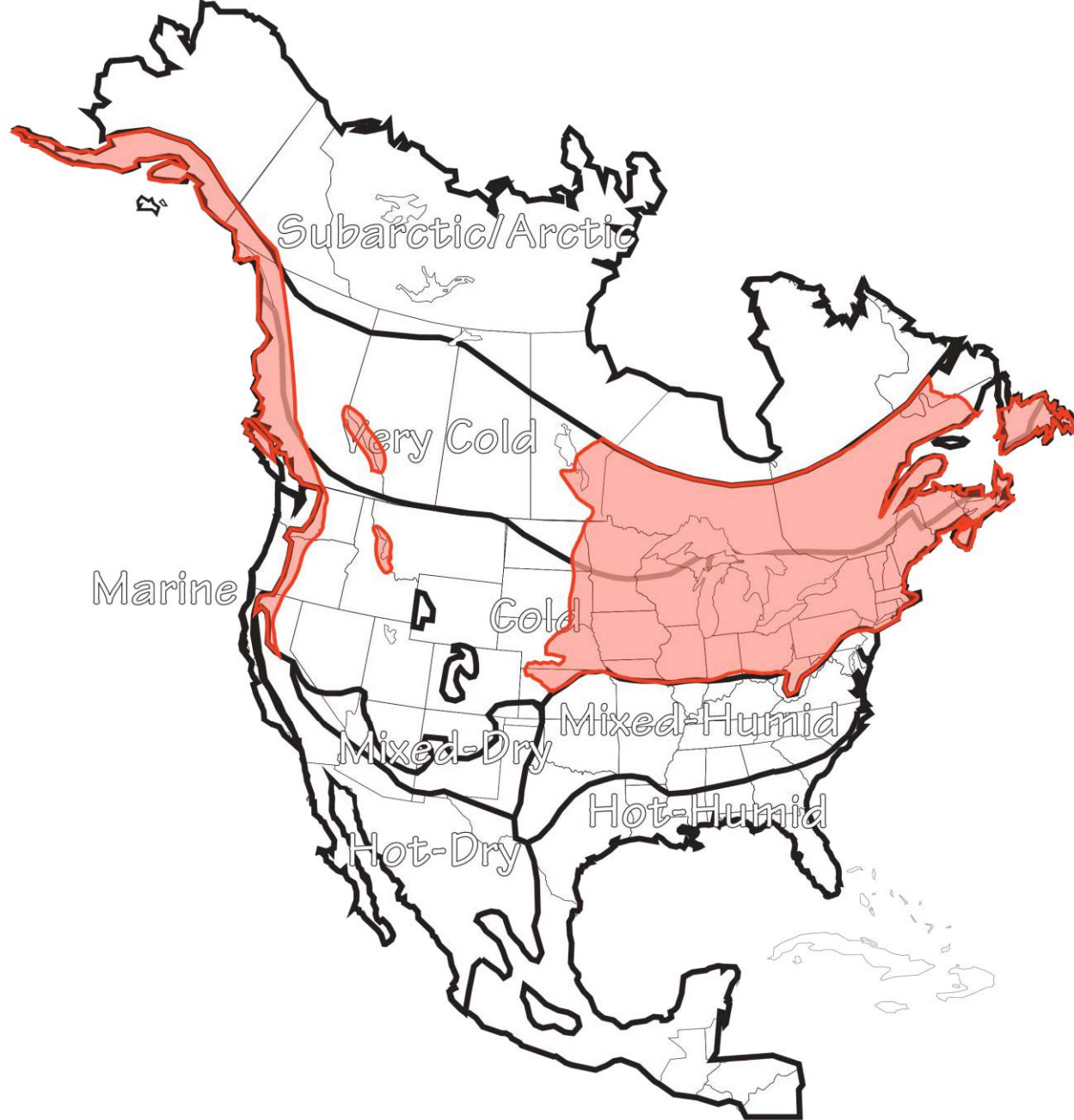




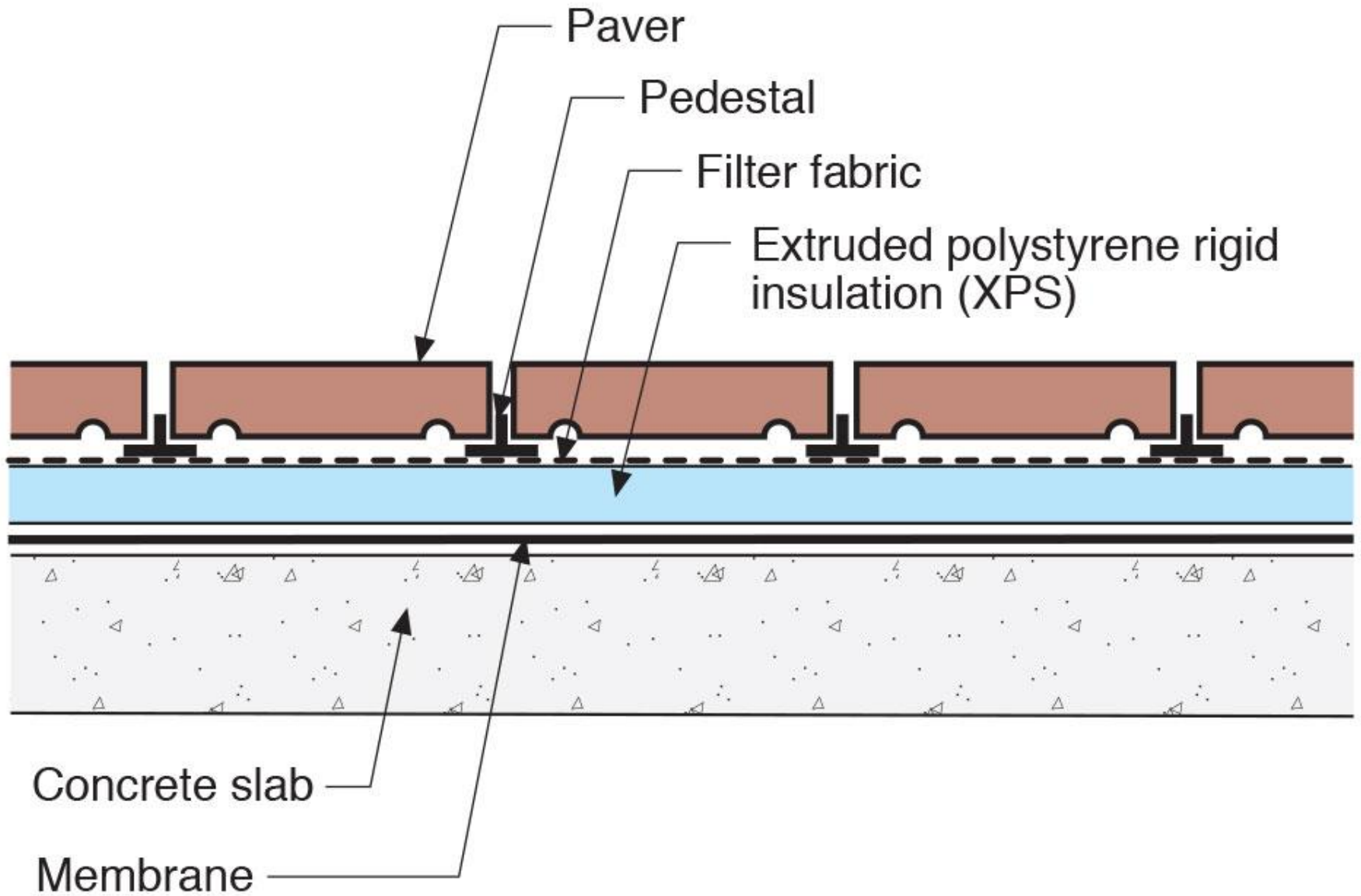


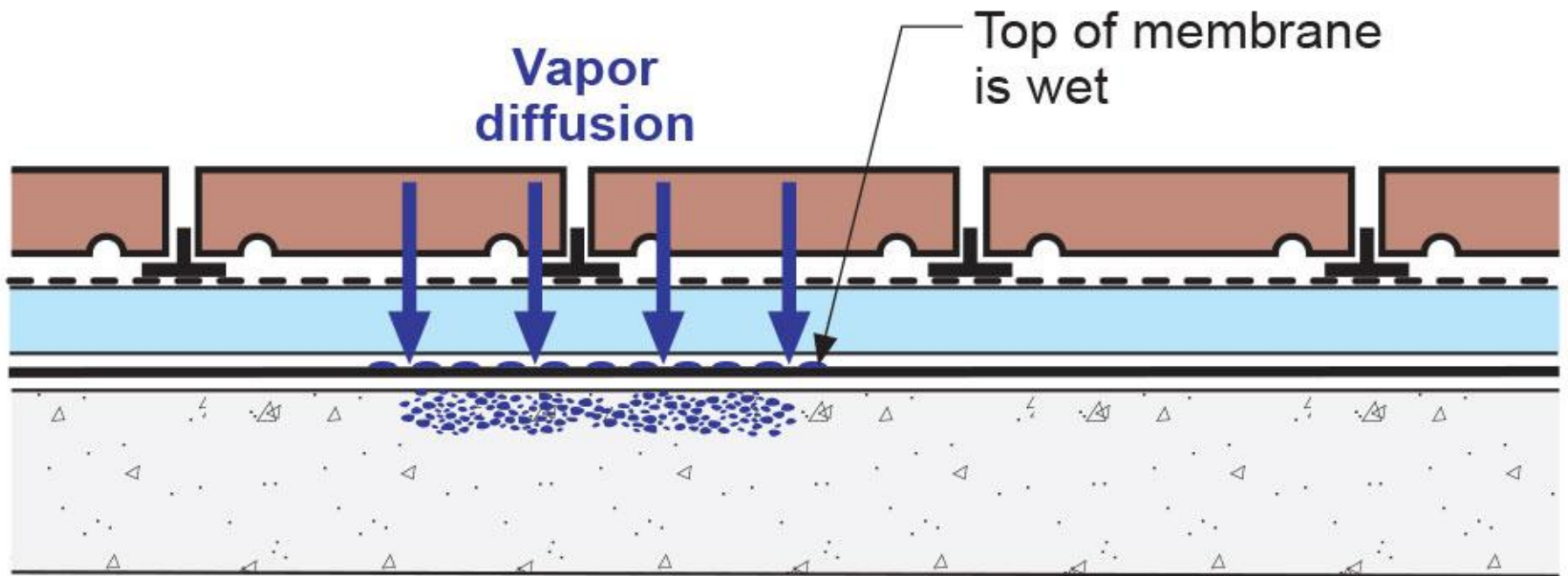
**Exposure**

- Extreme Over 60"
- High 40" - 60"
- Moderate 20" - 40"
- Low Under 20"

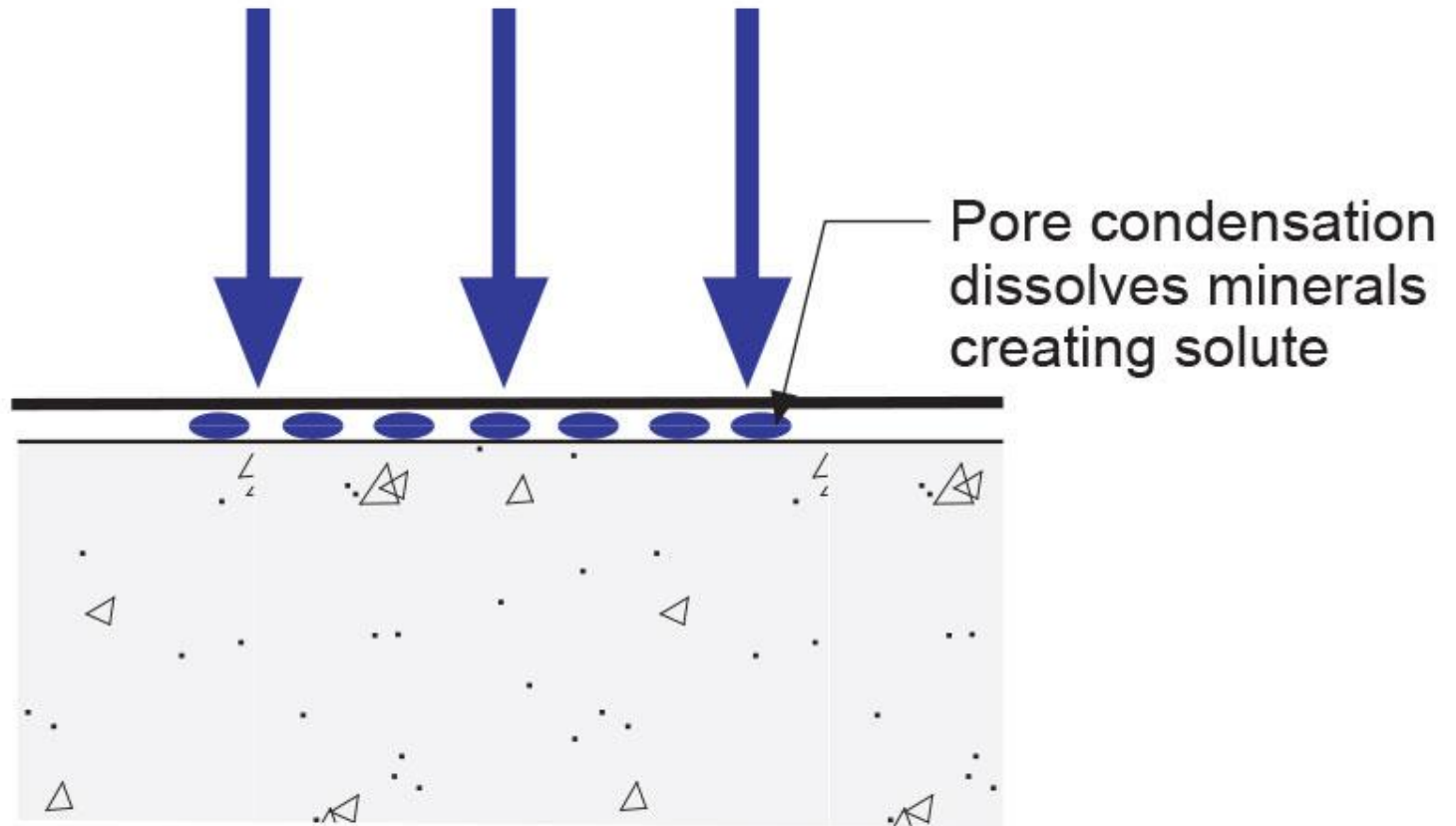


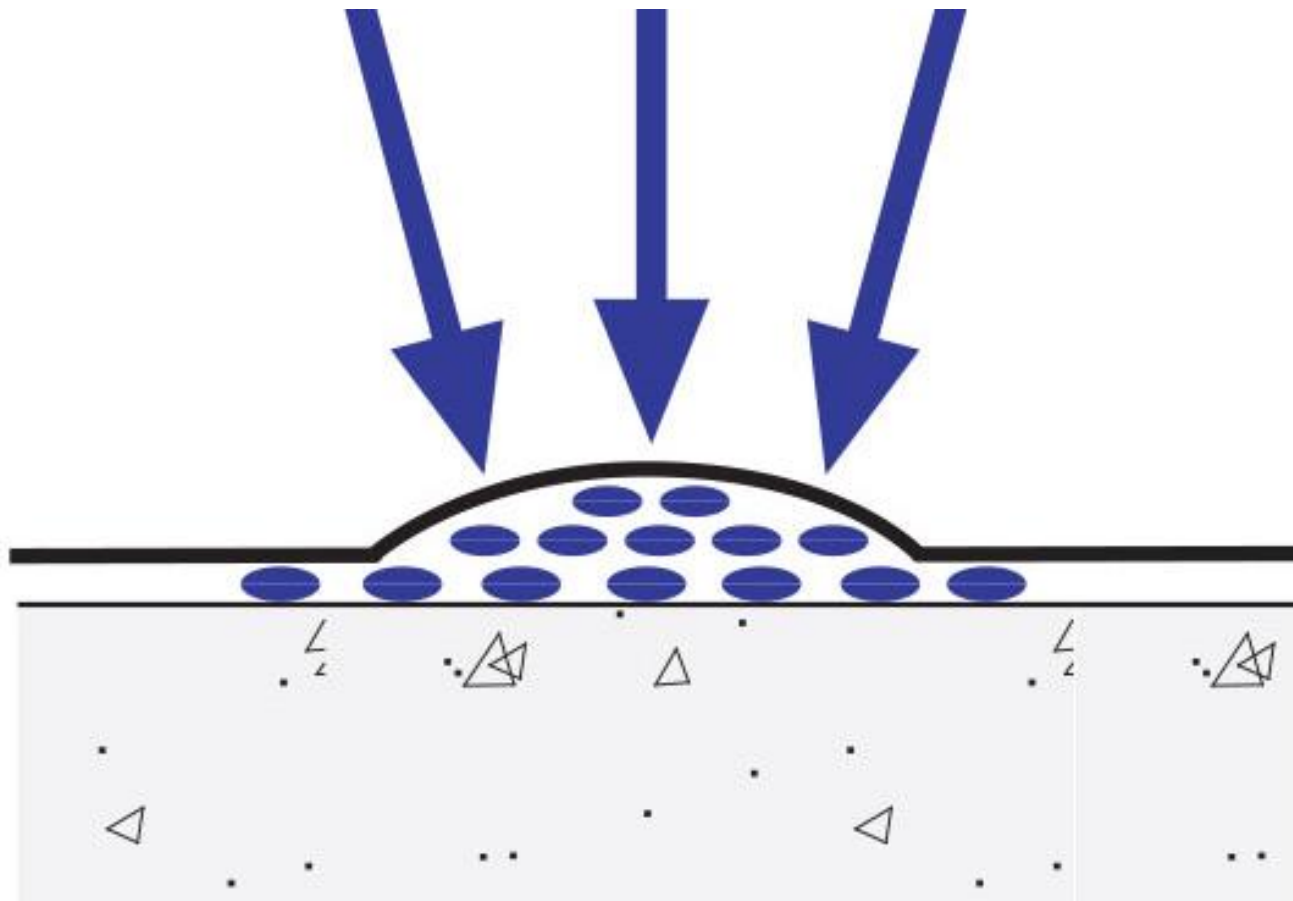
# More Osmosis





## Vapor diffusion









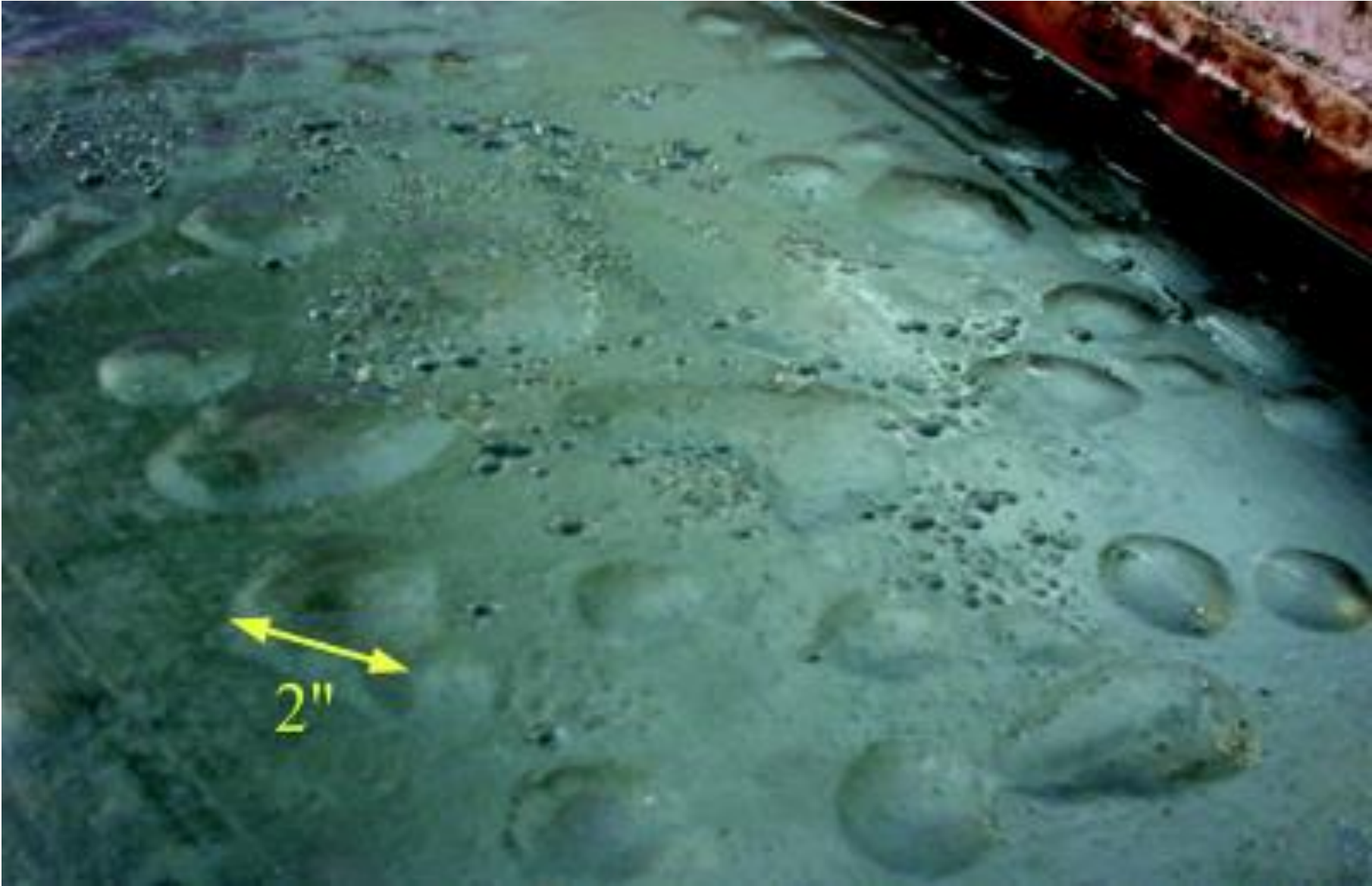
# Paver Water Beds!





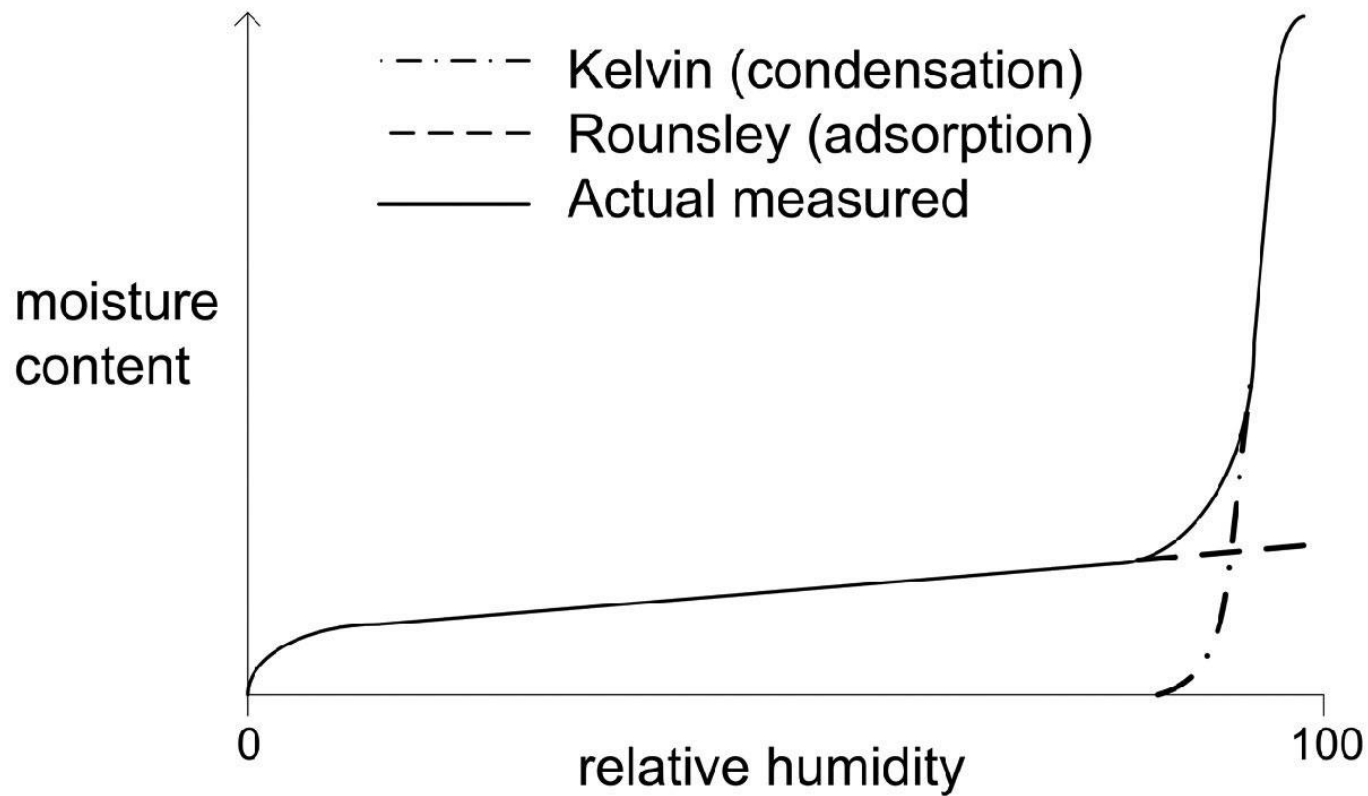
**Really Heavy  
Pink Stuff**

**Liquid Waterproofing  
over Concrete Deck**

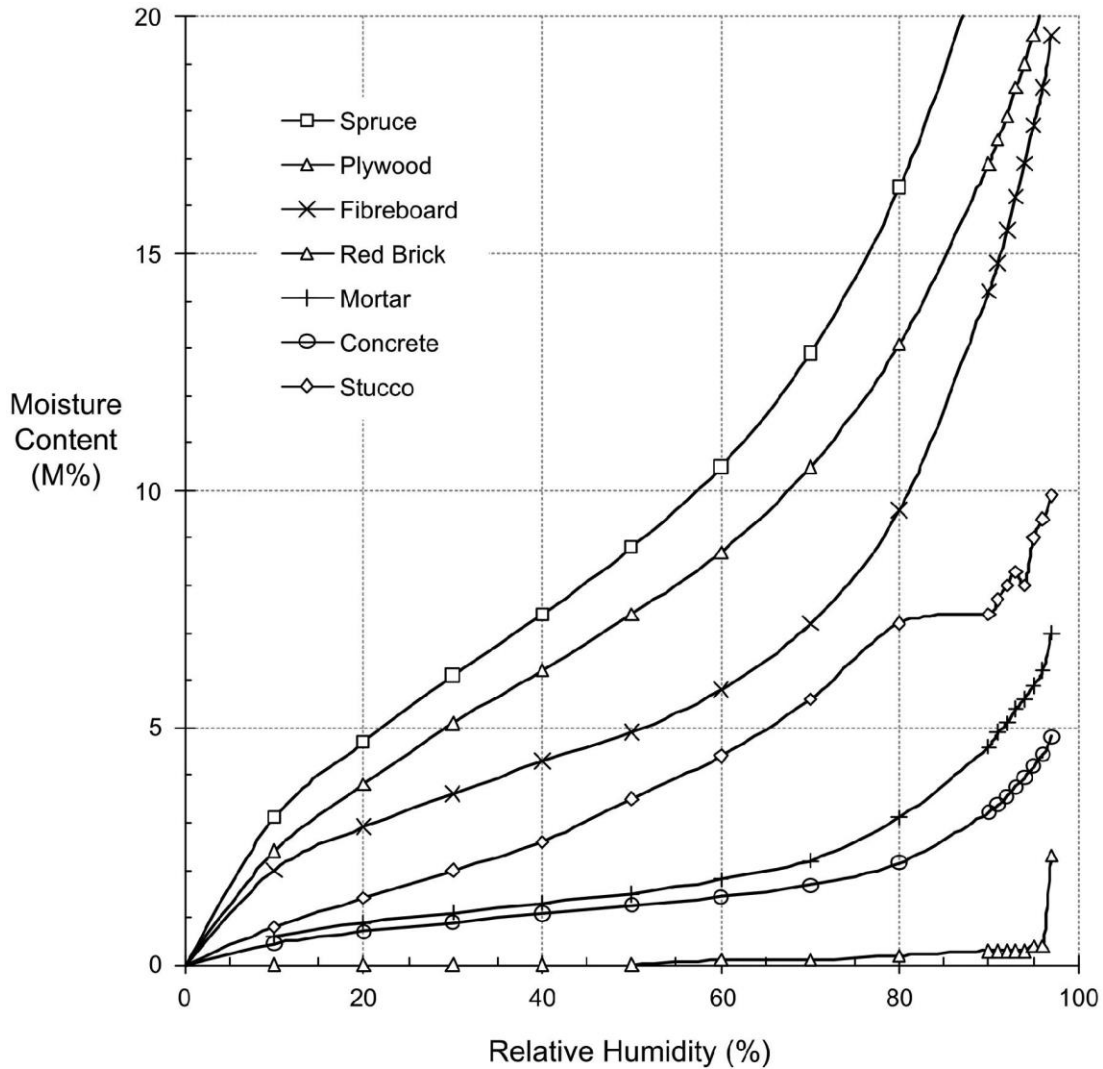




# Combined Flows



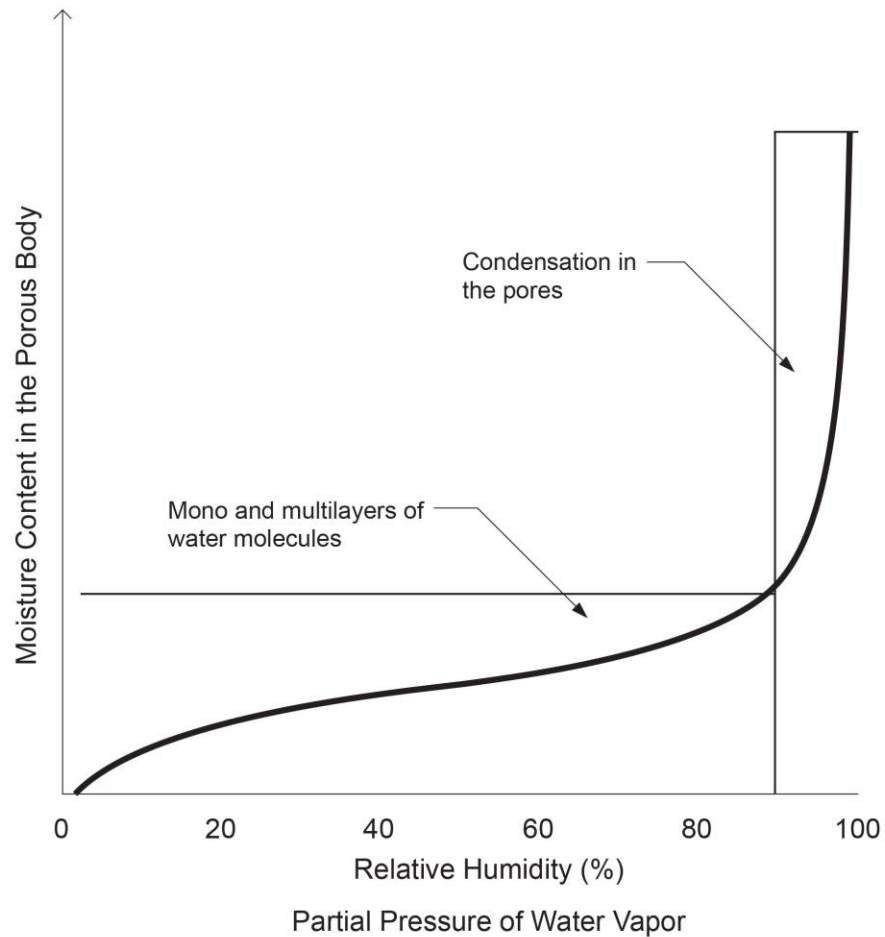
**Typical predicted sorption isotherm according to Kelvin equation  
and modified BET theory**  
From Straube & Burnett, 2005



Sorption isotherm for several building materials [Kumaran 2002]

From Straube & Burnett, 2005

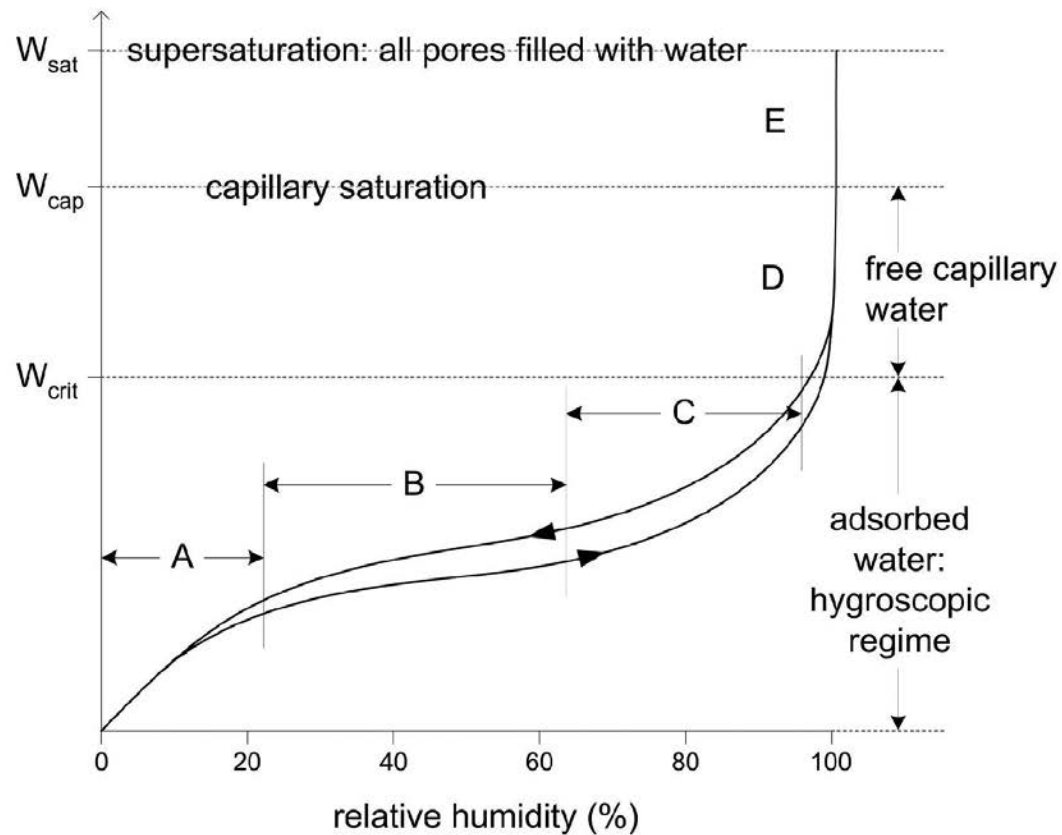




Change in the storage of moisture in a porous building material as the partial pressure of water vapor in the ambient air increases from zero to full saturation value at a given temperature.

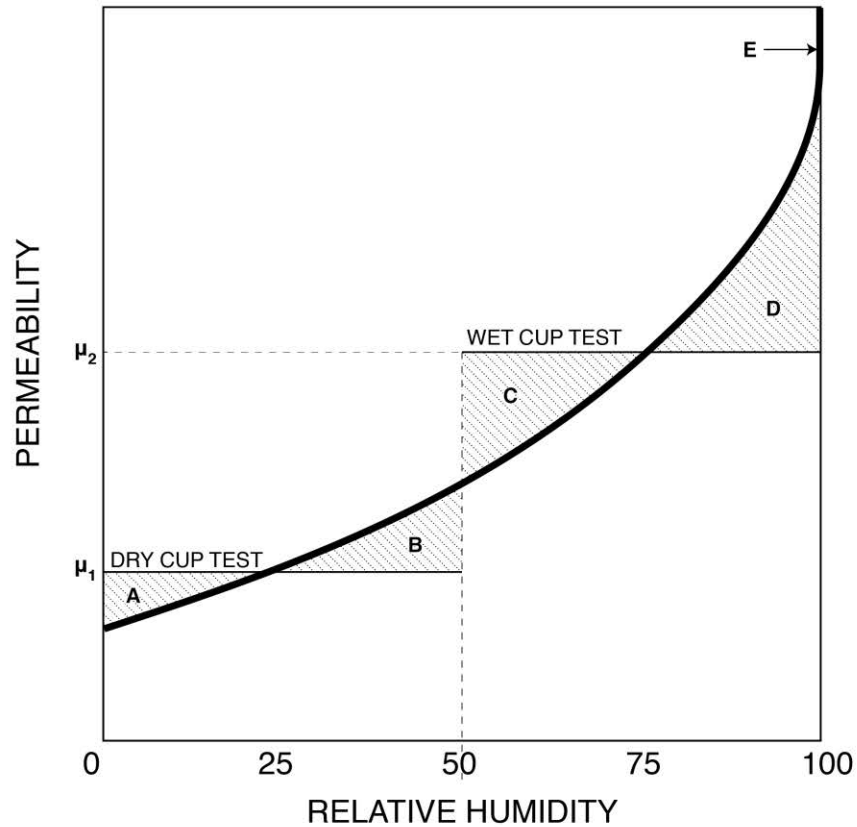
### Sorption Curve

From M.K. Kumaran, ASTM MNL 18-2nd Edition,  
Moisture Control in Buildings, 2009



- A: Single-layer of adsorbed molecules
- B: Multiple layers of adsorbed molecules
- C: Interconnected layers (internal capillary condensation)
- D: Free water in Pores, capillary suction
- E: Supersaturated Regime

**Regimes of moisture storage in a hygroscopic porous material**  
 From Straube & Burnett, 2005

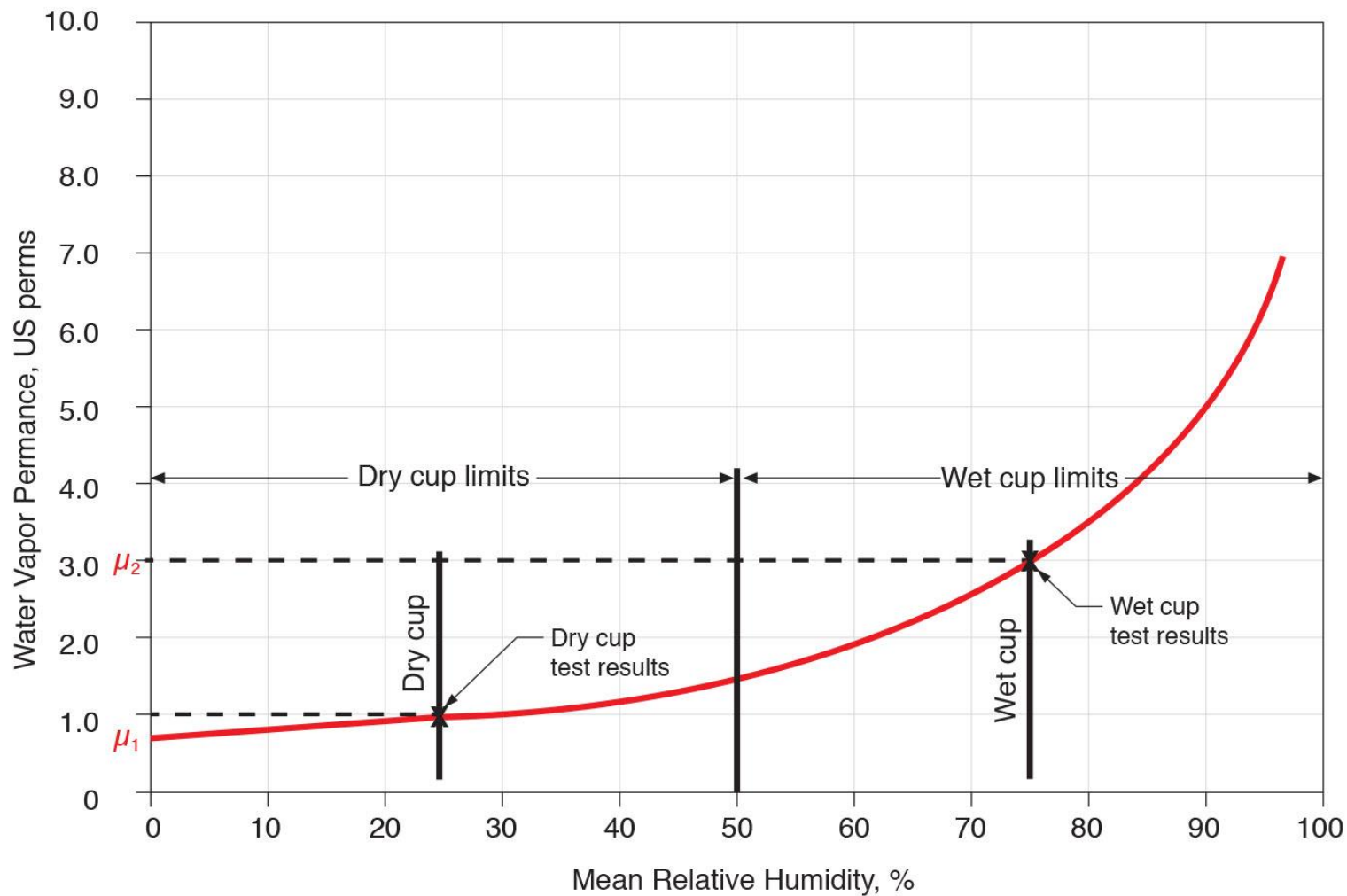


- A - Single-layer of absorbed molecules
- B - Multiple layers of absorbed molecules
- C - Interconnected layers (internal capillary condensation)
- D - Free water in pores, capillary suction
- E - Supersaturated regime

Relationship between Dry Cup and Wet Cup  
Adapted from Joy & Wilson, 1963



## Water Vapor Permeance vs. Relative Humidity



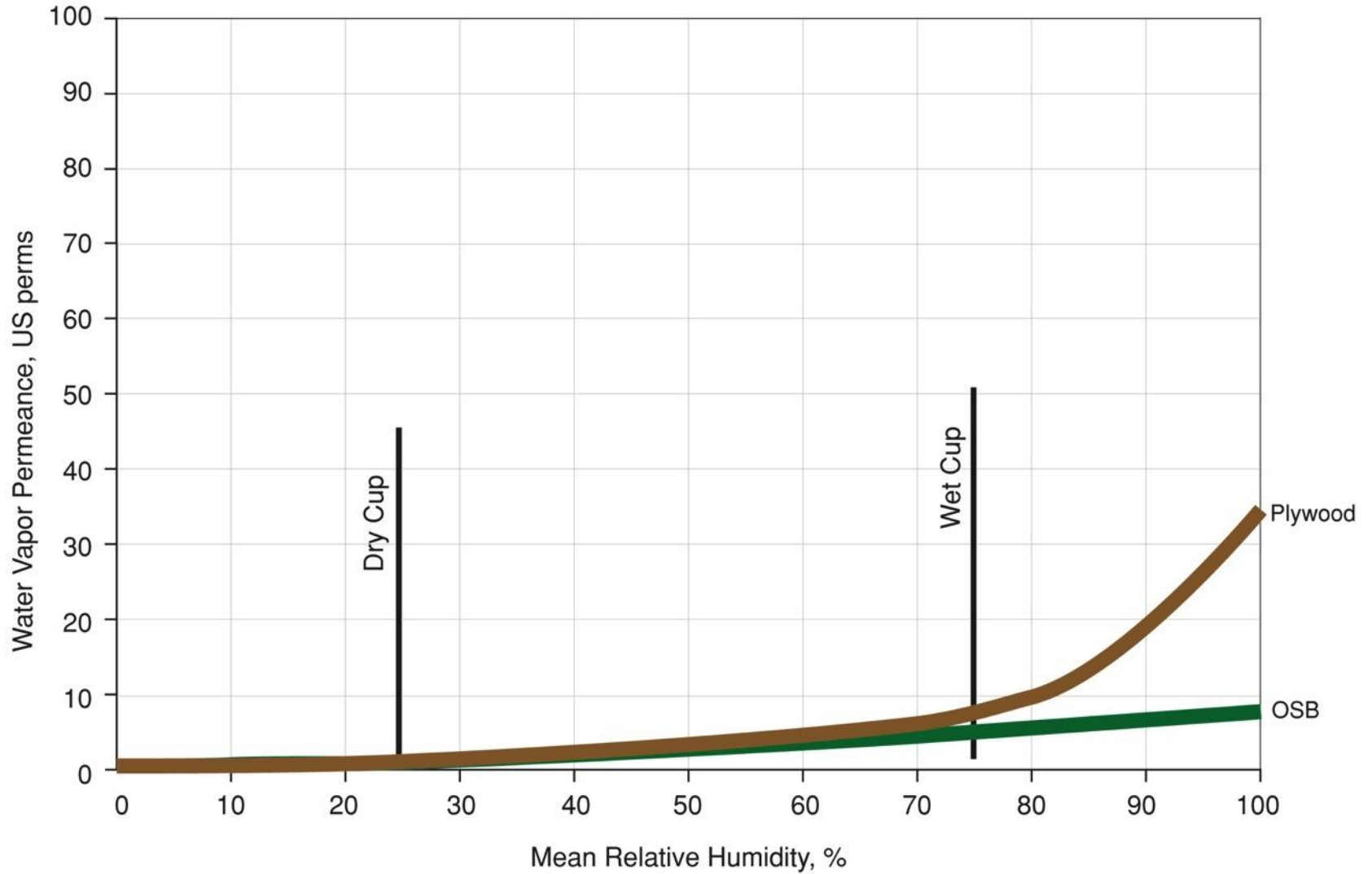
$\mu_1$  = Dry cup permeance

$\mu_2$  = Wet cup permeance



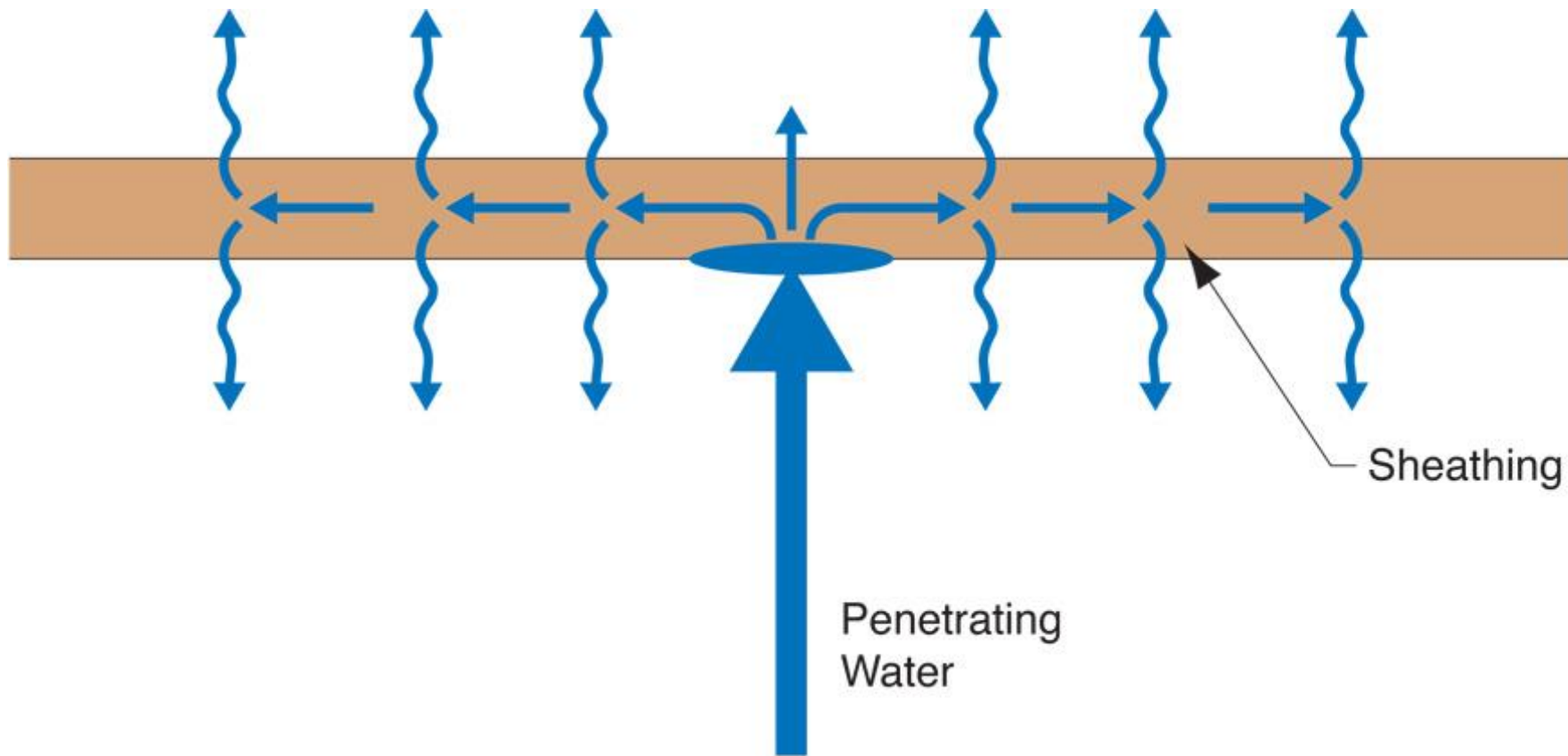


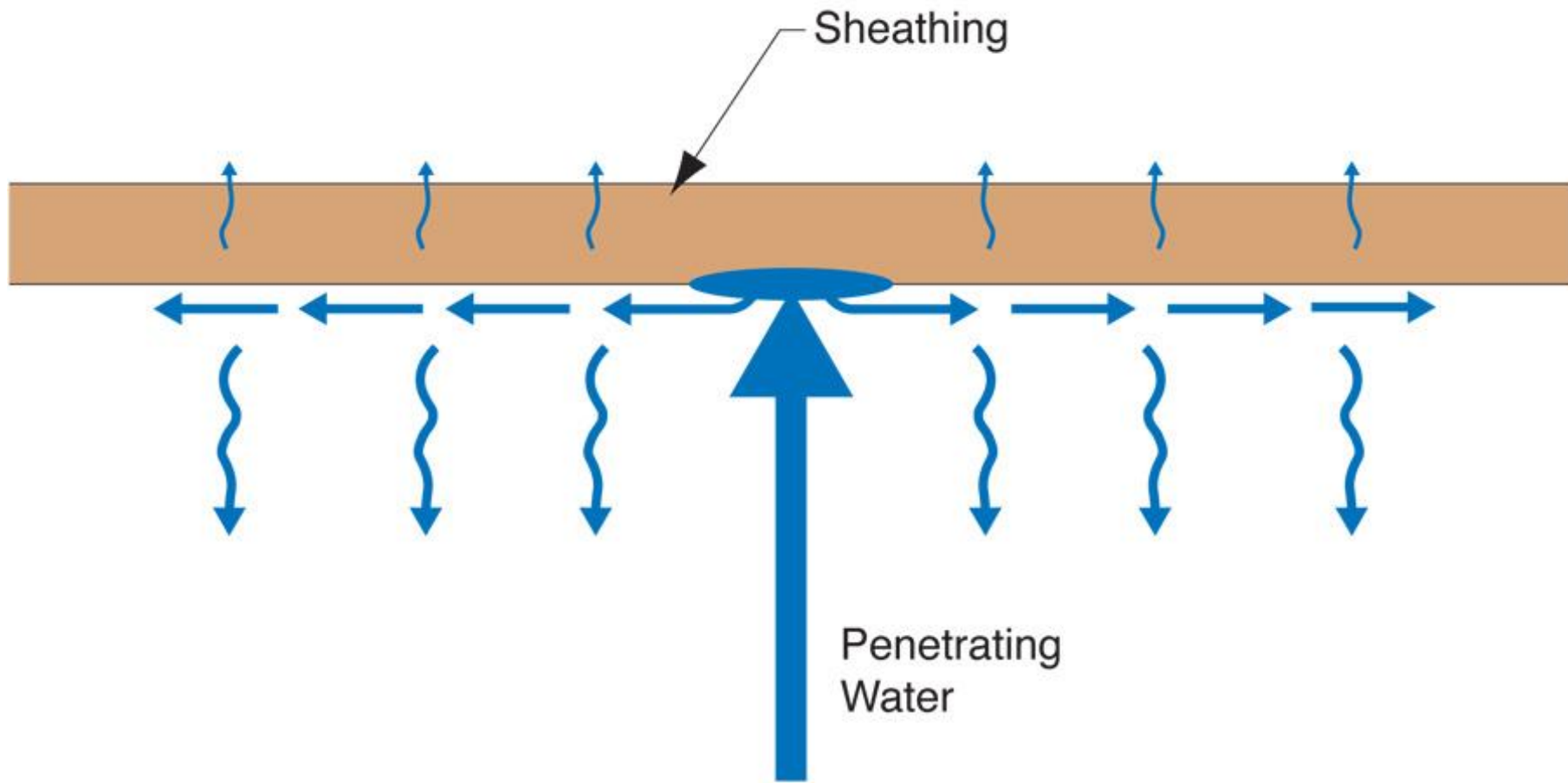
## Water Vapor Permeance of Sheathing Materials

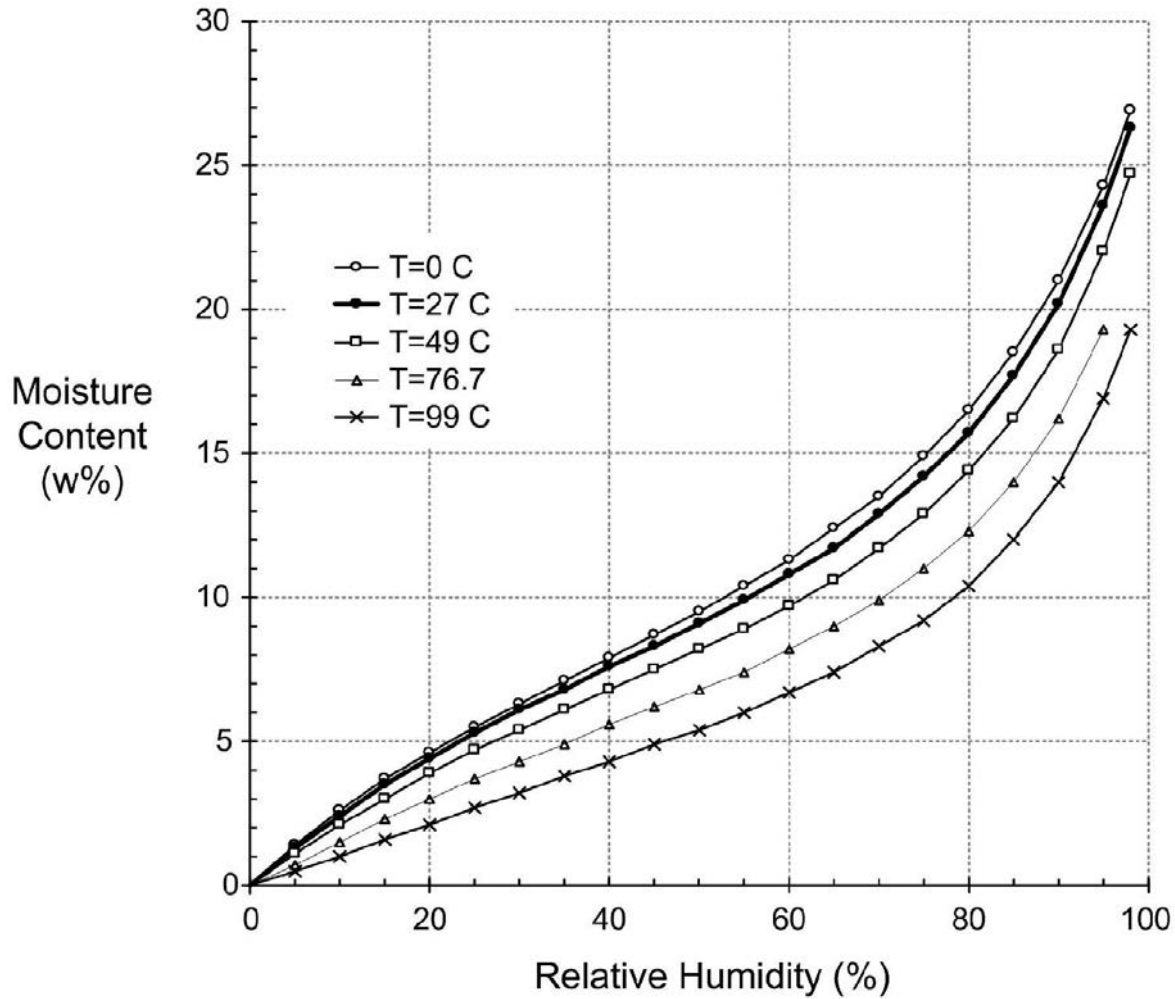






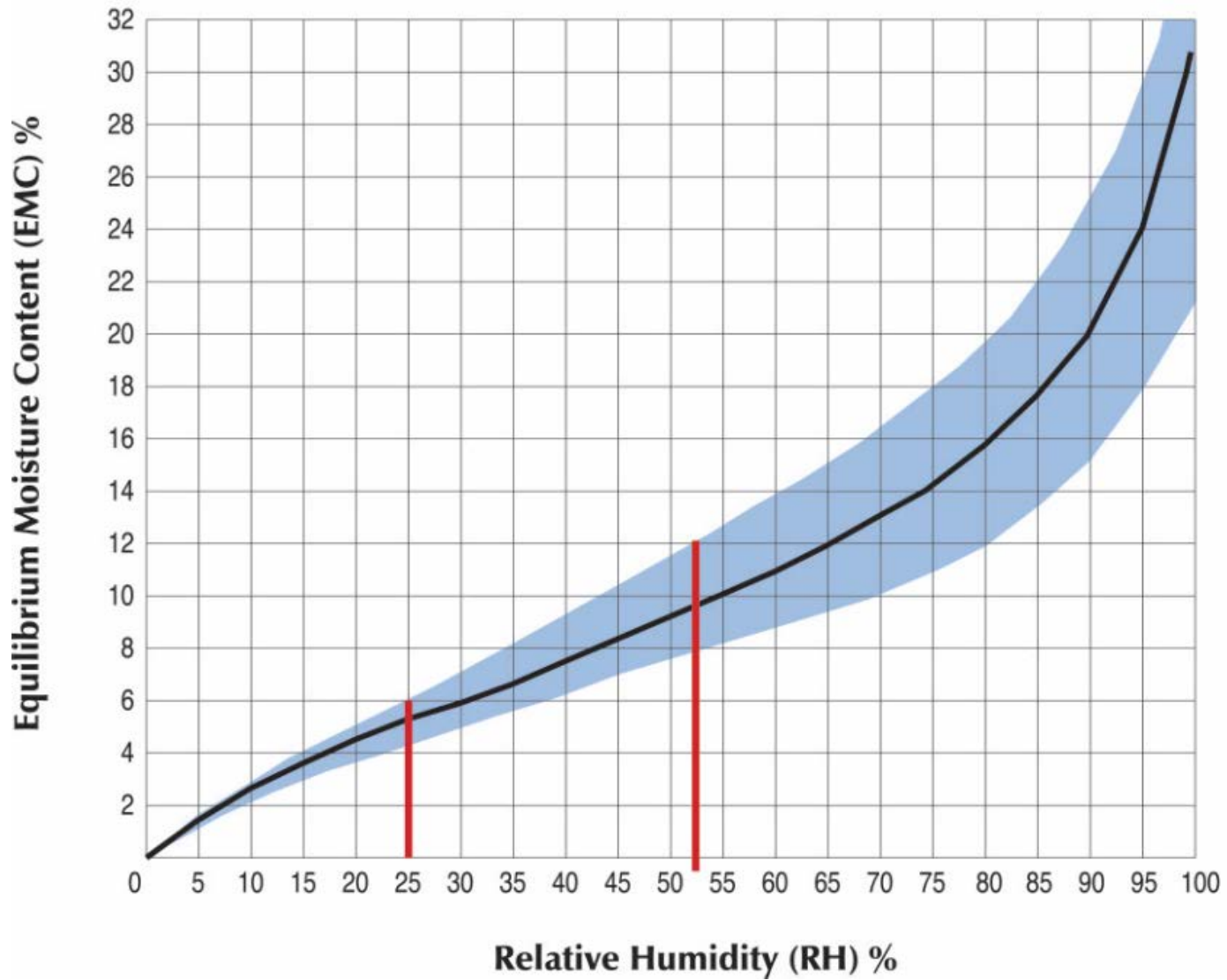






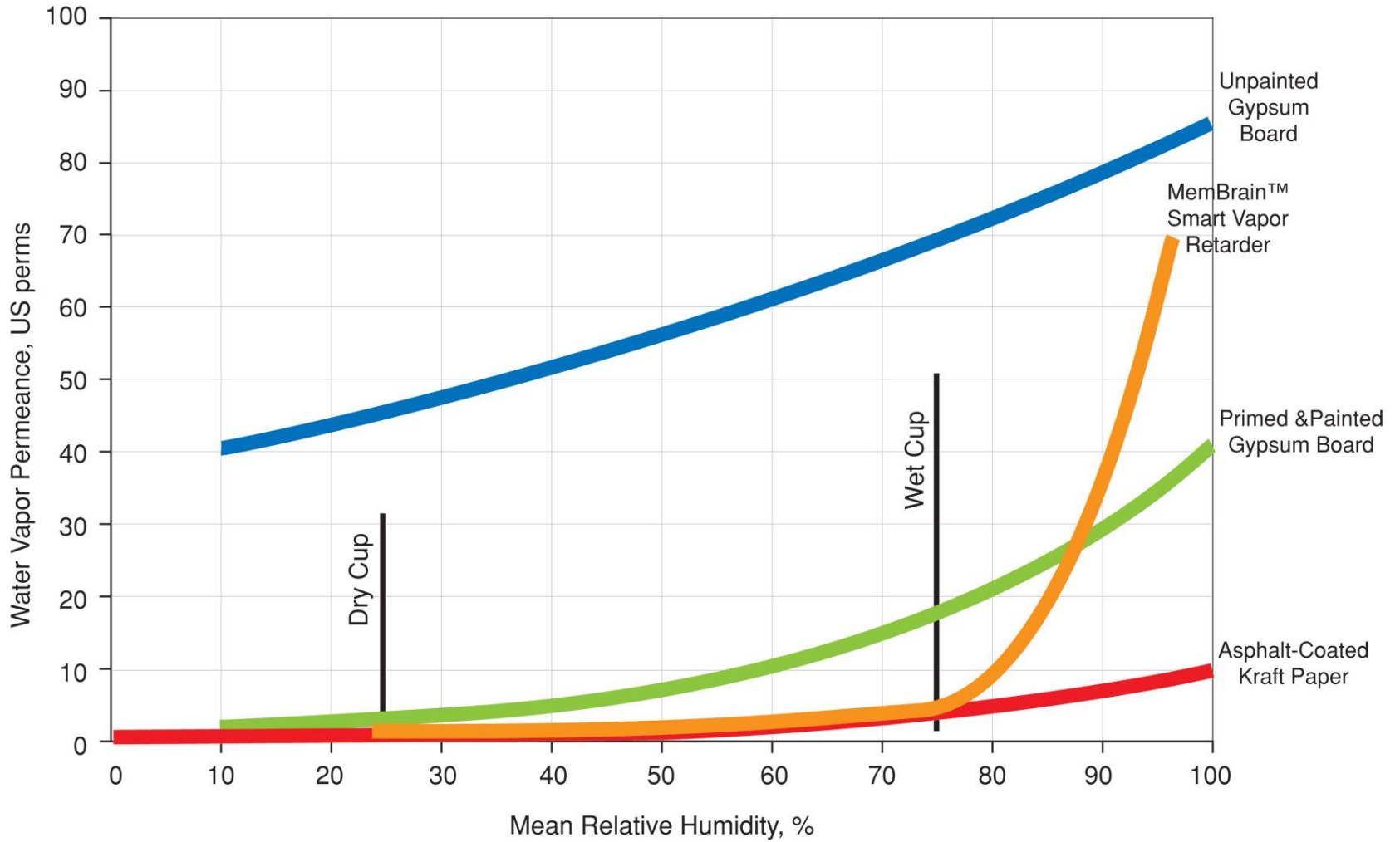
Average sorption isotherm for wood as a function of temperature  
 From Straube & Burnett, 2005

# Moisture Content vs. Relative Humidity

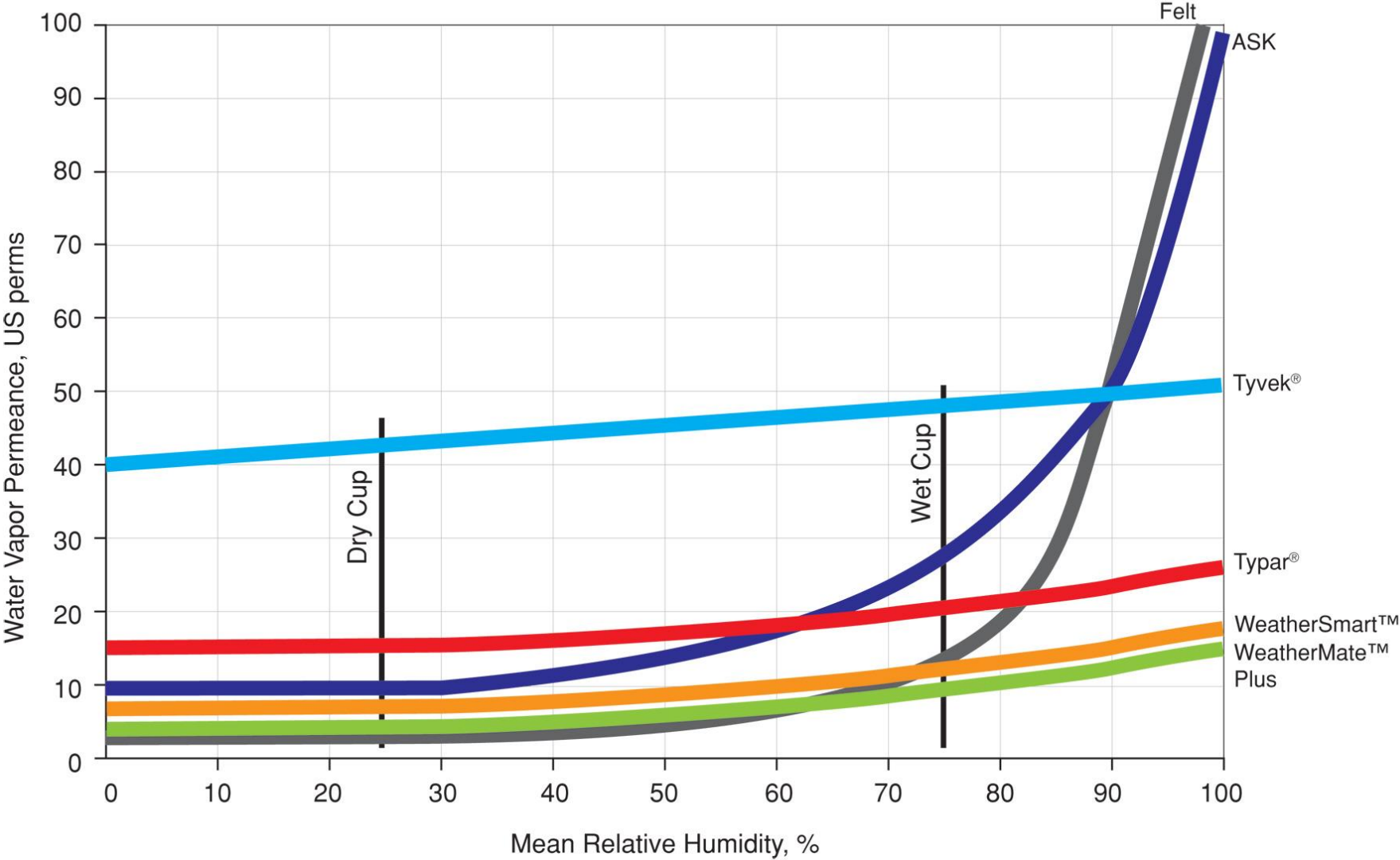




### Water Vapor Permeance of MemBrain™ Smart Vapor Retarder, Primed and Painted Gypsum Board, Unpainted Gypsum Board and Asphalt-Coated Kraft Paper



# Water Vapor Permeance of WRB's





# Laws of Thermodynamics

Zeroth Law – Equal Systems

First Law - Conservation of Energy

Second Law - Entropy

Third Law – Absolute Zero

# 2<sup>nd</sup> Law of Thermodynamics

In an isolated system, a process can occur only if it increases the total entropy of the system

Rudolf Clausius

Moisture Flow Is From Warm To Cold  
Moisture Flow Is From More To Less

Moisture Flow Is From Warm To Cold  
Moisture Flow Is From More To Less

Thermal Gradient – Thermal Diffusion  
Concentration Gradient – Molecular Diffusion

Moisture Flow Is From Warm To Cold  
Moisture Flow Is From More To Less

Thermal Gradient – Thermal Diffusion  
Concentration Gradient – Molecular Diffusion

Vapor Diffusion

# Thermodynamic Potential





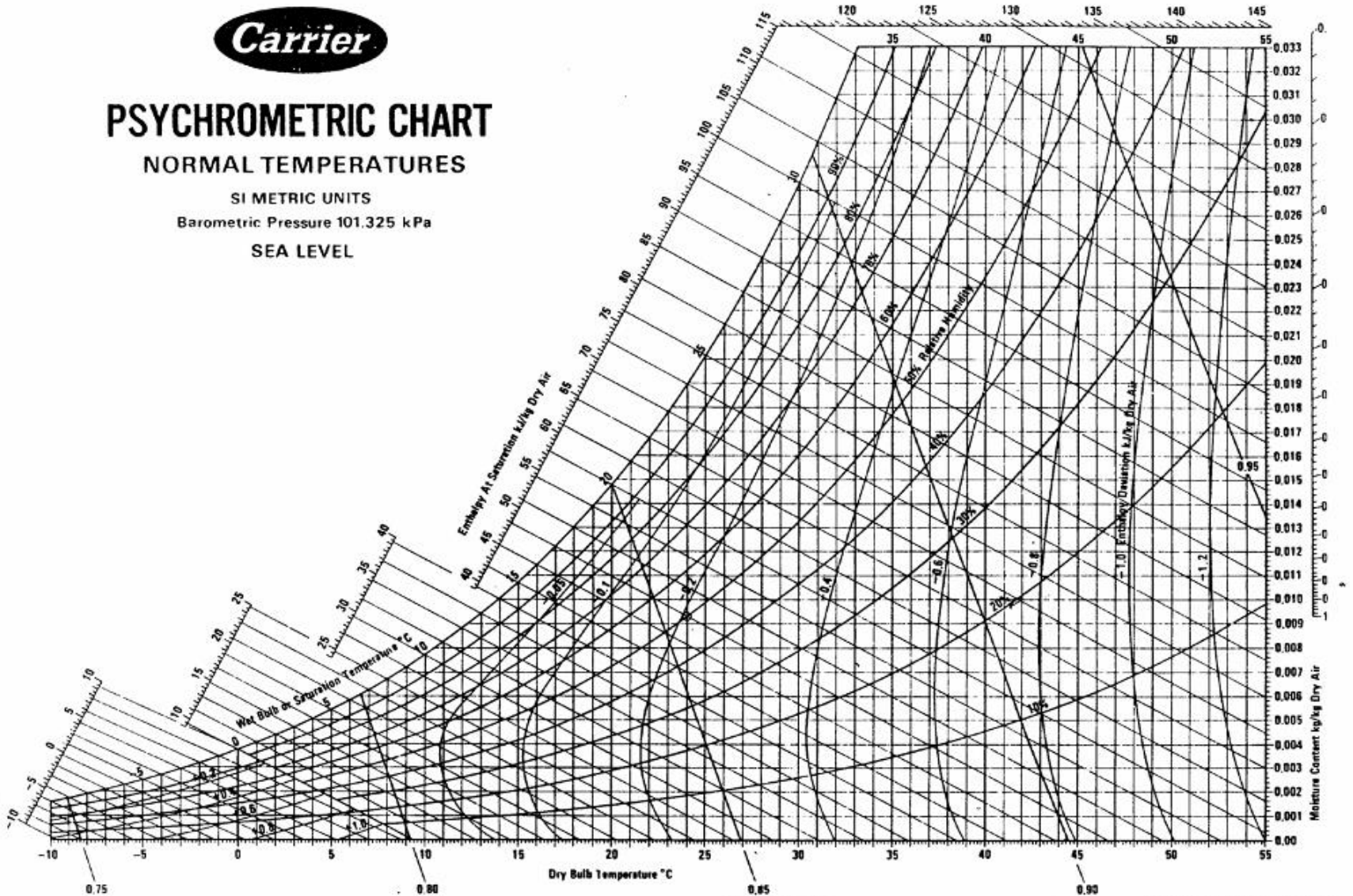
# PSYCHROMETRIC CHART

NORMAL TEMPERATURES

SI METRIC UNITS

Barometric Pressure 101.325 kPa

SEA LEVEL



Below 0°C Properties and Enthalpy Deviation Lines Are For Ice

Copyright ©Carrier Corporation 1975  
Cat. No. 794-002 Printed in U.S.A.

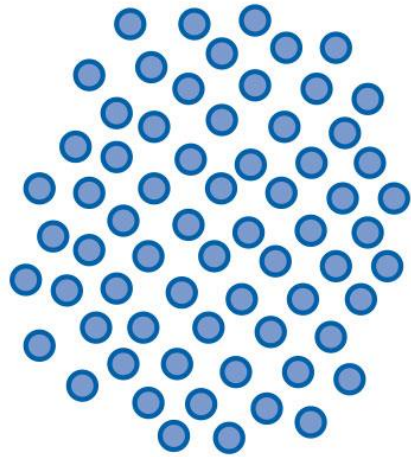
Vapor

Diffusion

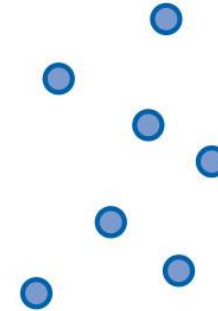
Convective Flow

Vapor Concentration

Air Pressure



**DIFFUSION**

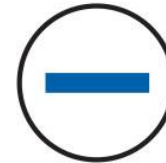


**Higher Dewpoint Temperature  
Higher Water Vapor Density  
or Concentration  
(Higher Vapor Pressure)  
on Warm Side of Assembly**

**Low Dewpoint Temperature  
Lower Water Vapor Density  
or Concentration  
(Lower Vapor Pressure)  
on Cold Side of Assembly**

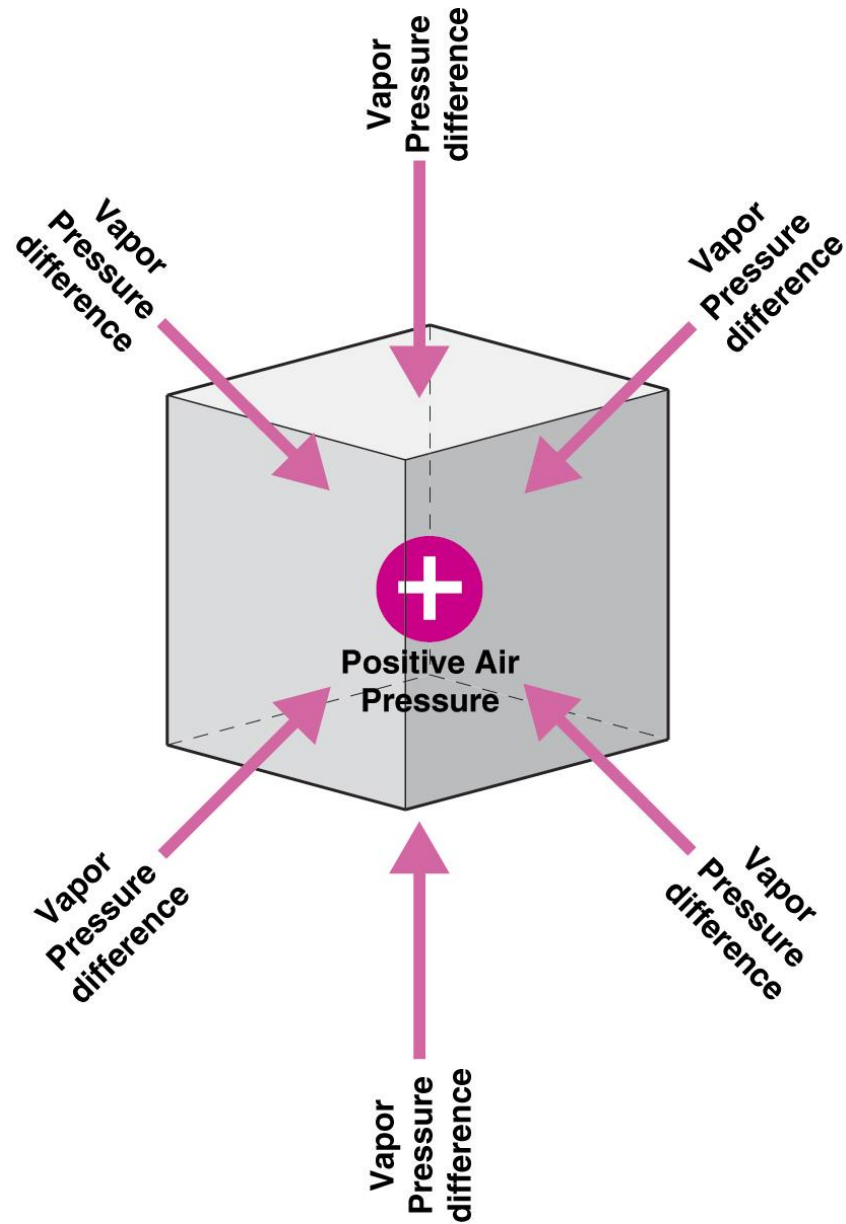


**AIR TRANSPORT**



**Higher Air  
Pressure**

**Lower Air  
Pressure**

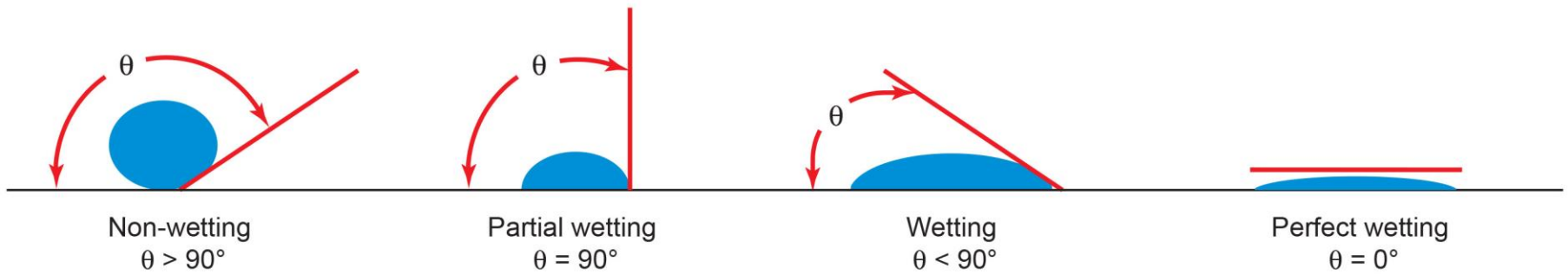


# How Does Wetting Occur?



- “non-wettable” surface
- water repellent surface
- hydrophobic surface
- water more attracted to itself than to surface
- surface energy of water greater than surface energy of surface
- water “beads up”
- “greasy” surface
- high contact angle “ $\theta$ ”

- “wettable” surface
- non-water repellent surface
- hydrophilic surface
- water more attracted to surface than itself
- surface energy of surface greater than surface energy of water
- water “spreads out”
- “non-greasy” surface
- low contact angle “ $\theta$ ”

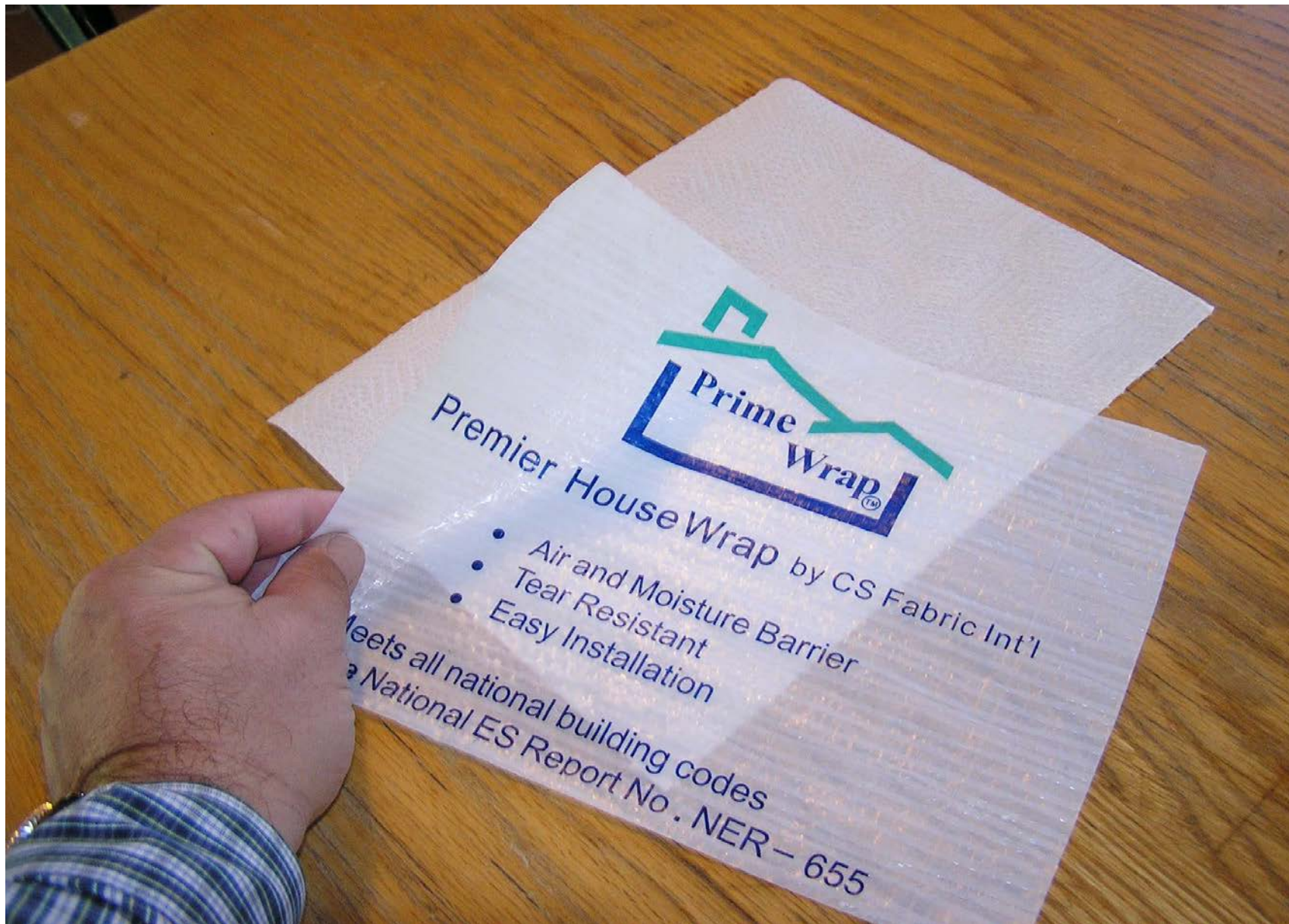


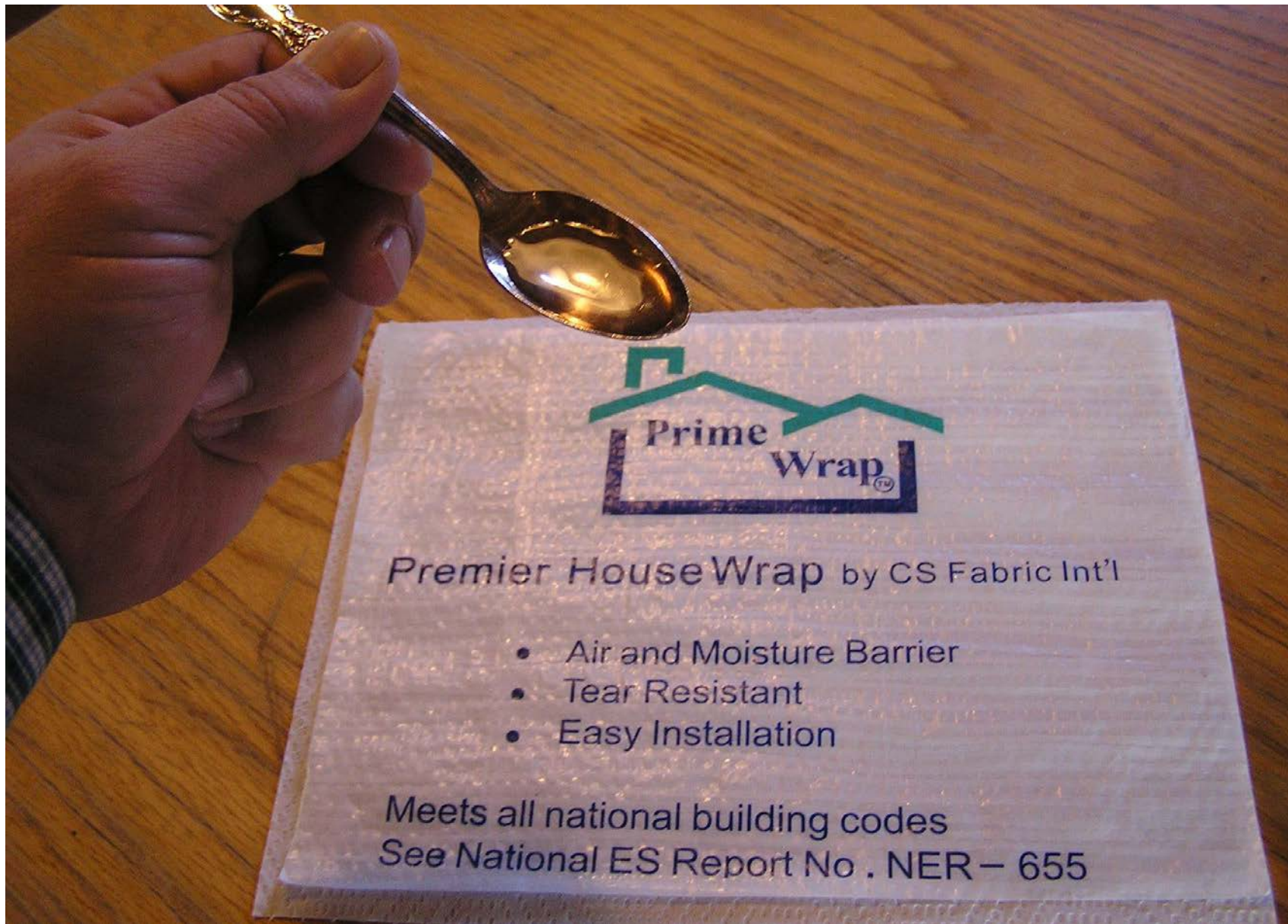












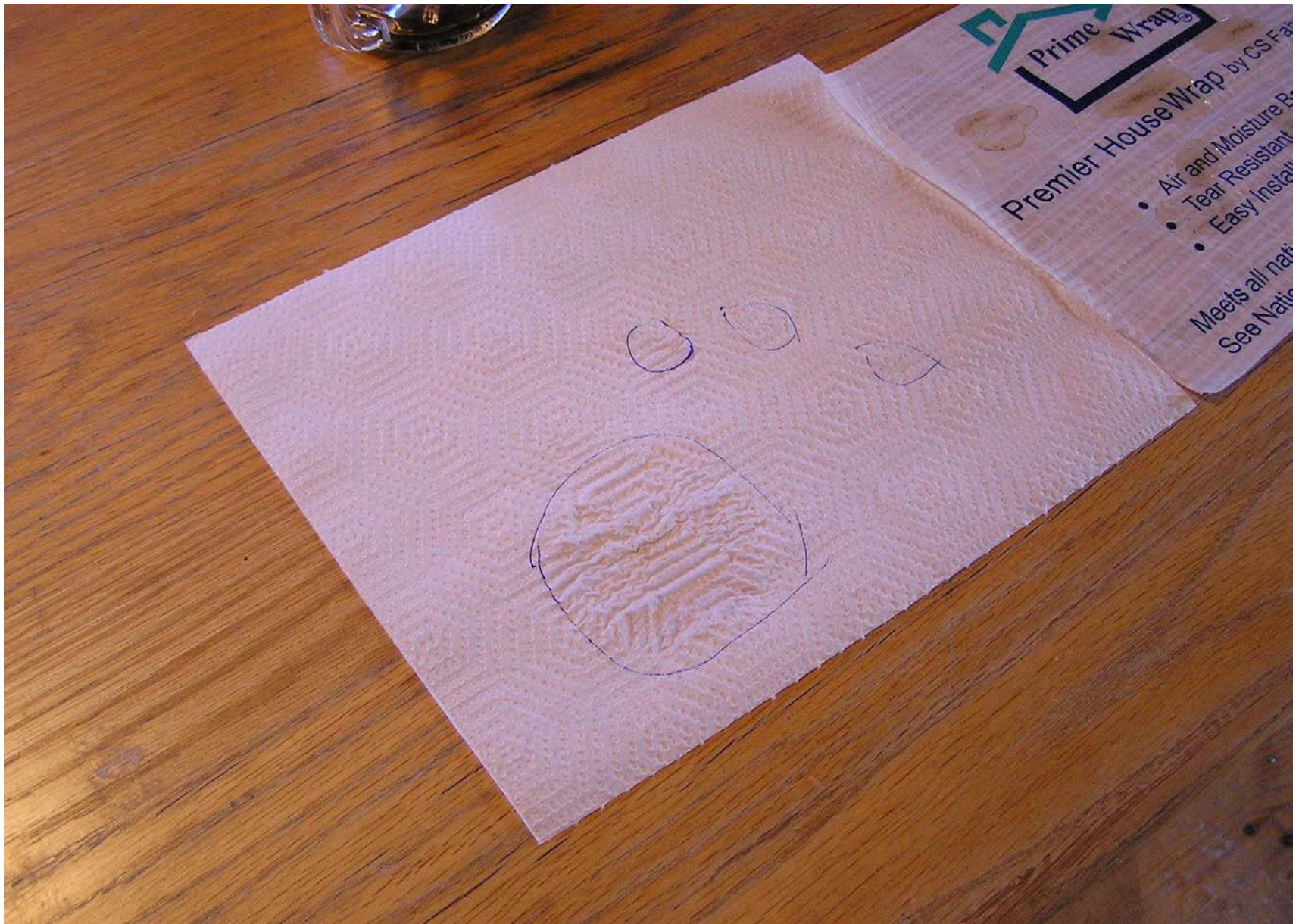














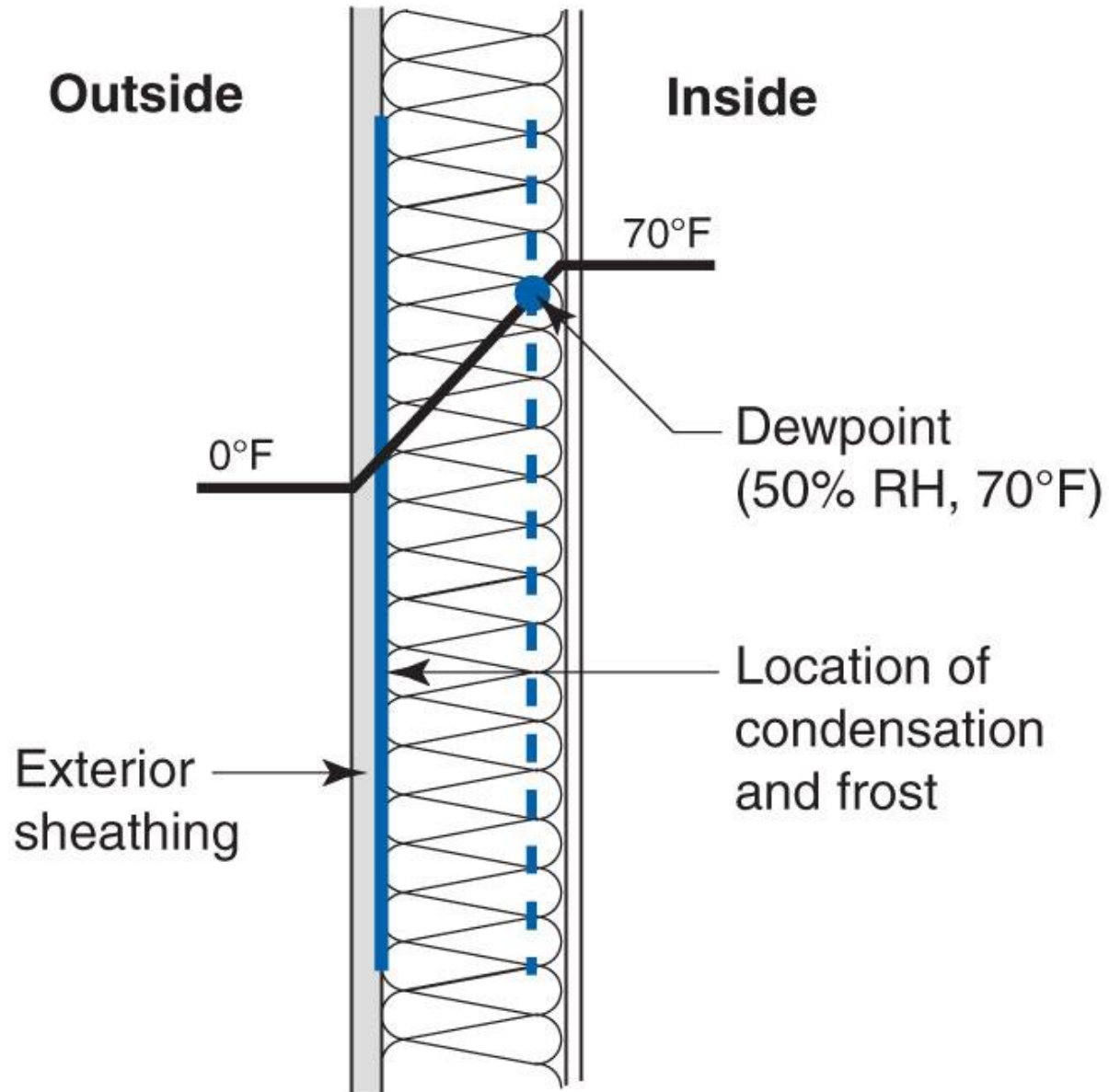




## Surface Energy

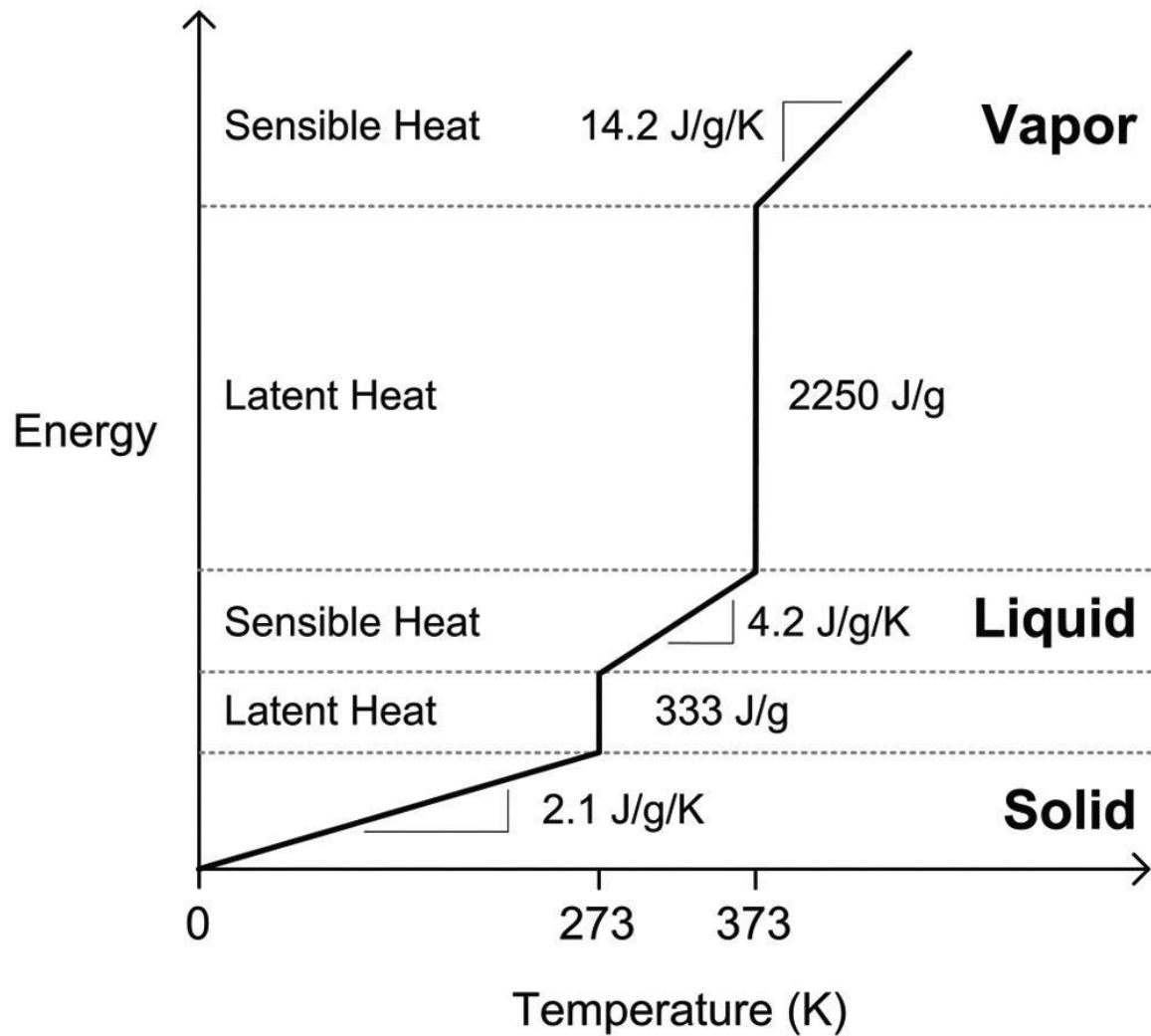
Water (20 C)	73 dynes/cm
Water (100 C)	59 dynes/cm
Epoxy	46 dynes/cm
Polyethylene	31 dynes/cm
Soapy water	30 dynes/cm
Paraffin wax	25 dynes/cm
Silicone	24 dynes/cm
Teflon	18 dynes/cm

# When Phases Change







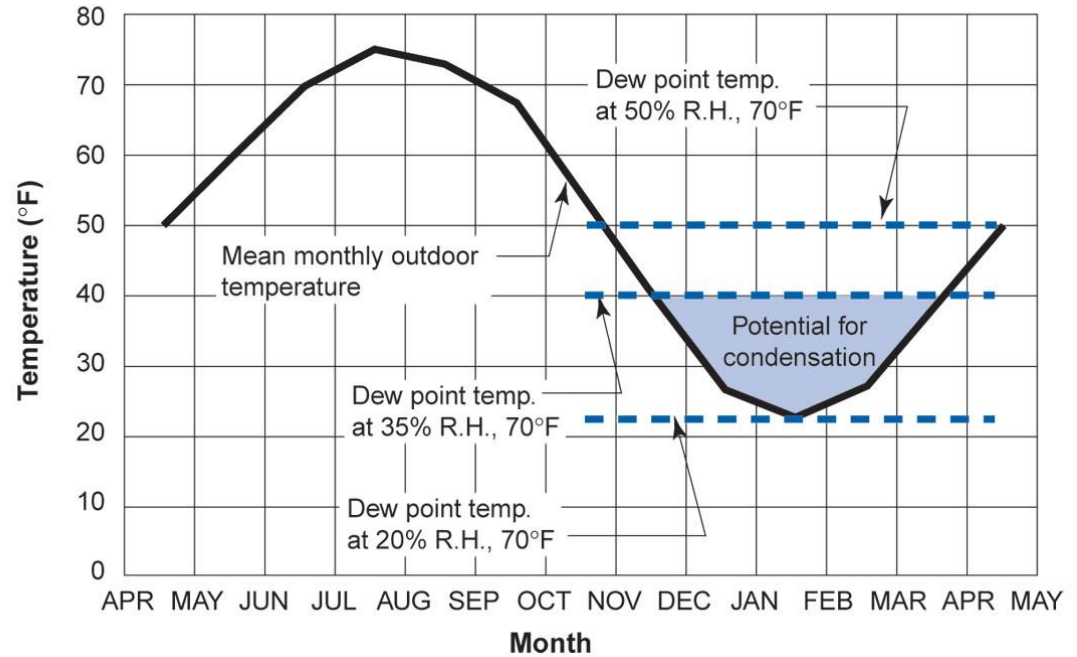
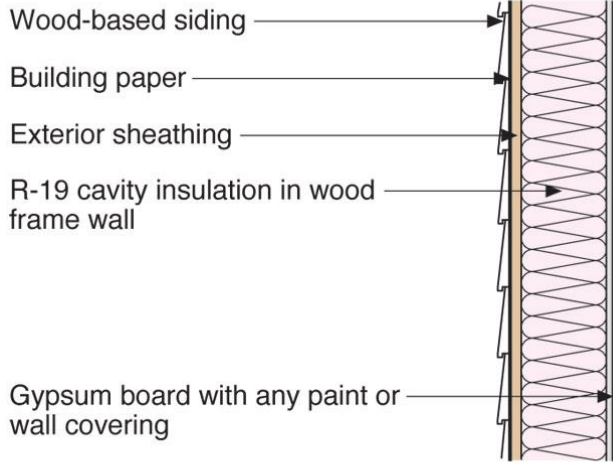


**Simple linearized energy-temperature relation for water**

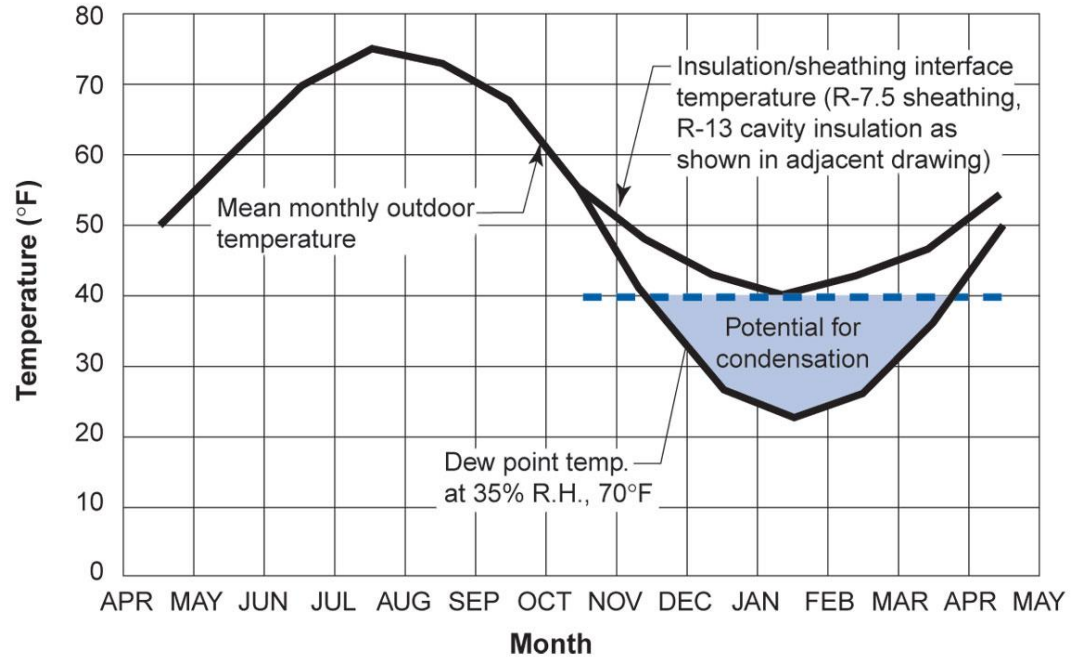
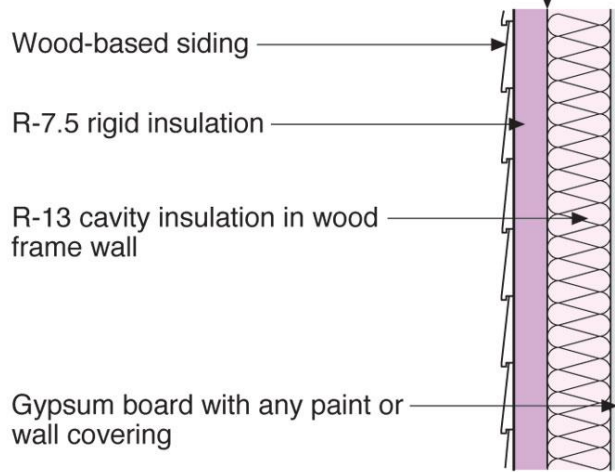
From Straube & Burnett, 2005



The inside face of the exterior sheathing is the condensing surface of interest



The inside face of the insulating sheathing is the condensing surface of interest



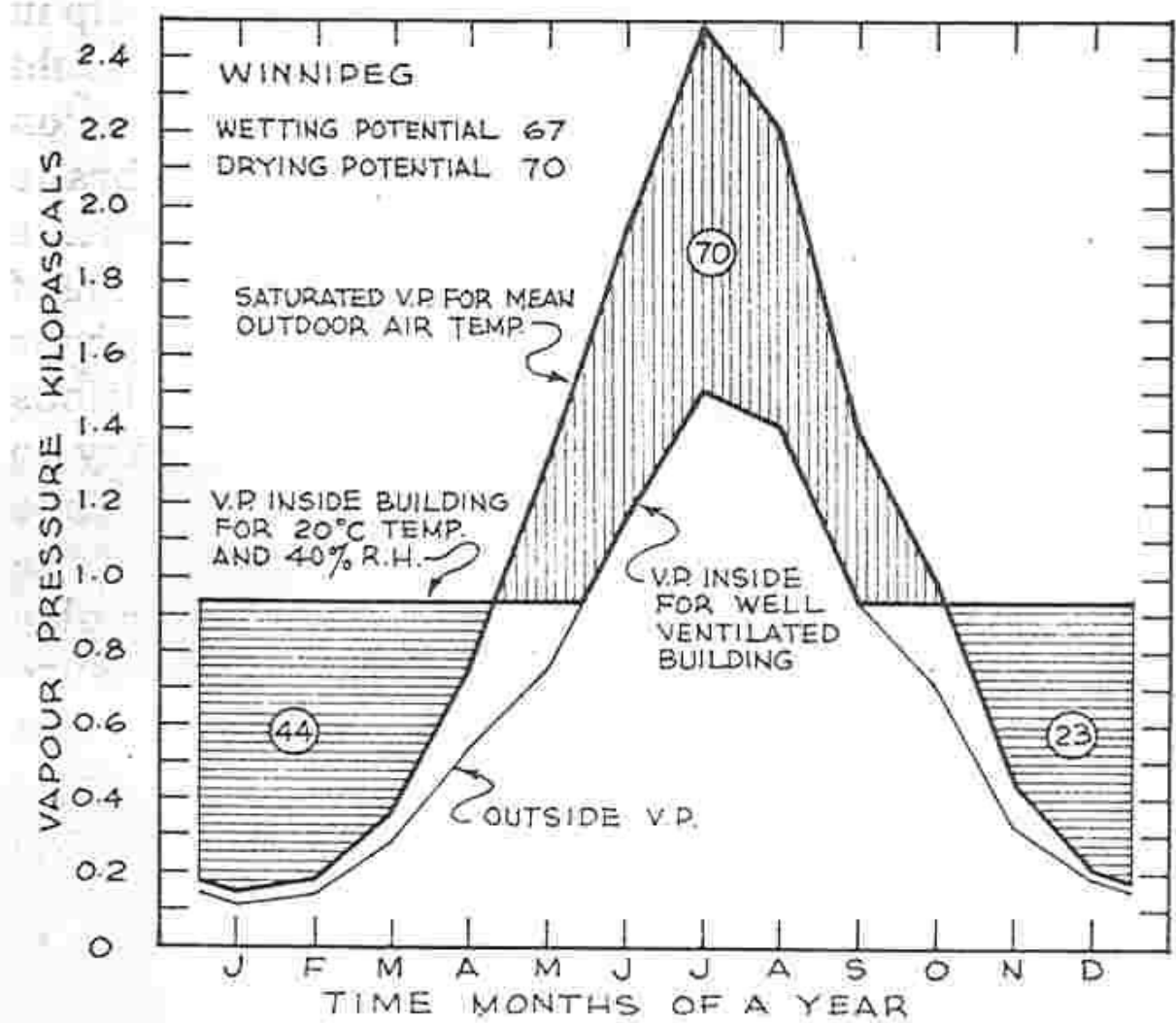
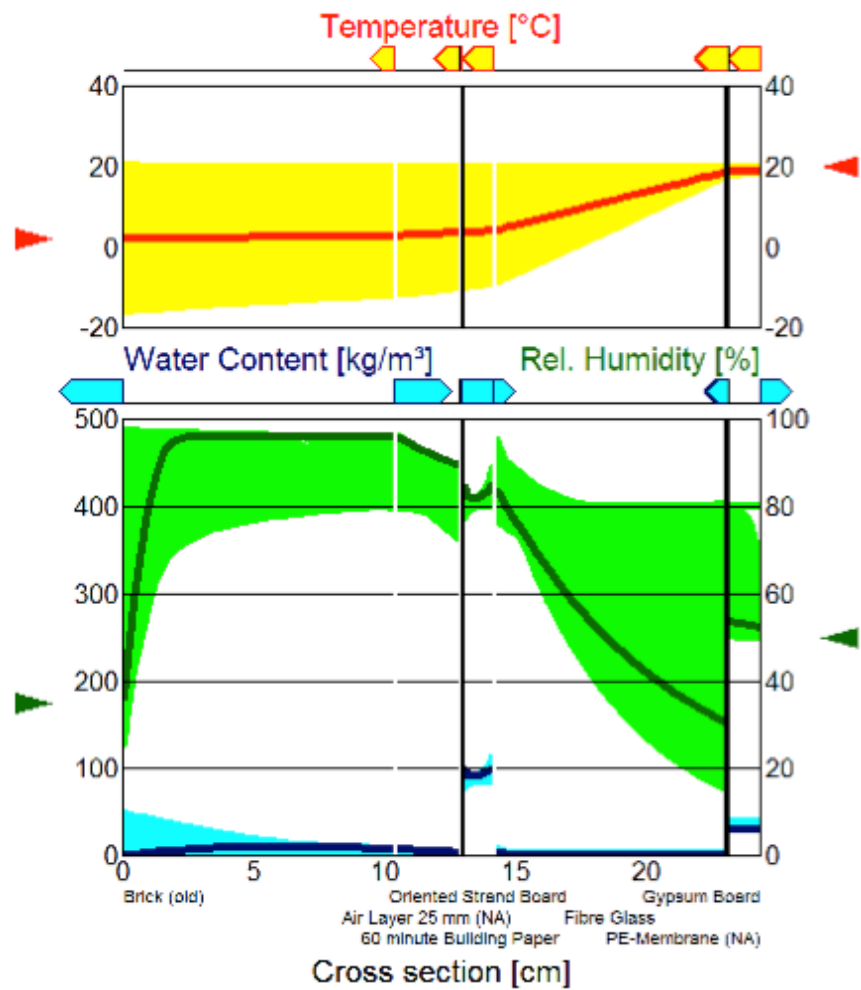


Figure 8-7. Outside vapour pressure, saturated vapour pressure and inside vapour pressure for Winnipeg.



WUFI® 3.3 Pro. IBP  
Run

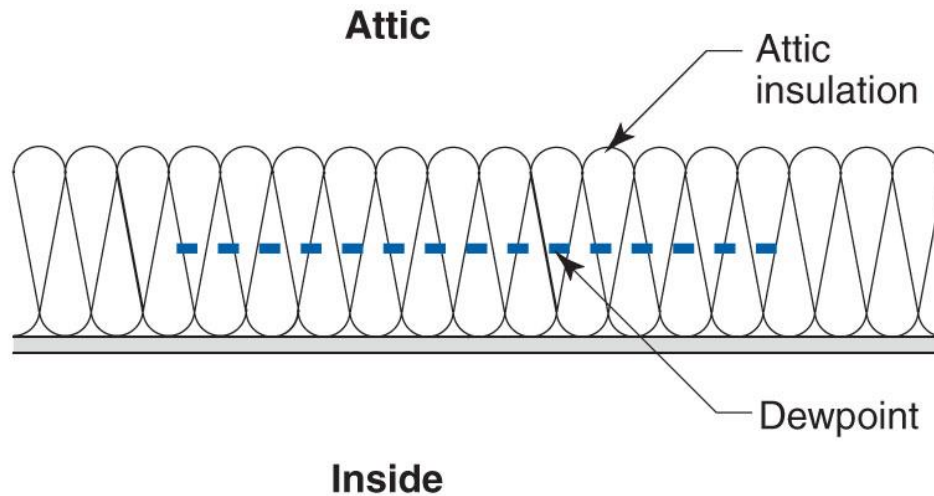
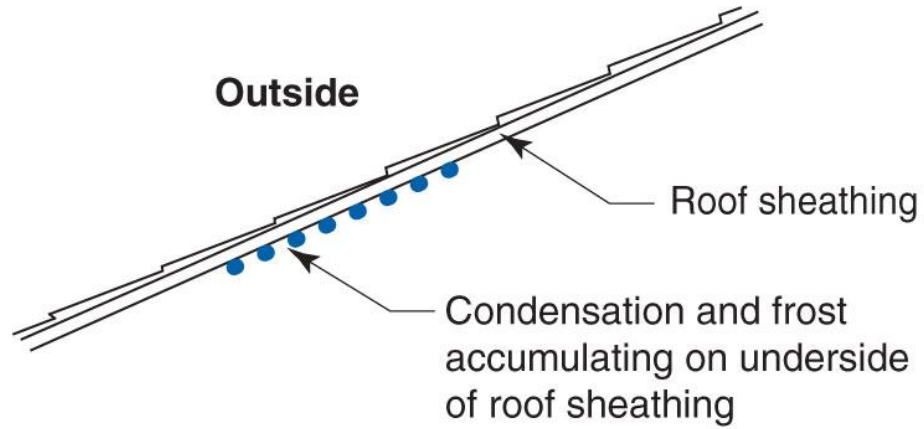
16 Feb  
2001

100%

100%

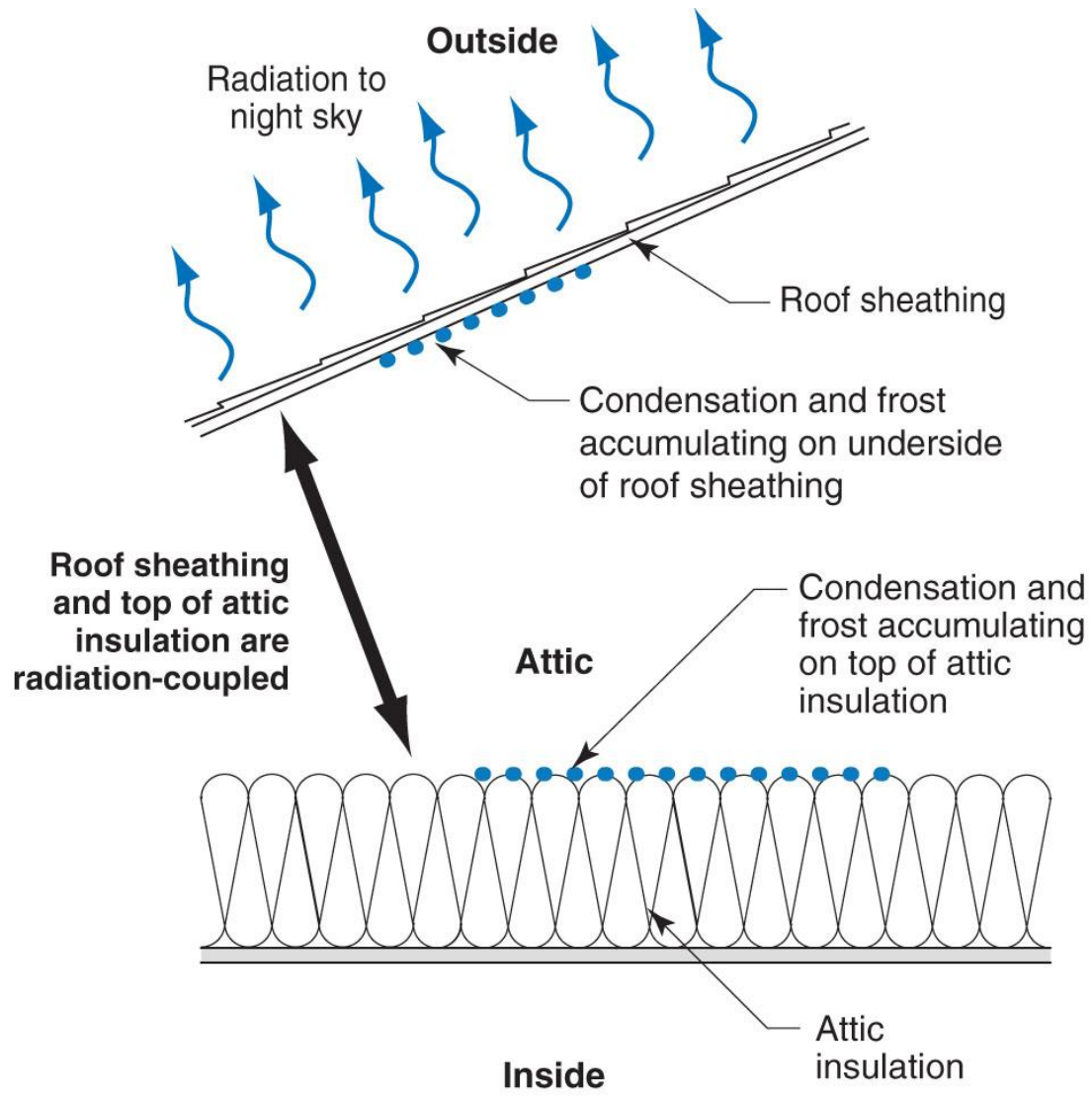
100%

100%





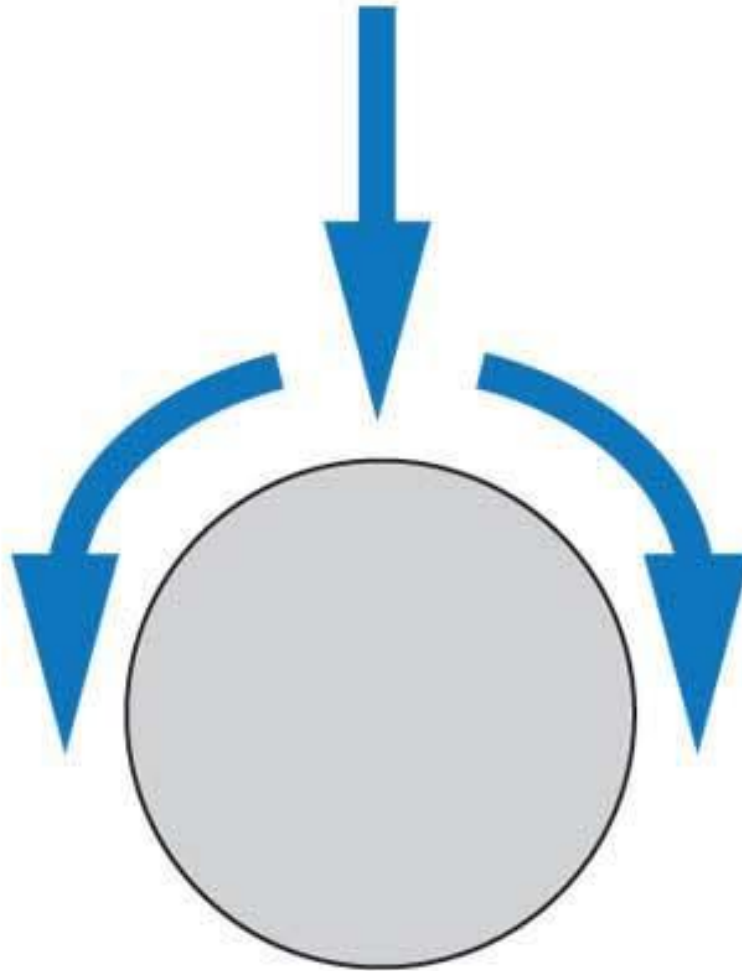


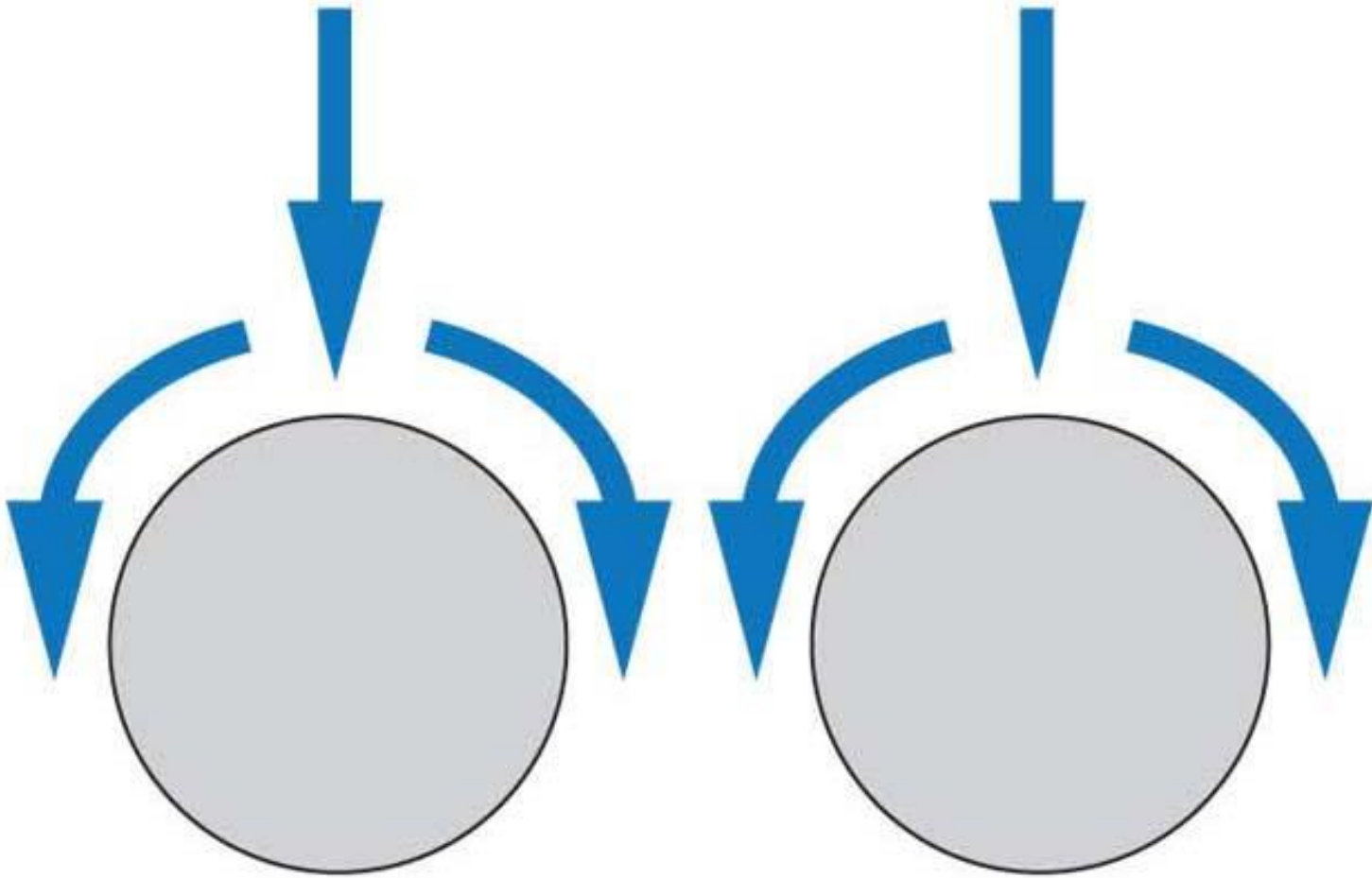


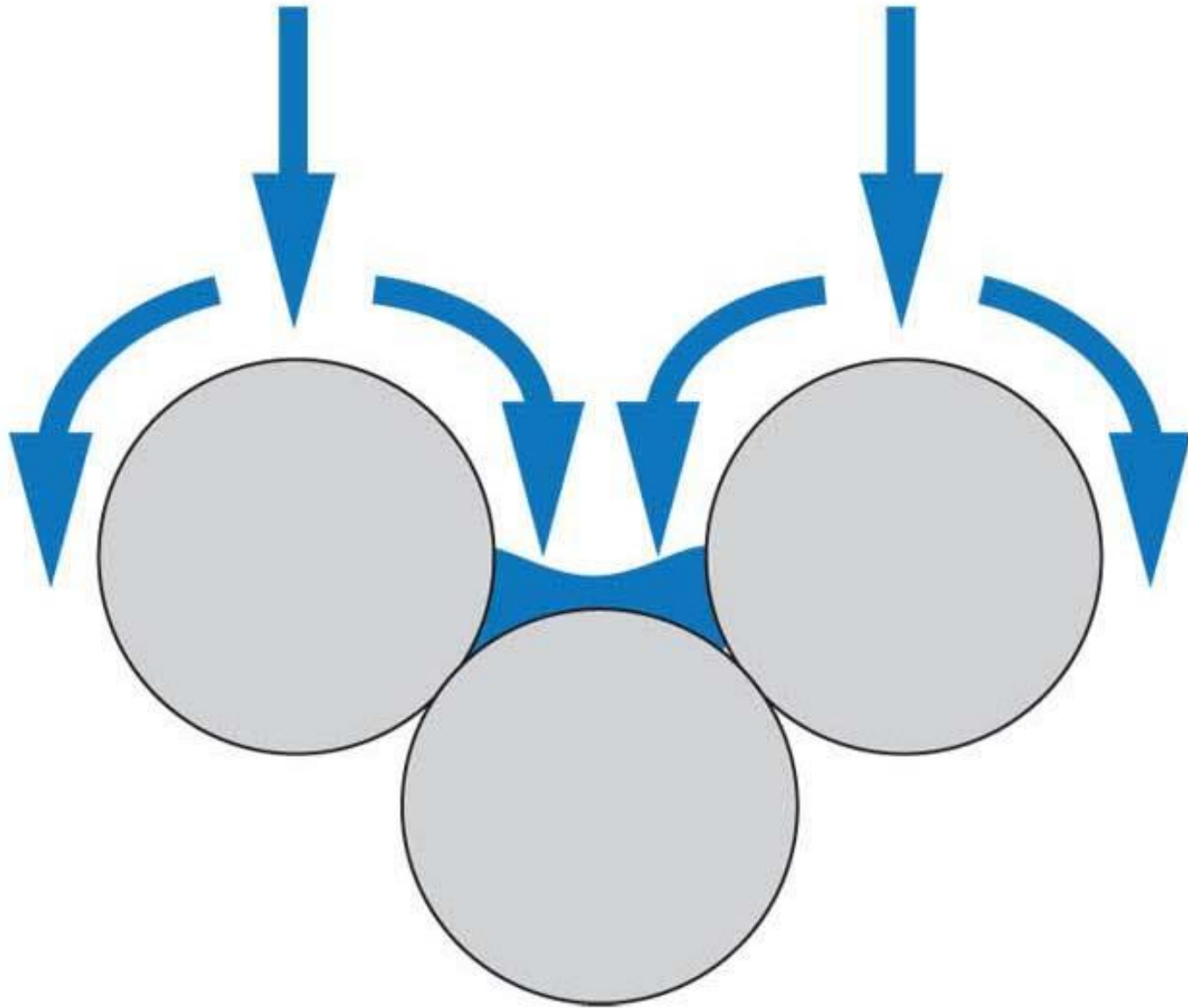


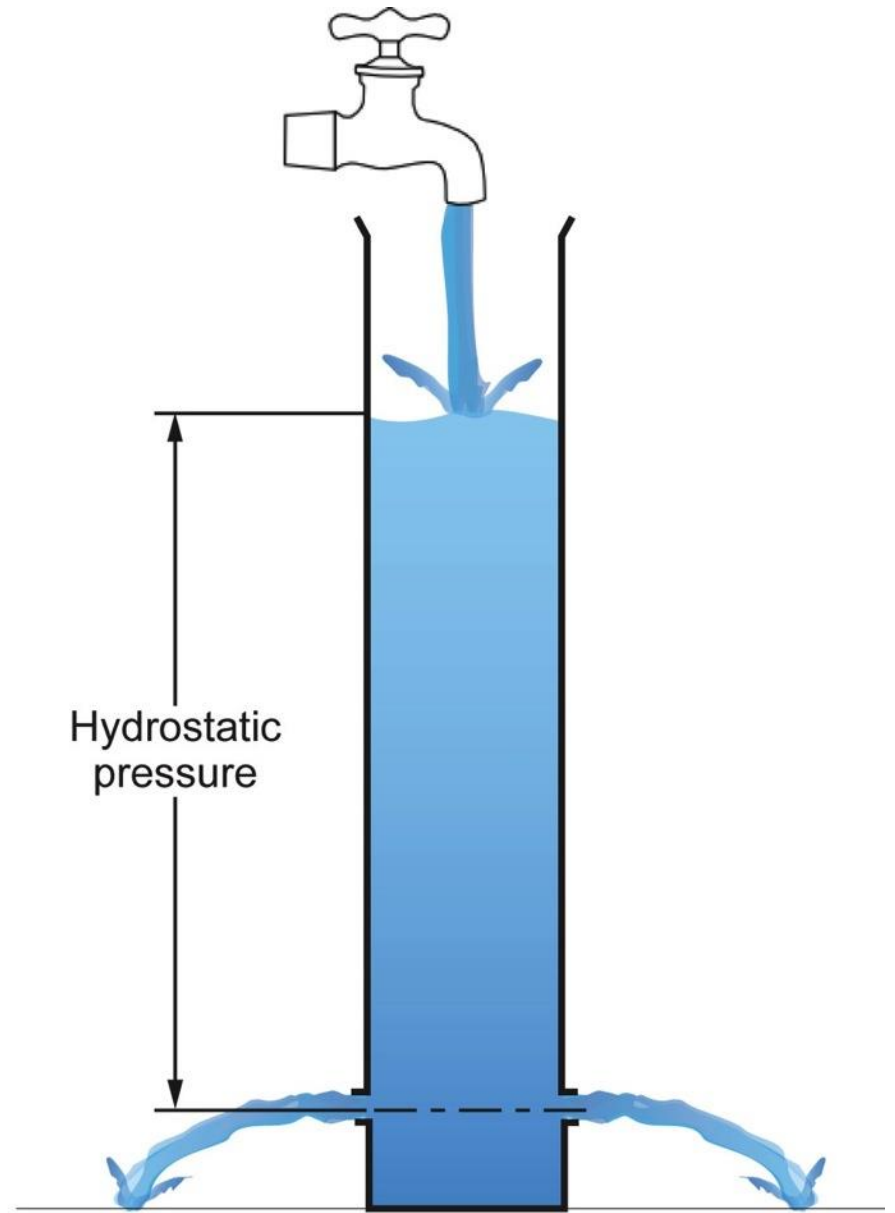
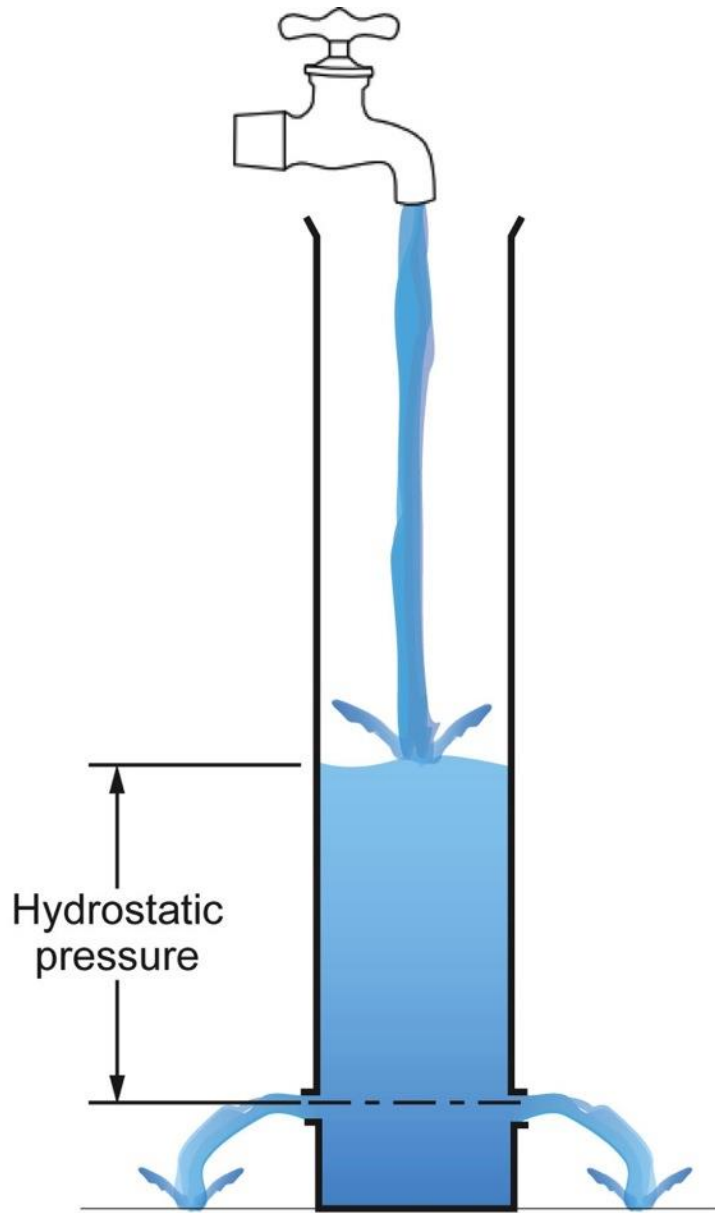




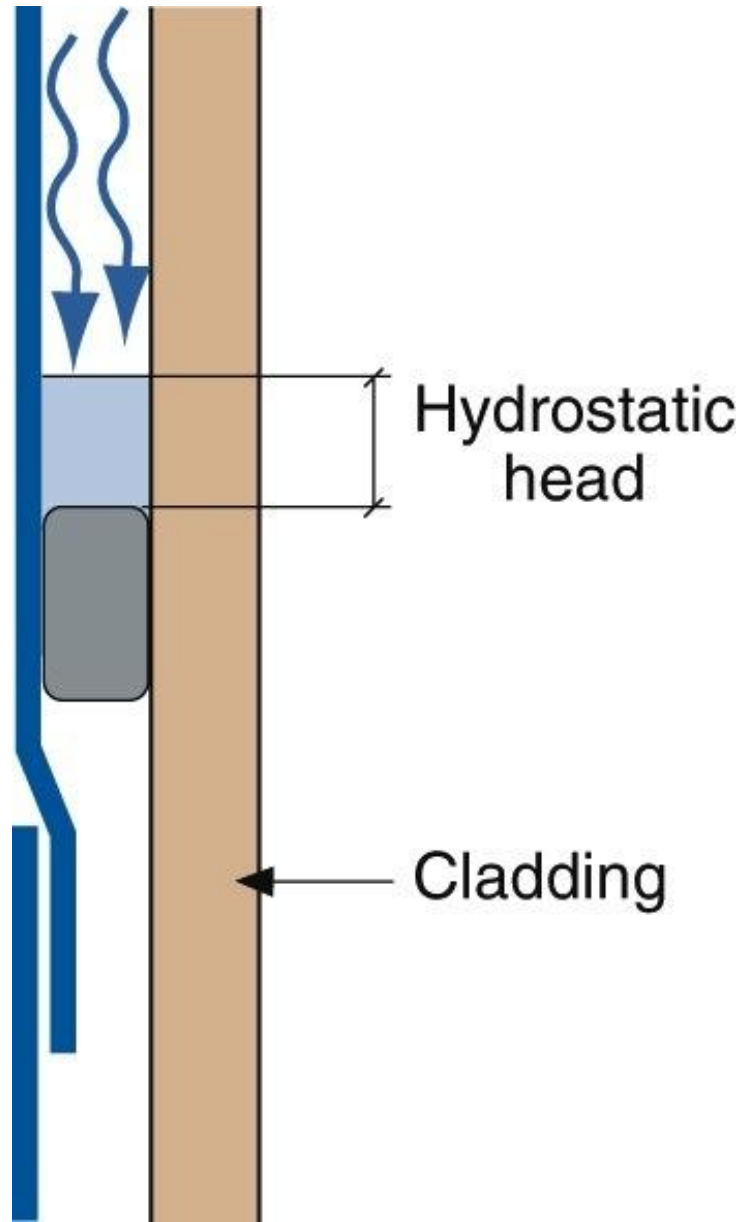


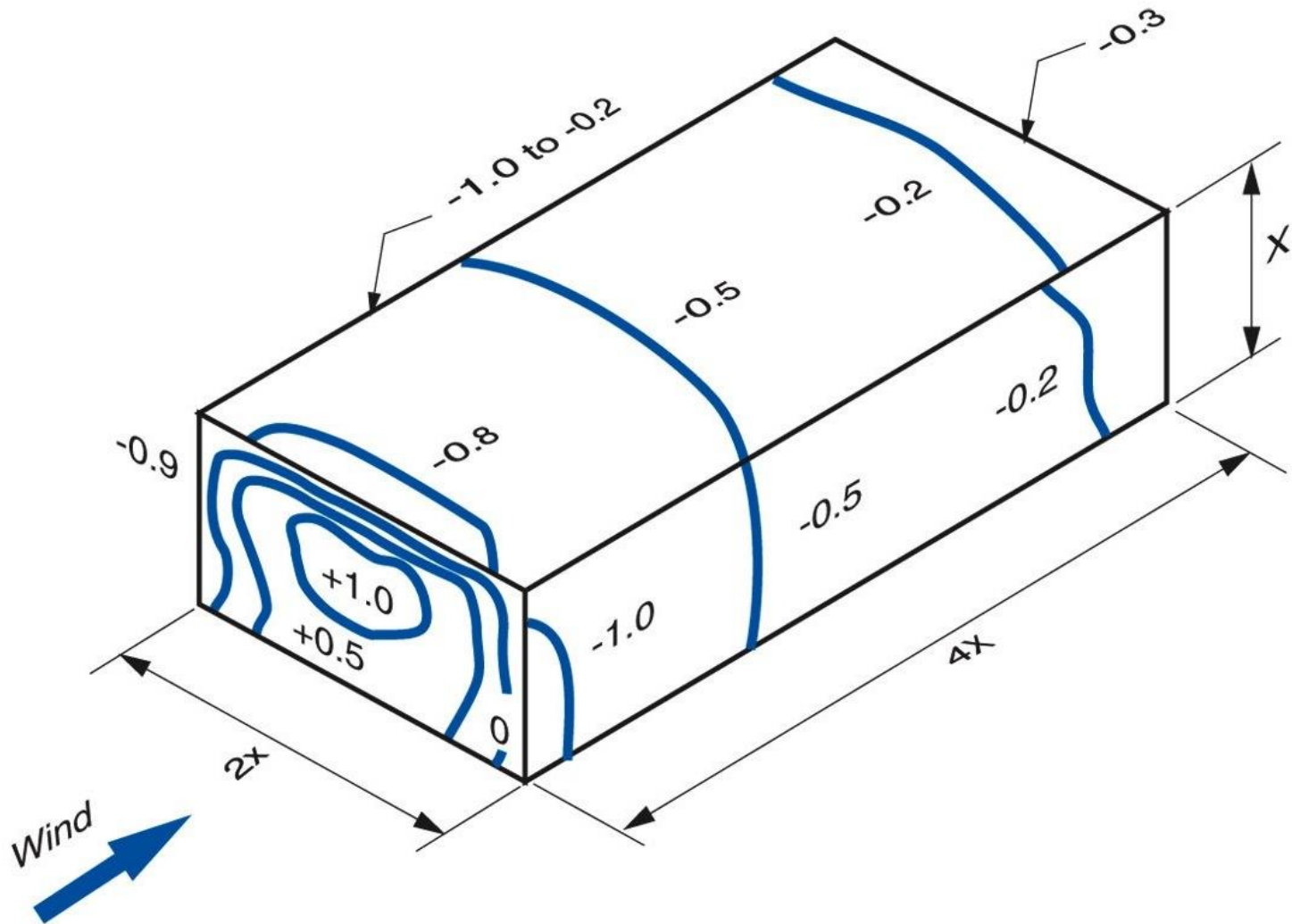








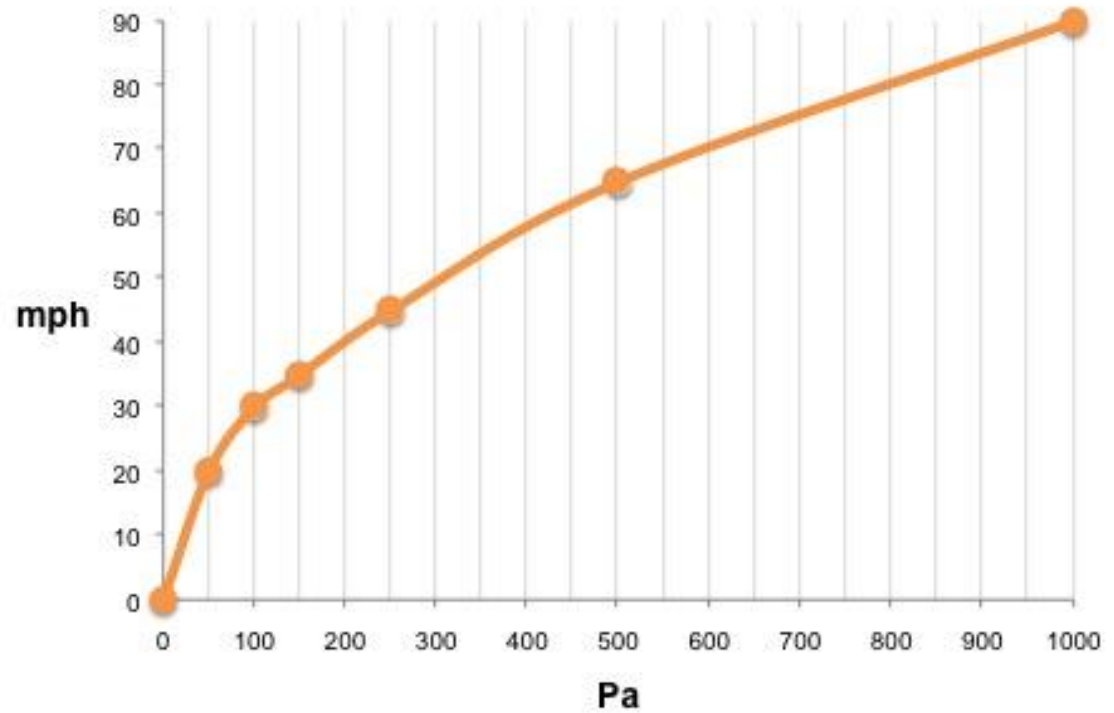




### Pascals    mph

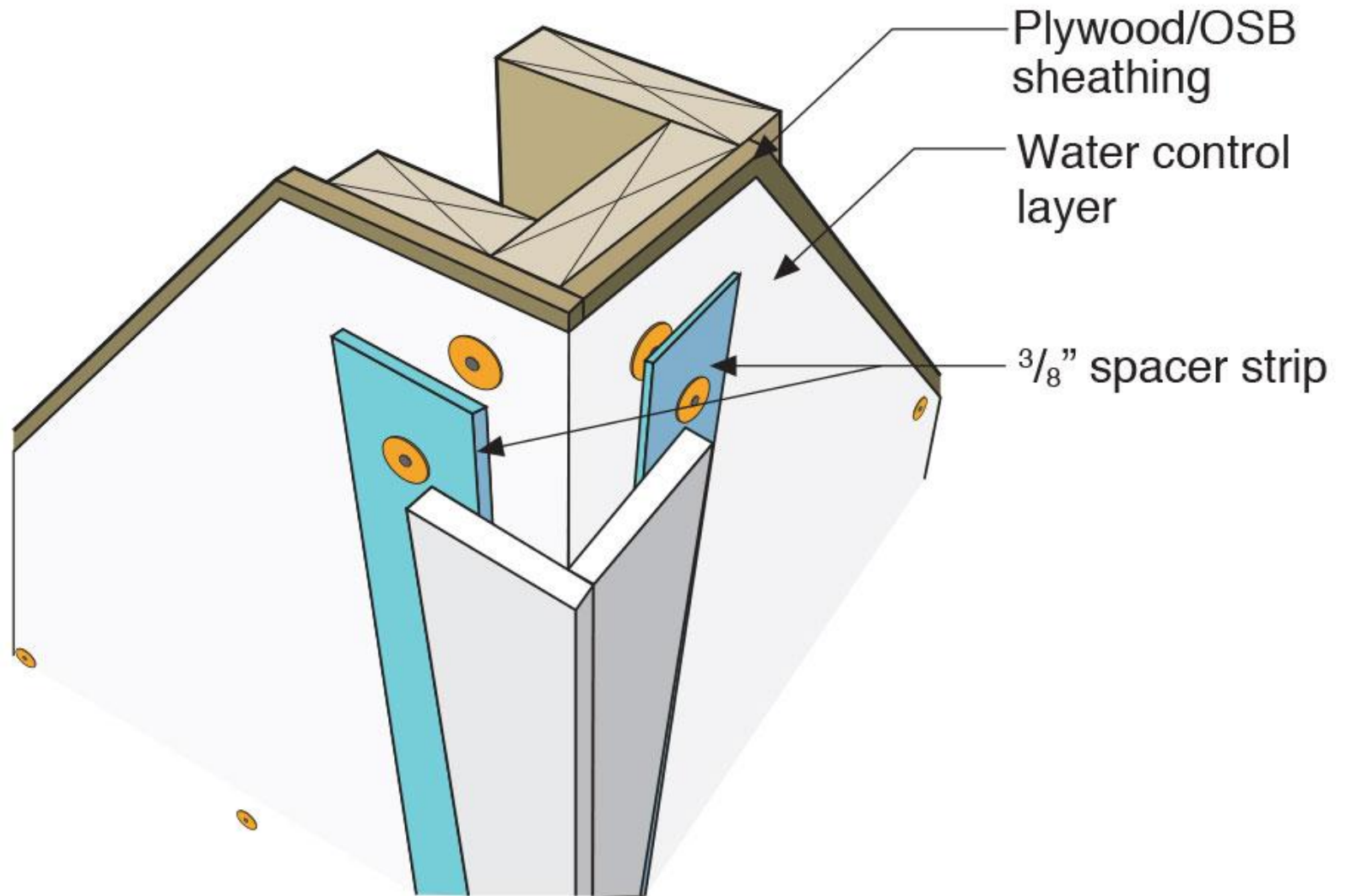
50	Pa =	20	mph
100	Pa =	30	mph
150	Pa =	35	mph
250	Pa =	45	mph
500	Pa =	65	mph
1,000	Pa =	90	mph

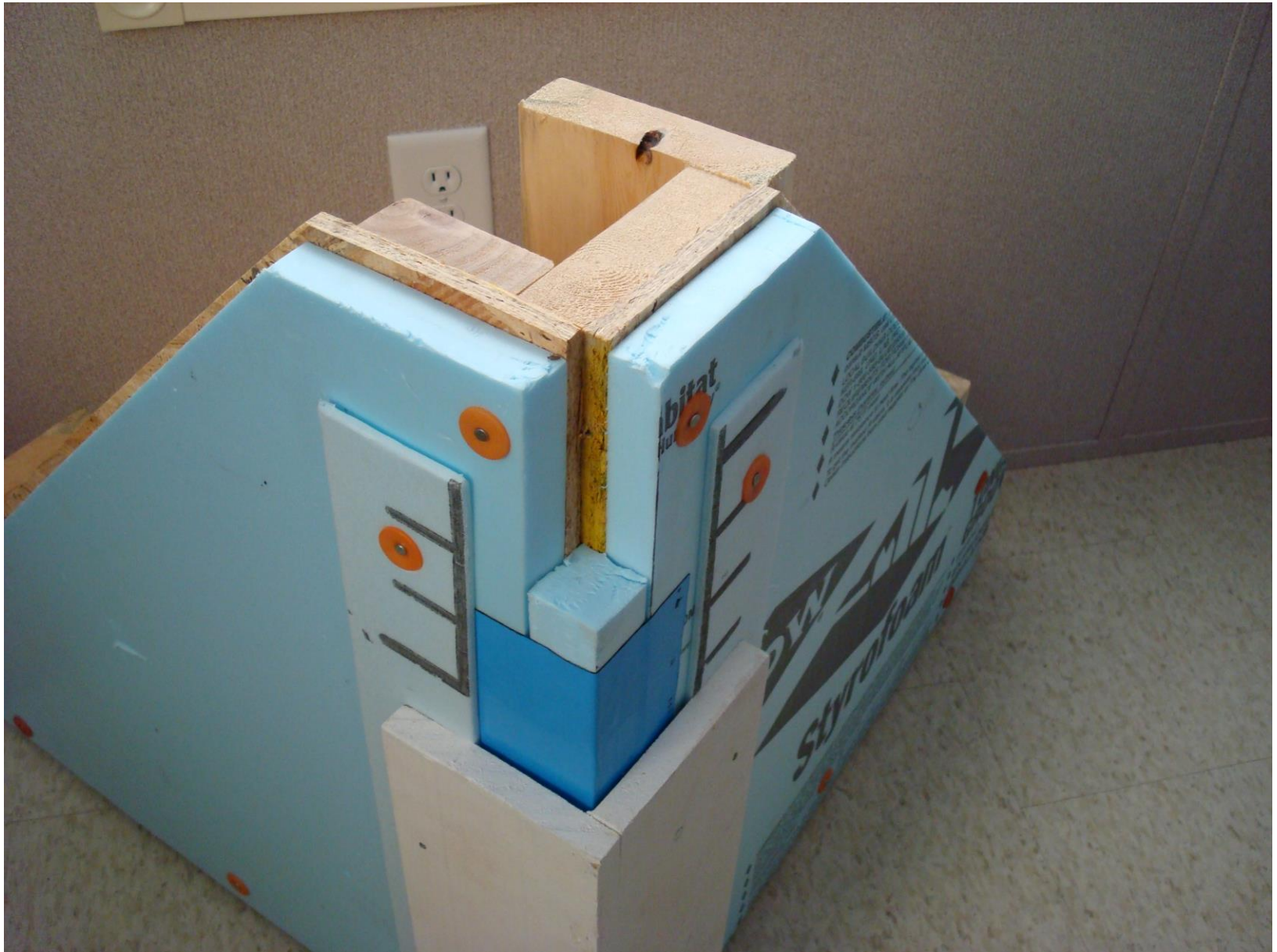
### Wind Speed (mph) vs. Stagnation Pressure (Pa)





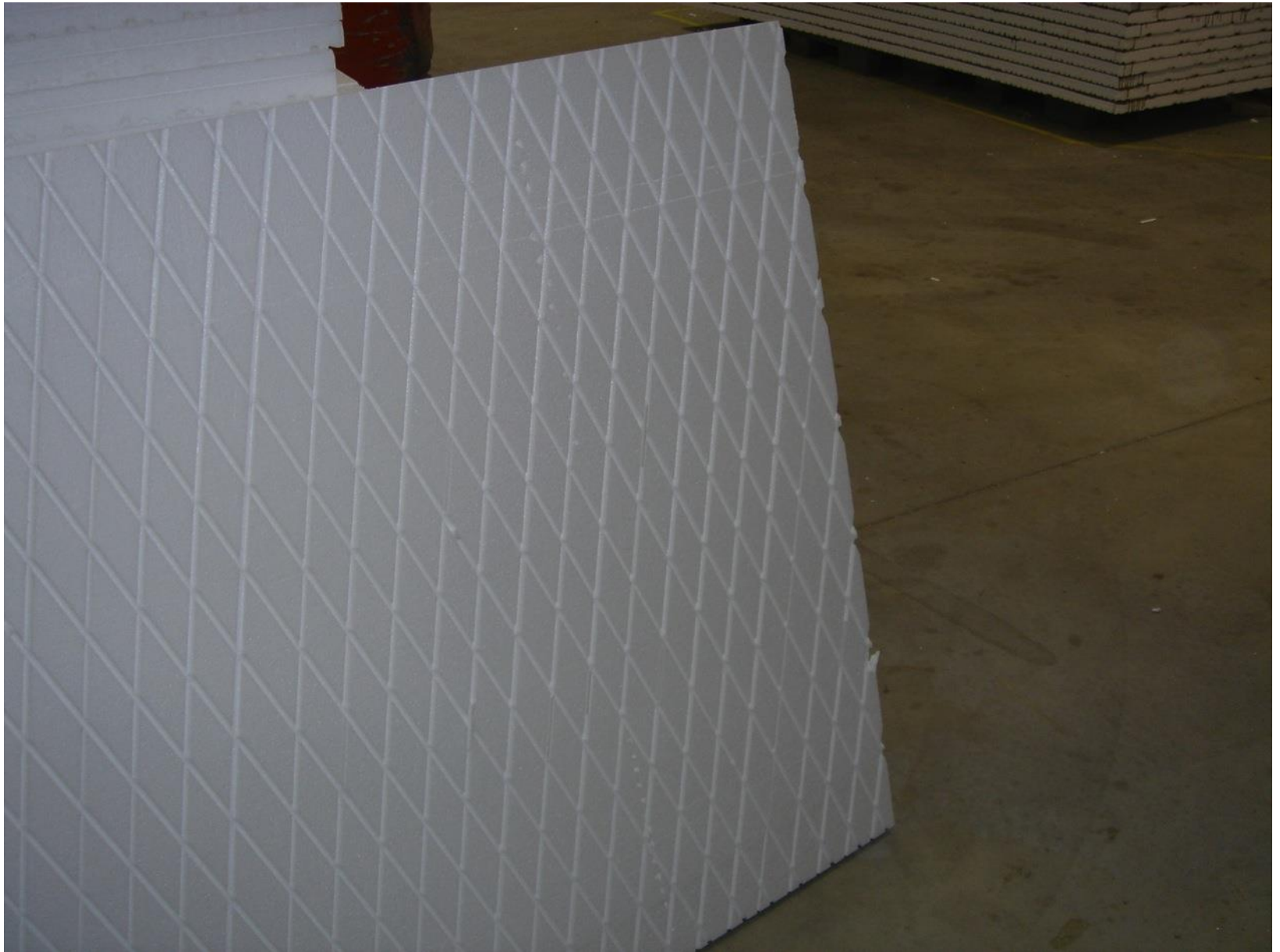




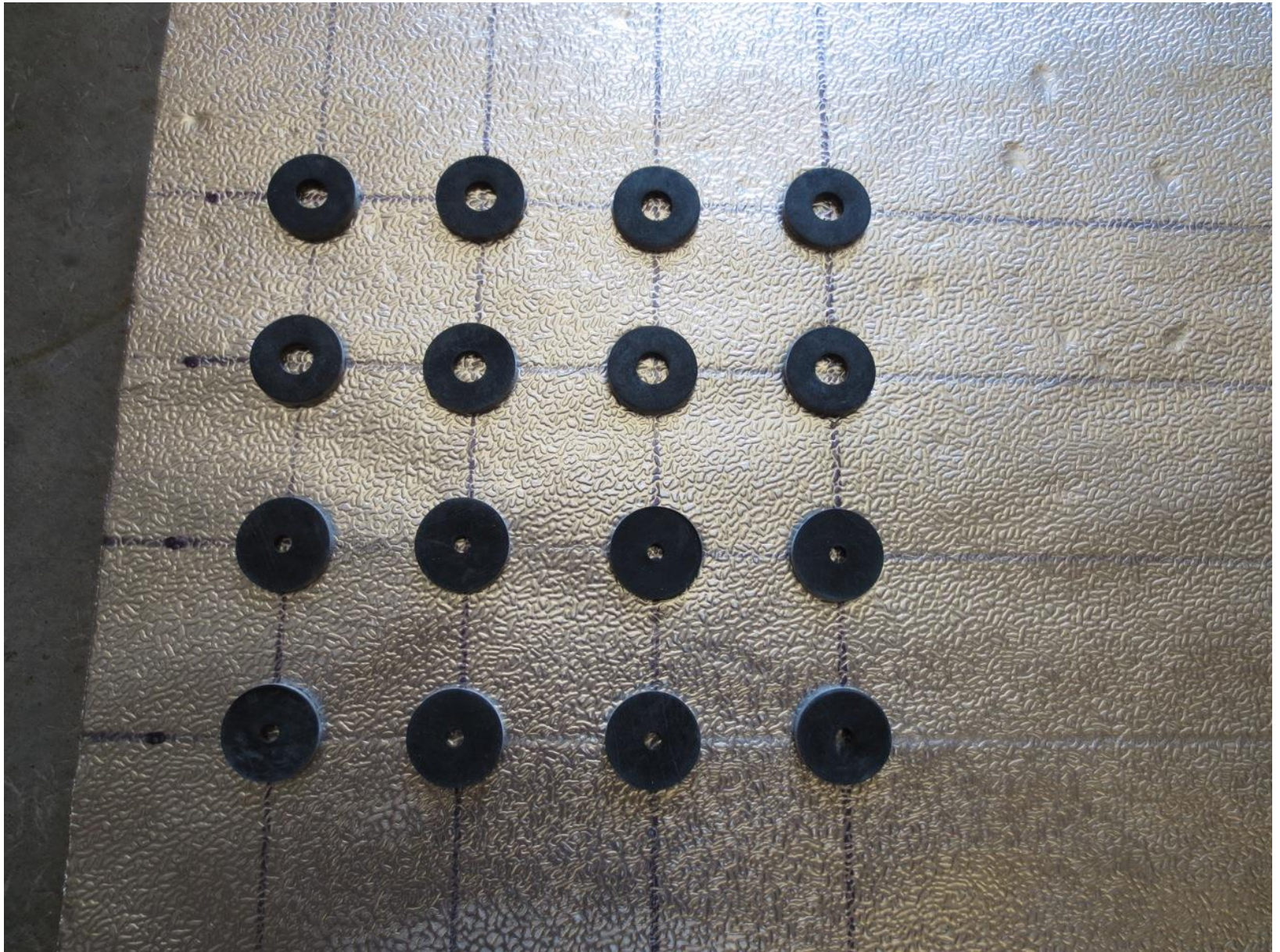








# Rain Screen



# Beer Screen?

