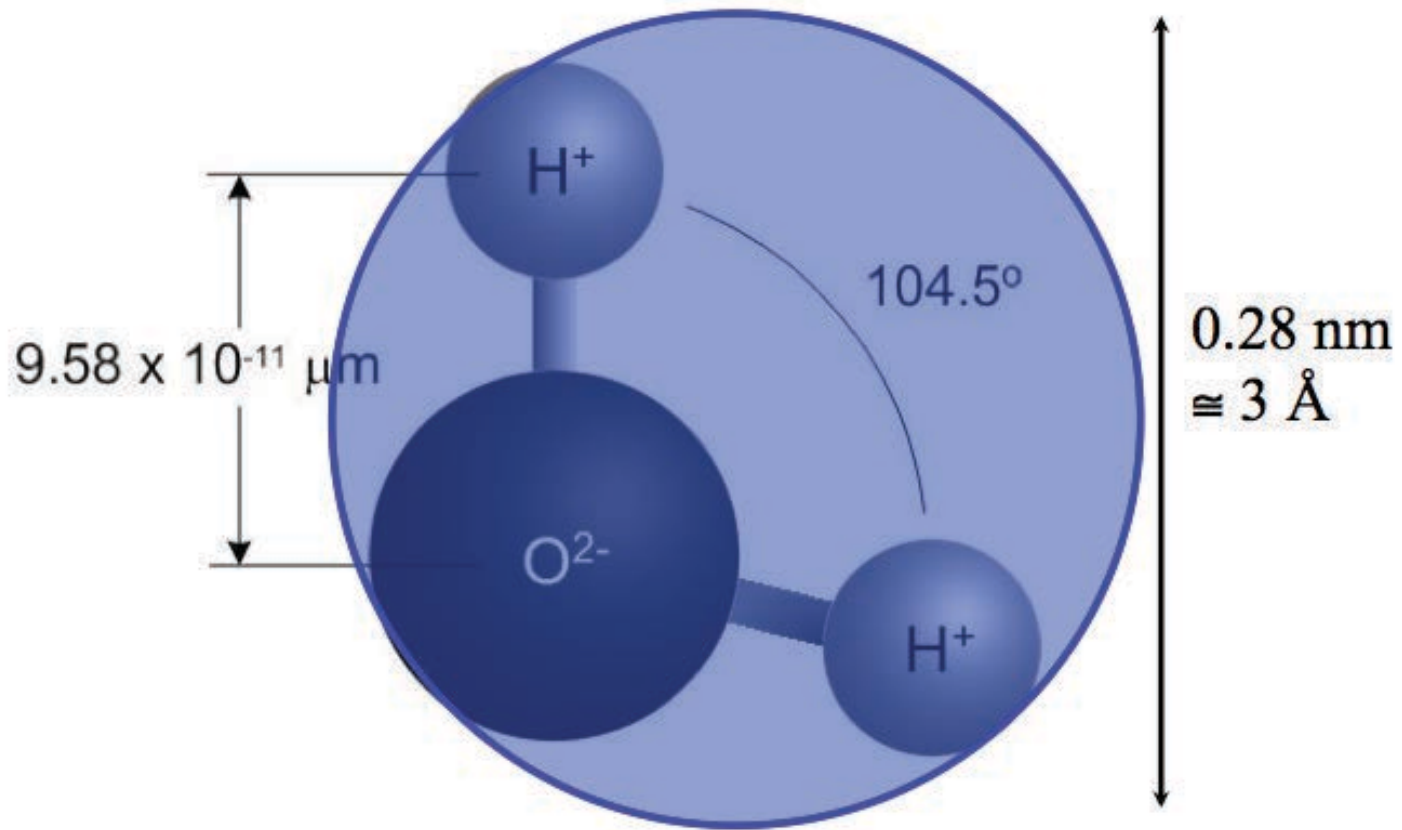


Joseph Lstiburek, Ph.D., P.Eng, ASHRAE Fellow

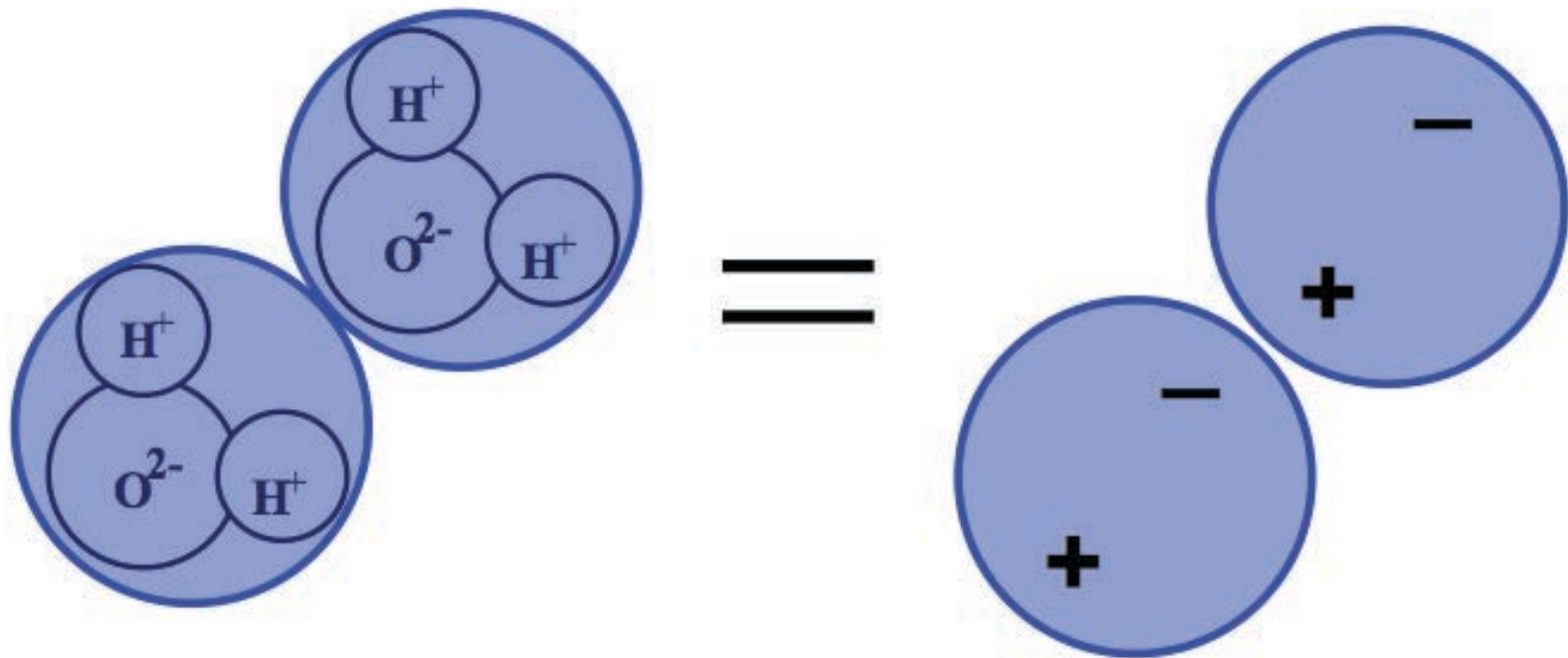
Building Science

The Water Molecule

Water Molecules



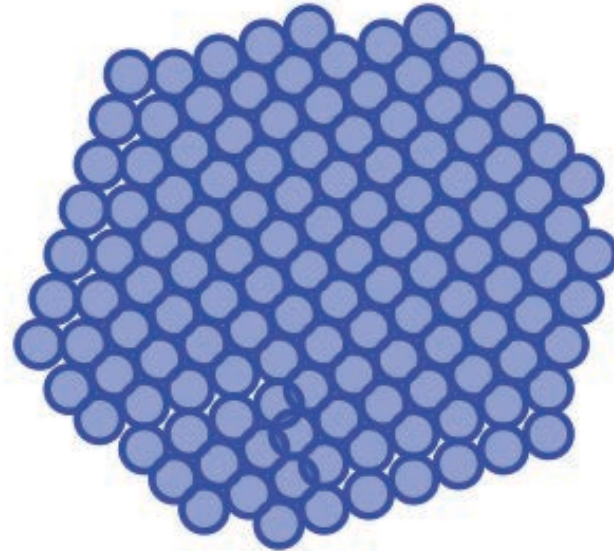
Polar Molecule



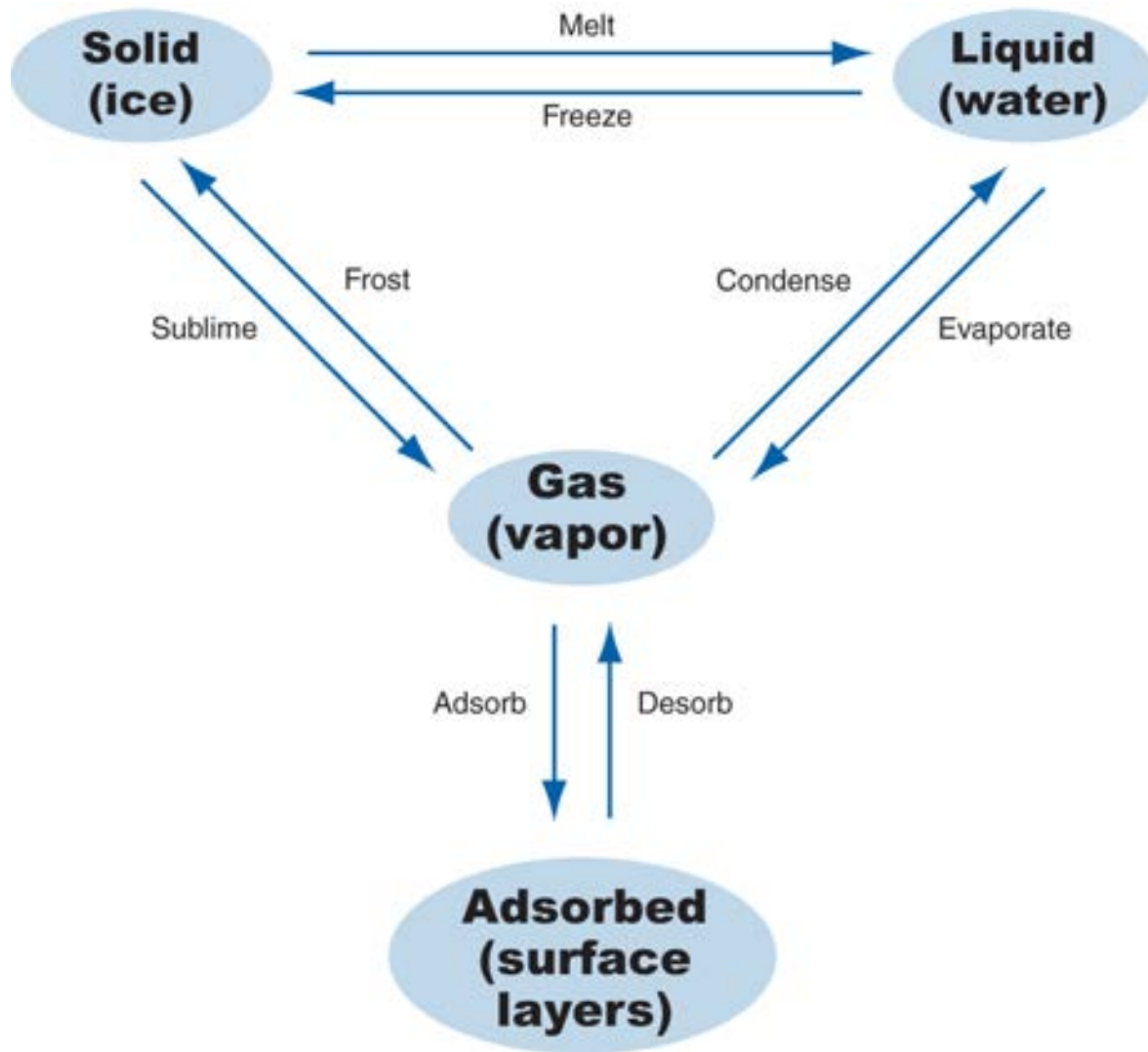
Size Matters

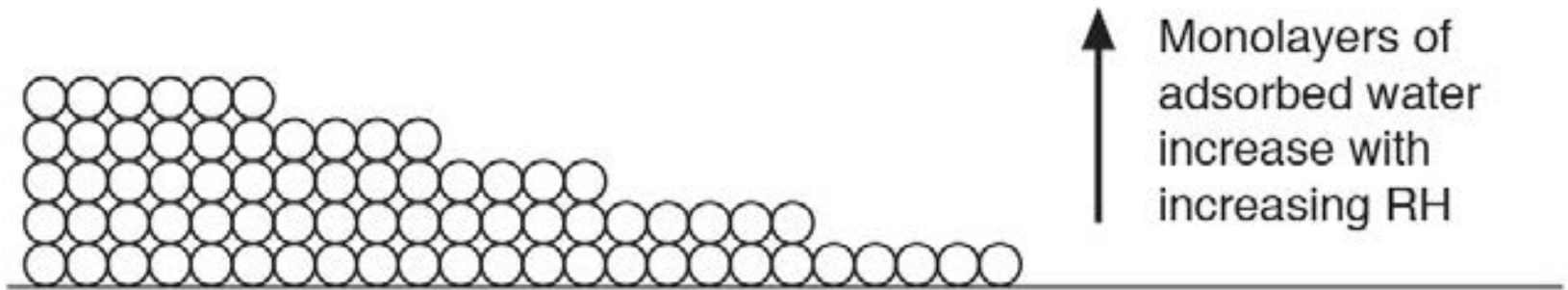


Vapor



Liquid







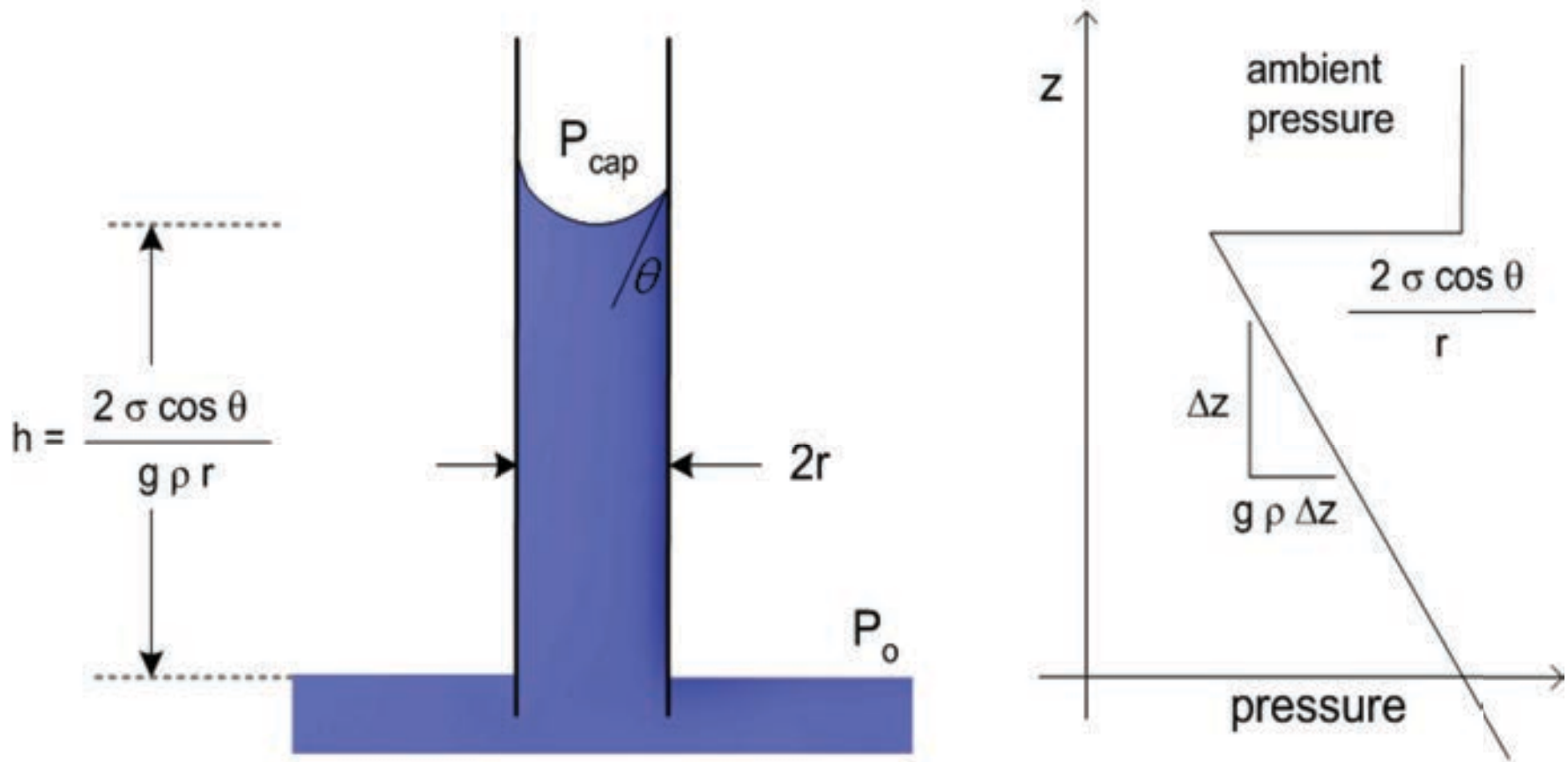
William Thomson

William Thomson – Lord Kelvin

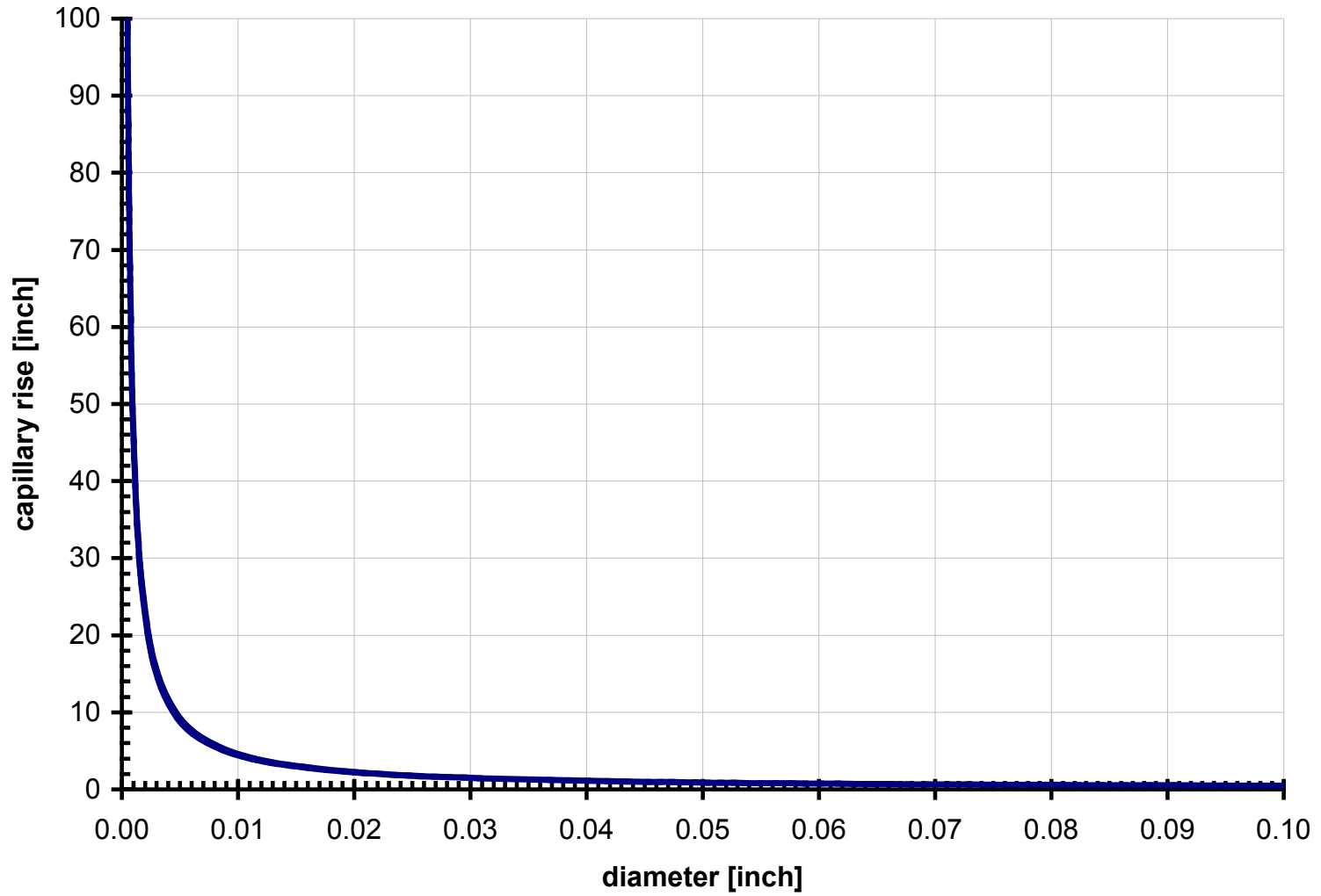
Kelvin Equation

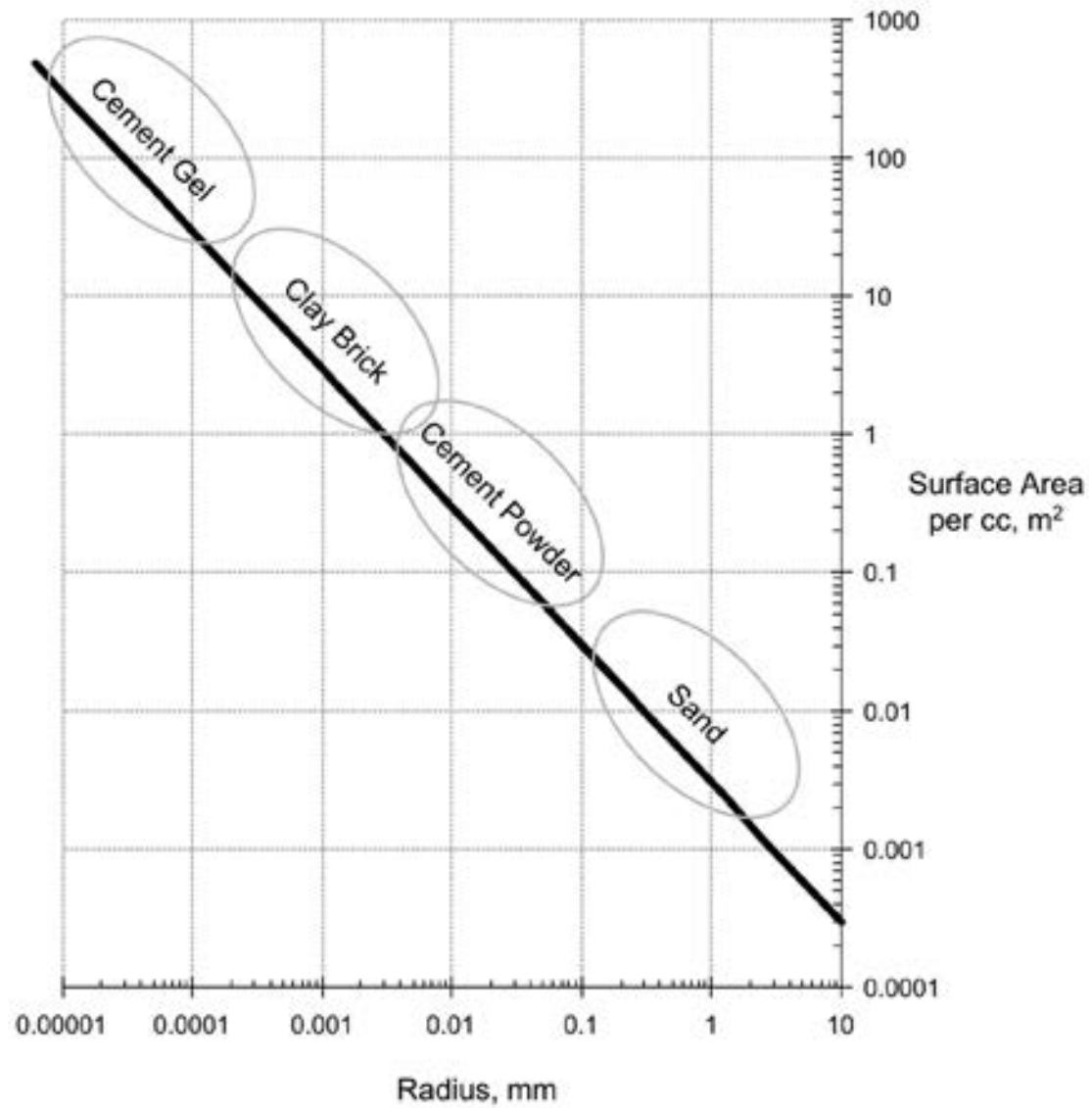
$$\ln \frac{p}{p_0} = \frac{2\gamma V_m}{rRT}$$

Calculating capillary rise



Capillary rise versus diameter





Surface area vs. particle size
From Straube & Burnett, 2005

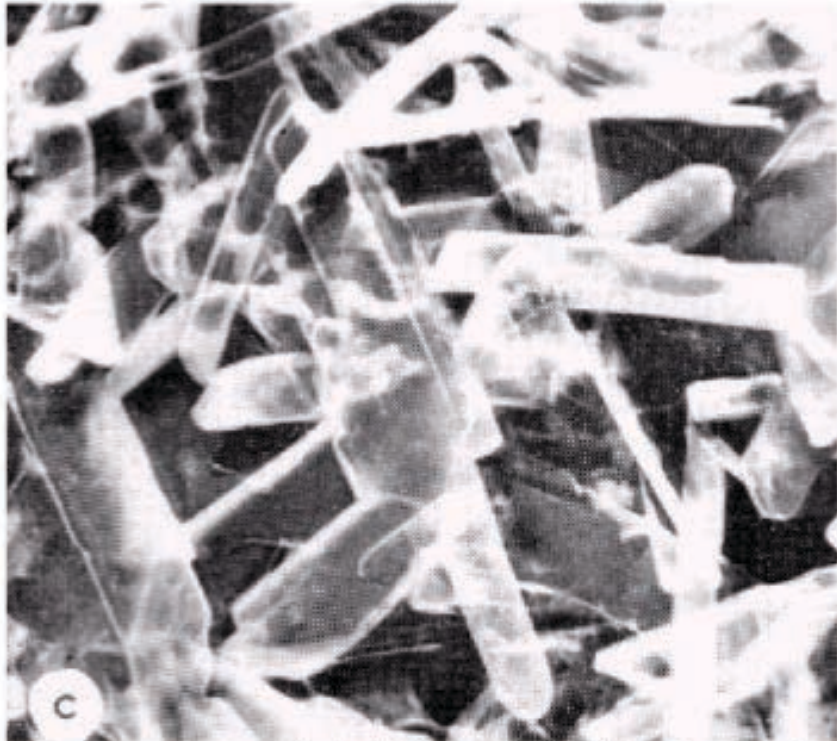


Figure 1c. Gypsum, hydrated from plaster of paris and water, porosity 30 per cent.

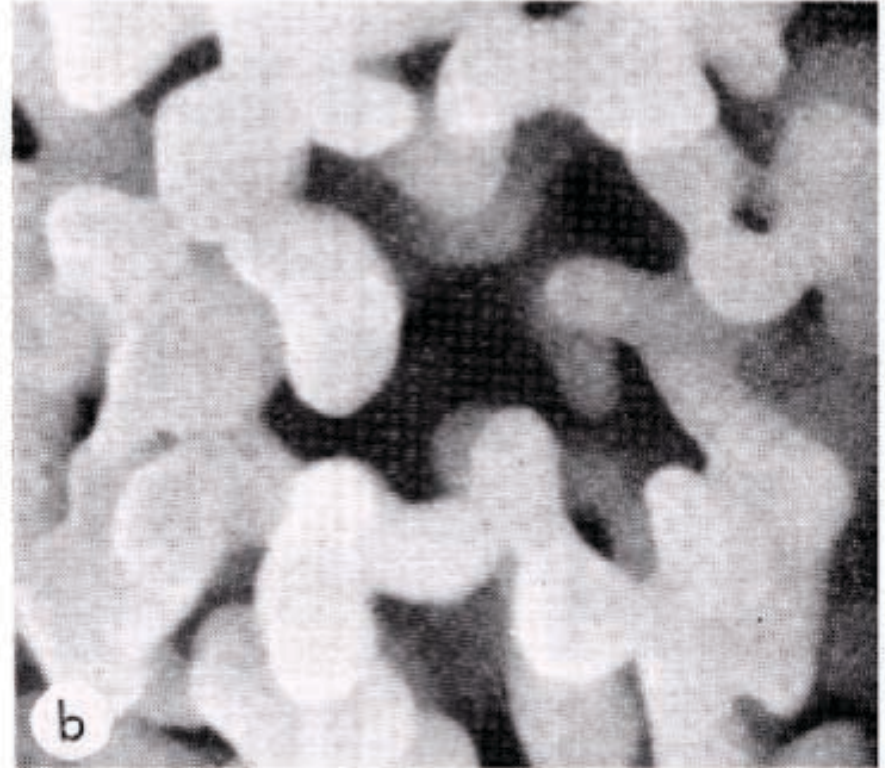
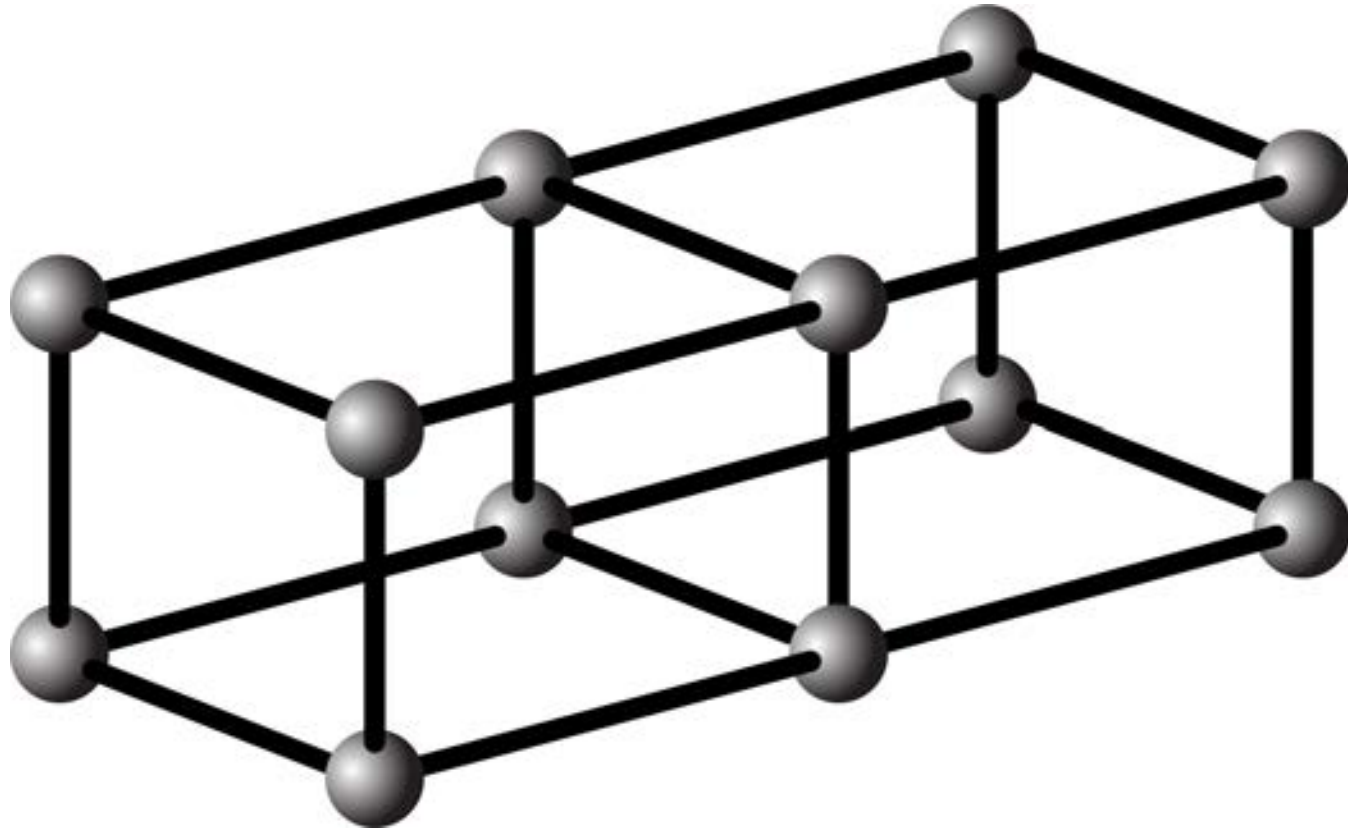
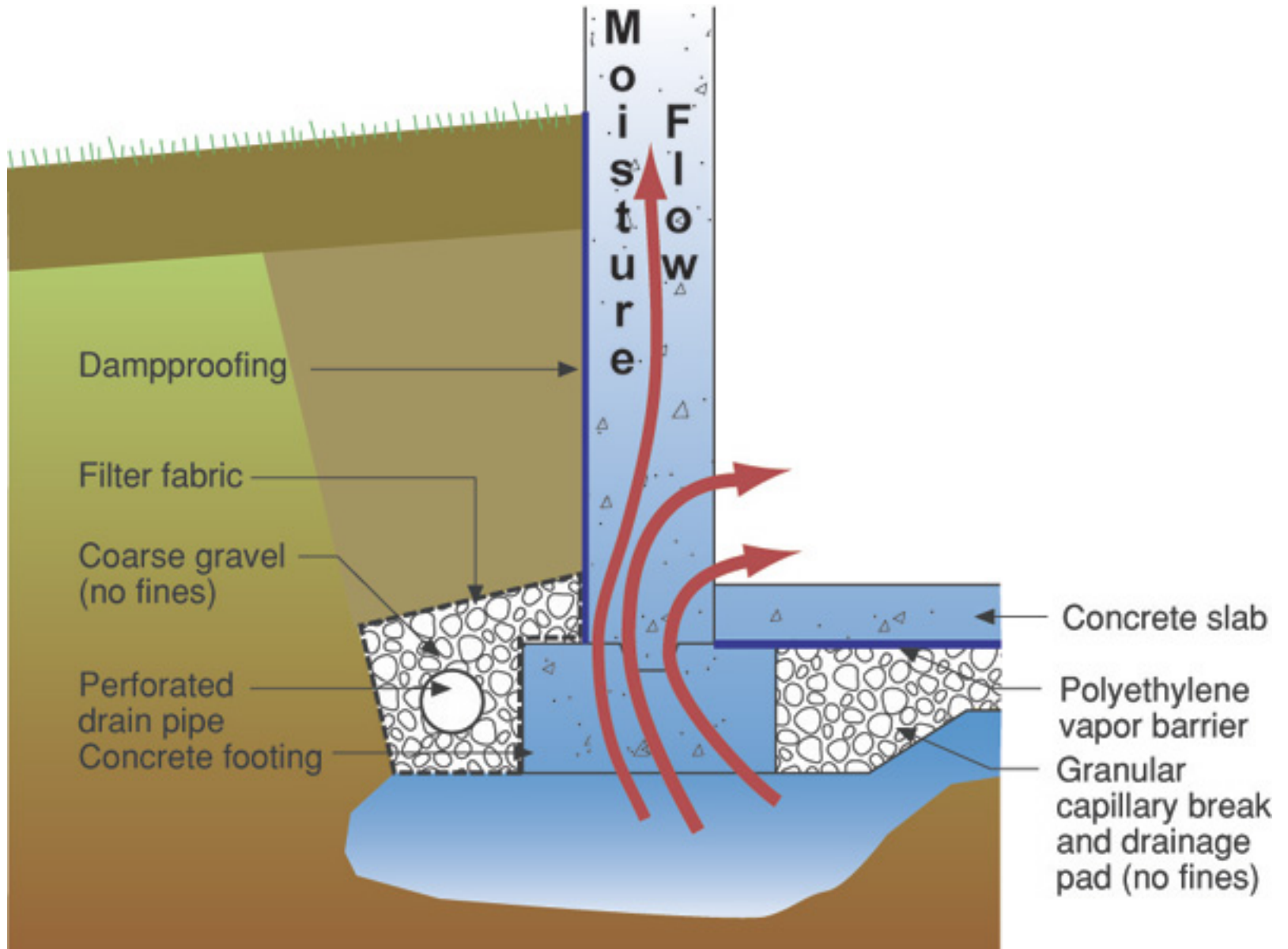
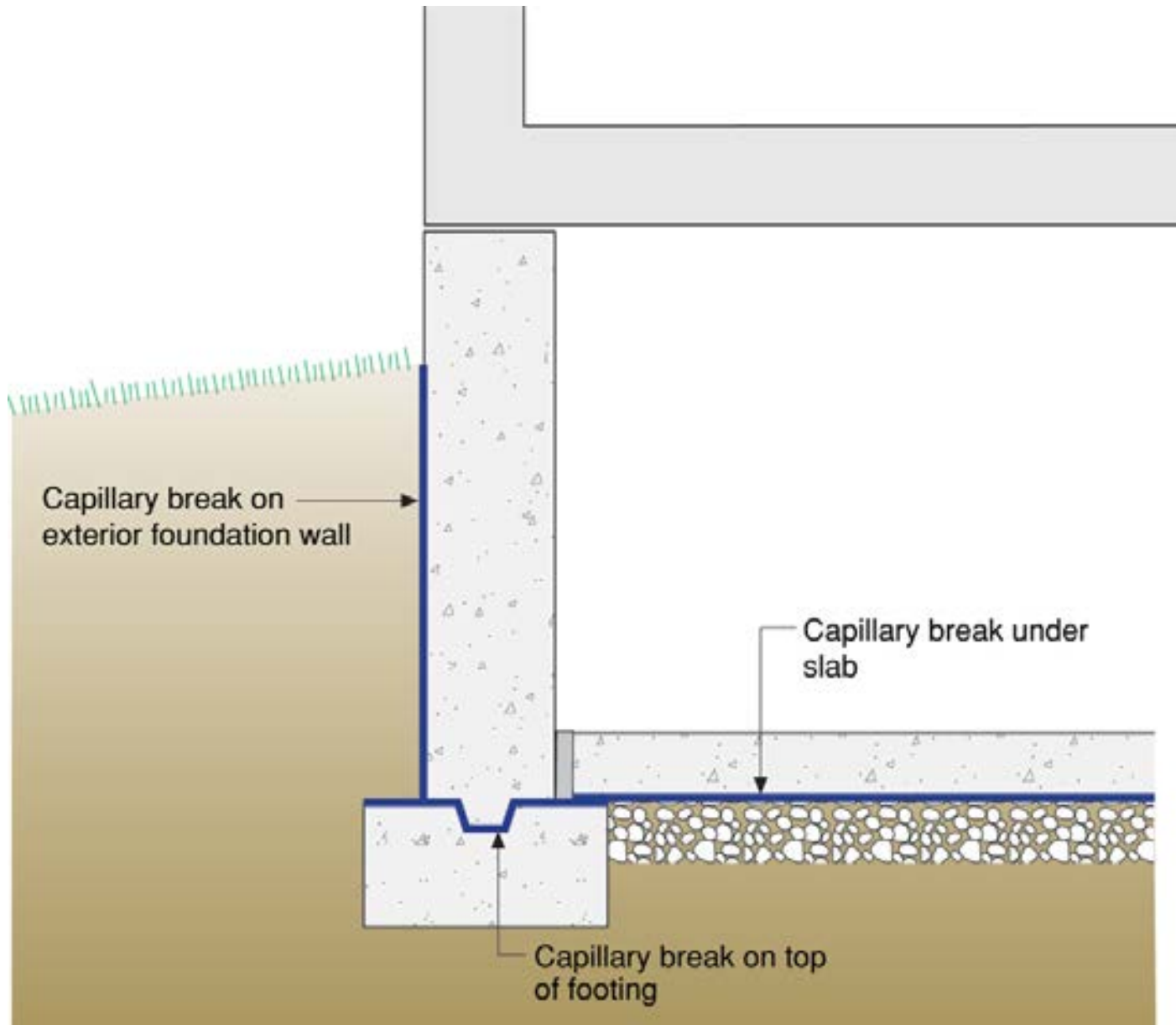


Figure 1b. Brick, sintered clay, porosity 40 per cent.



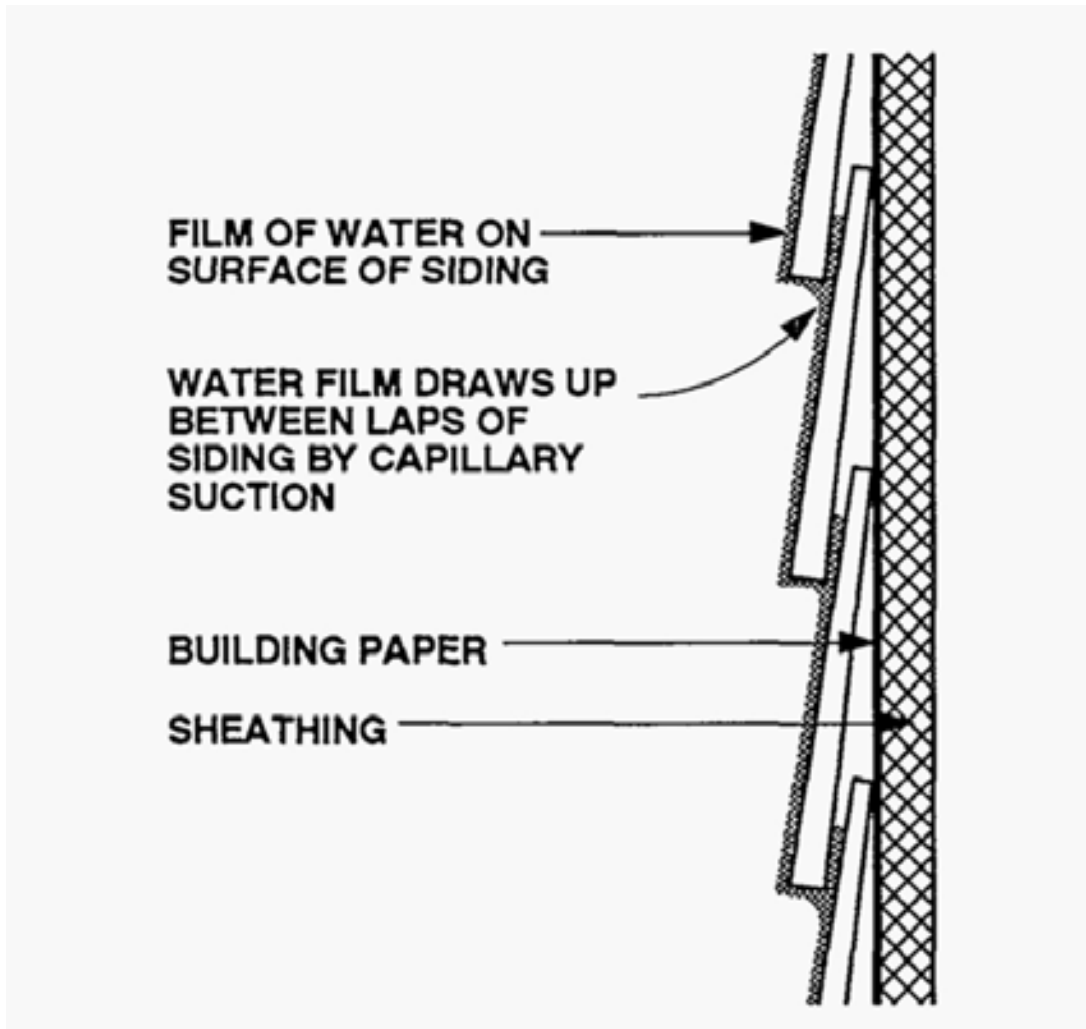




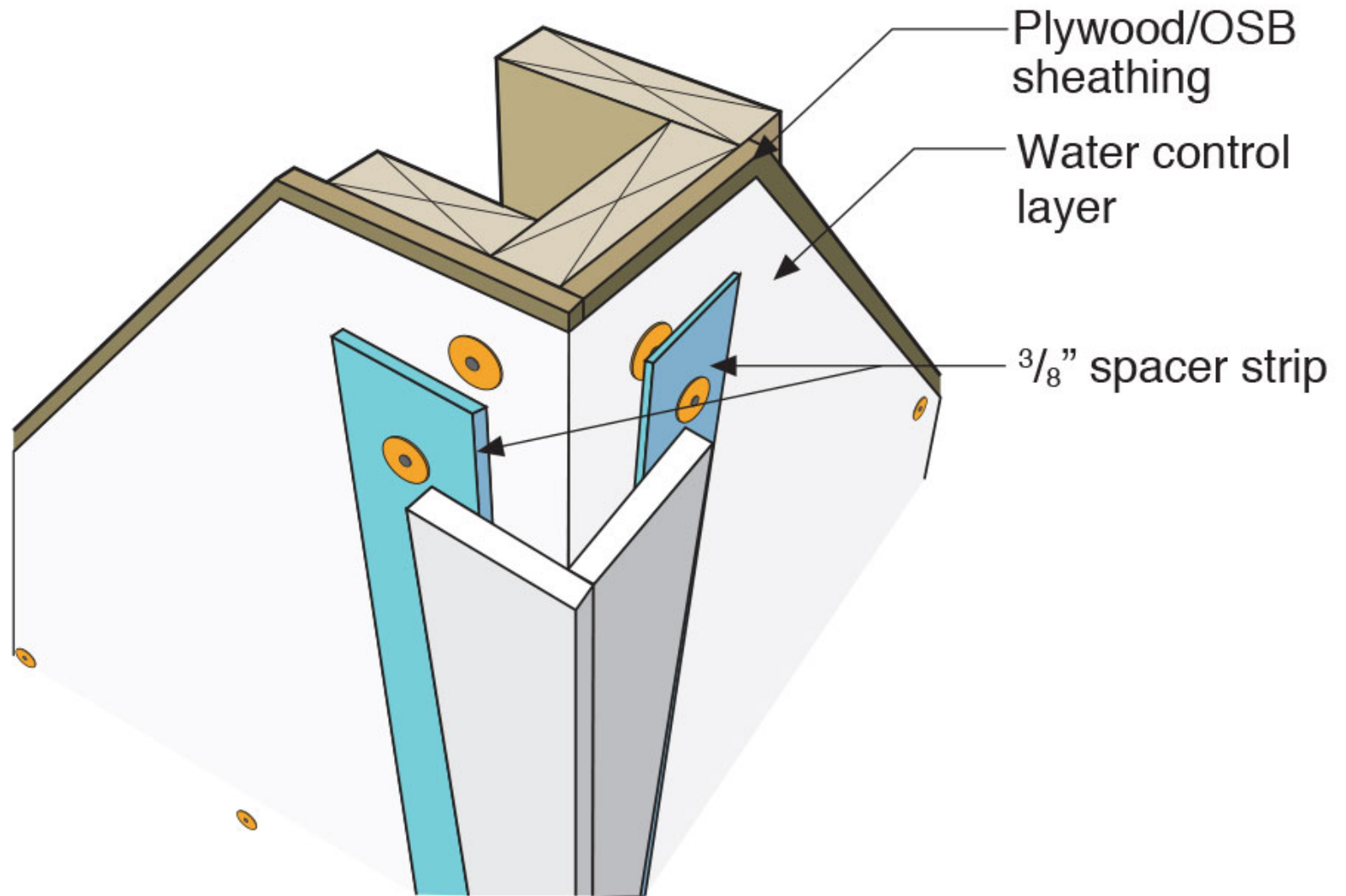


Siding Laps













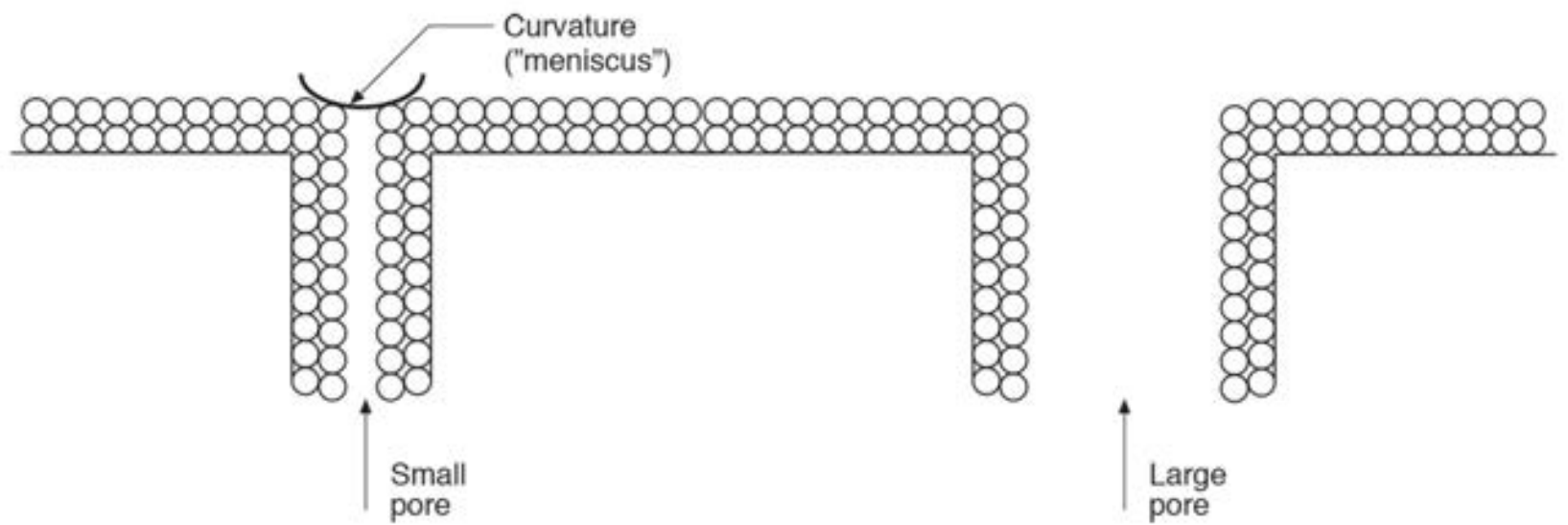






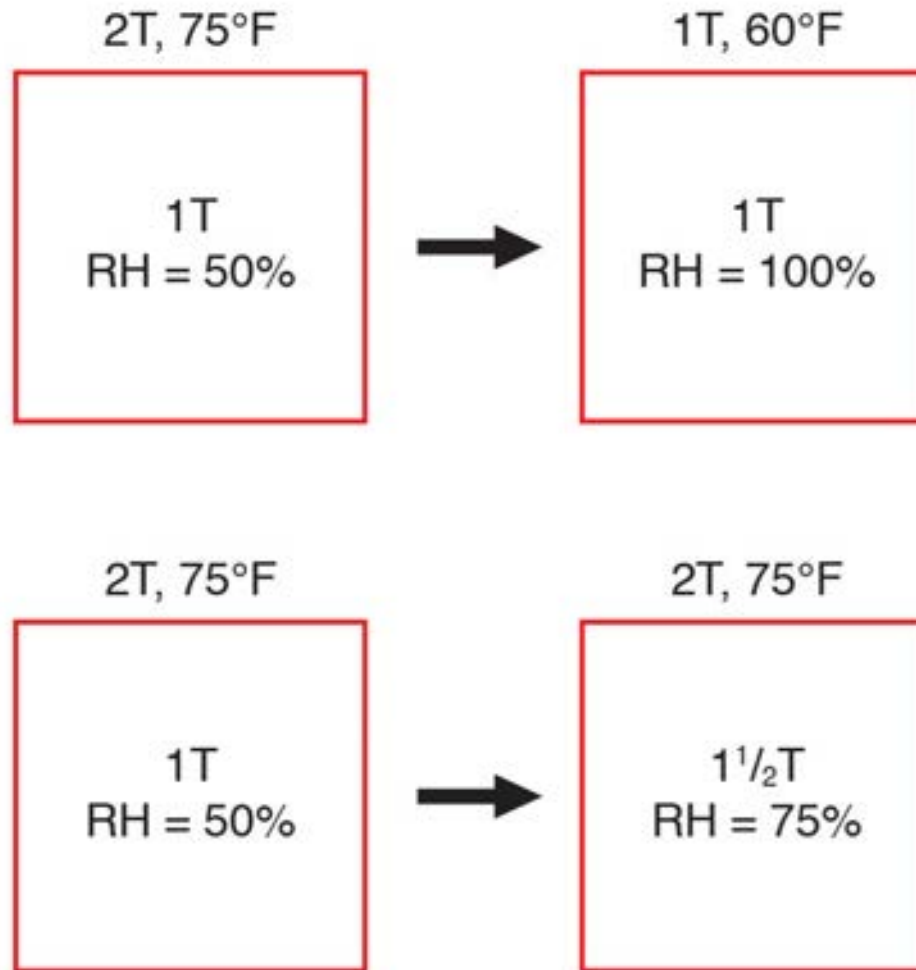
Kelvin Equation Again....

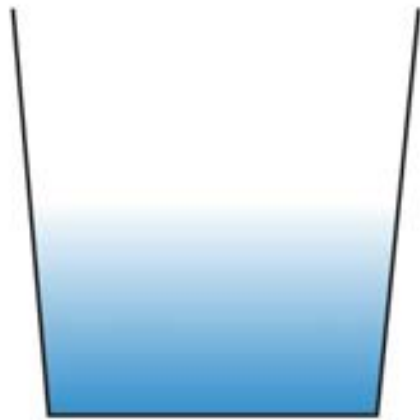
$$\ln \frac{p}{p_0} = \frac{2\gamma V_m}{rRT}$$



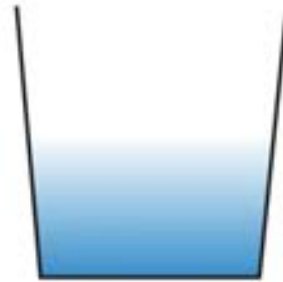


Relative Humidity Vapor Pressure





90°F
50% RH



75°F
50% RH



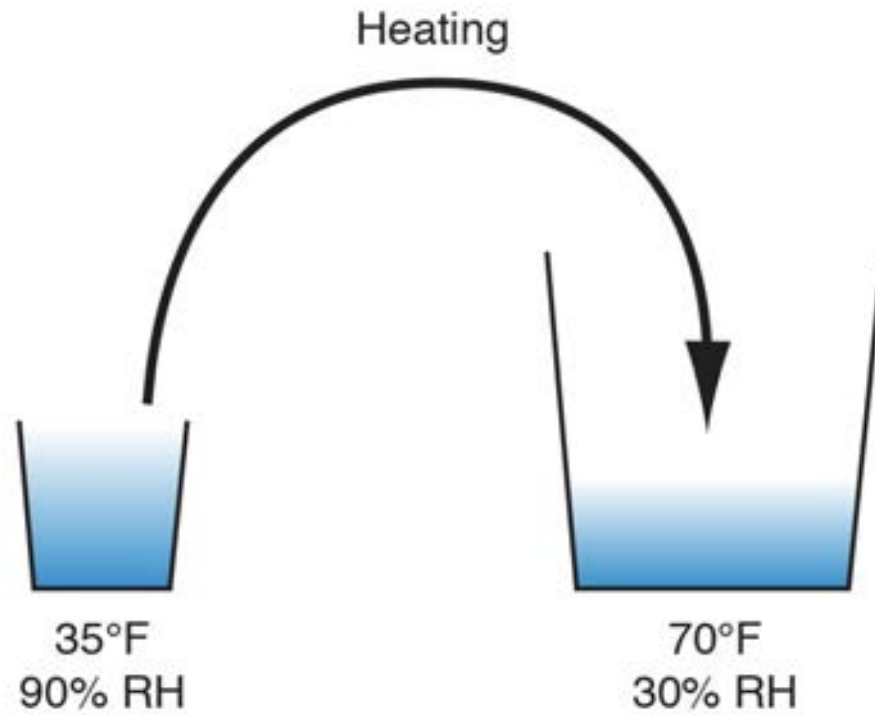
60°F
50% RH

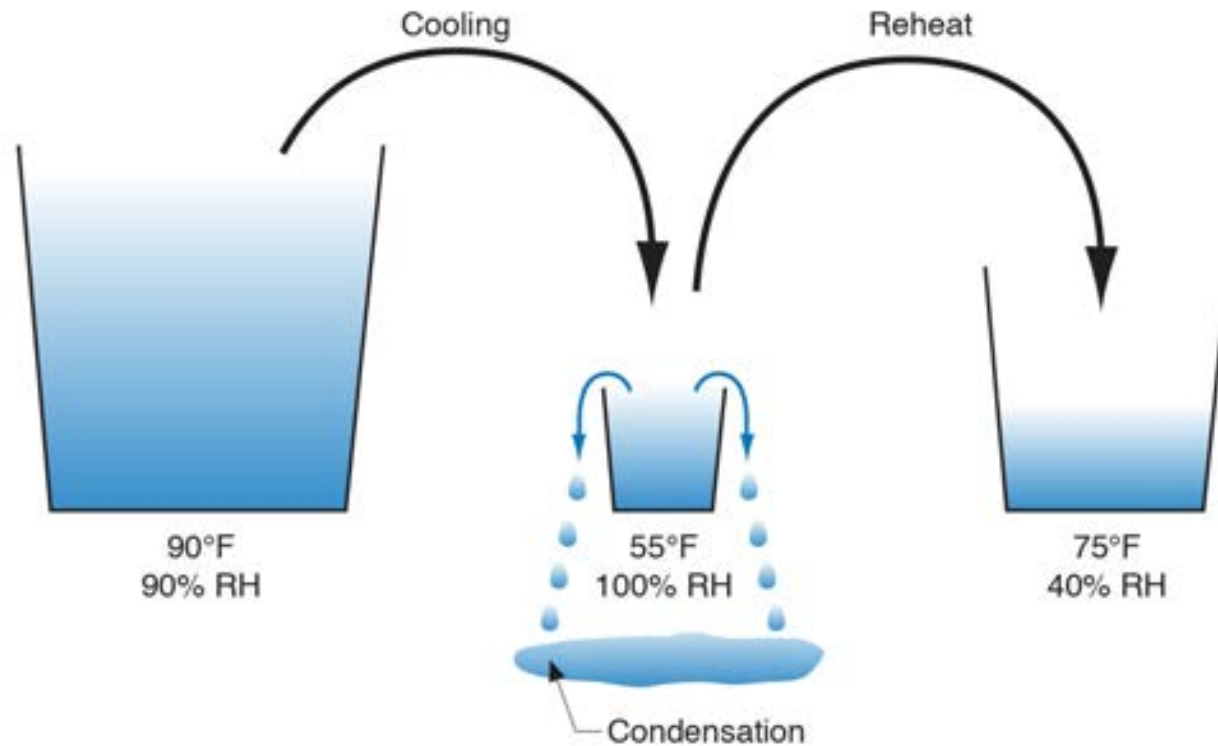


45°F
50% RH

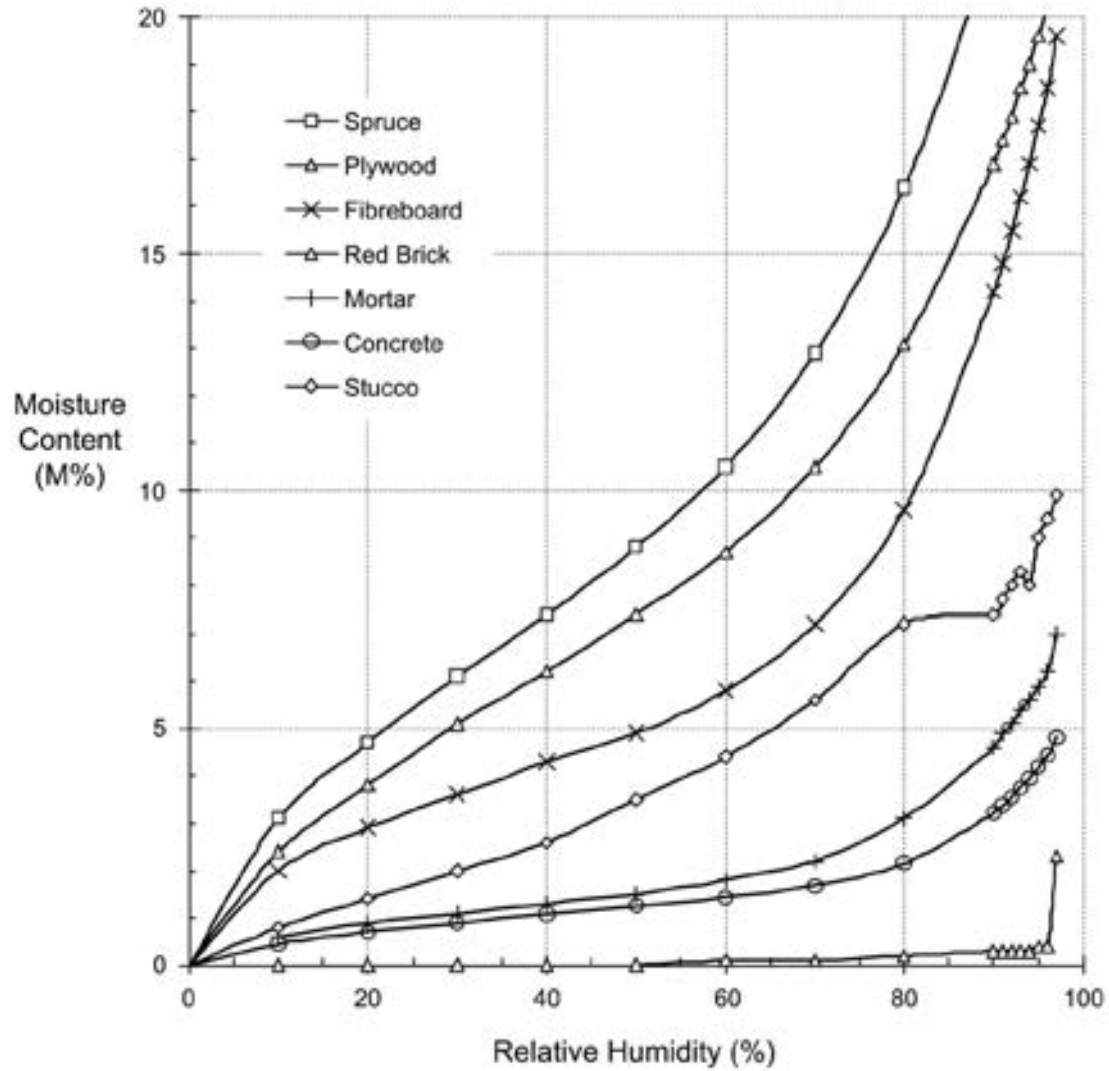


30°F
50% RH





Sorption



Sorption isotherm for several building materials [Kumaran 2002]
 From Straube & Burnett, 2005

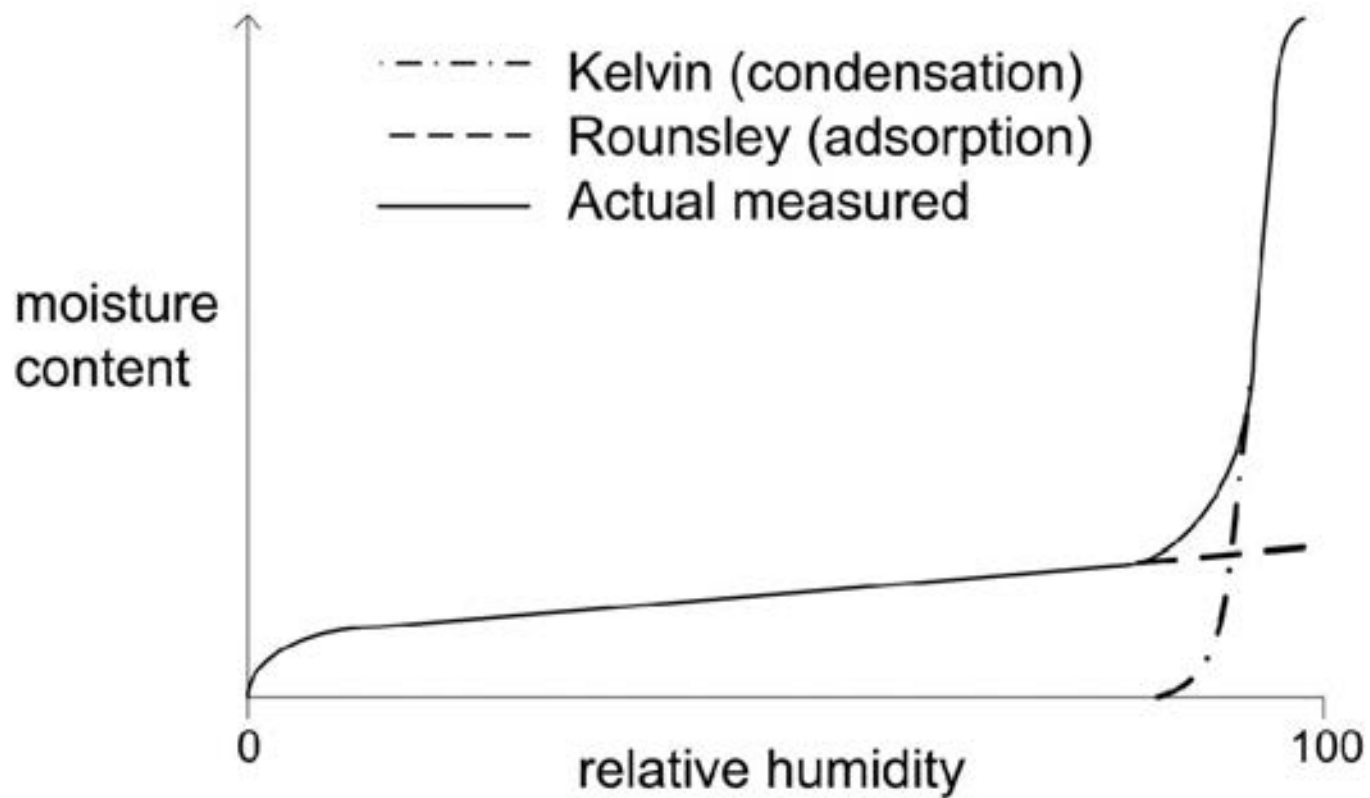
BET Theory

BET Theory

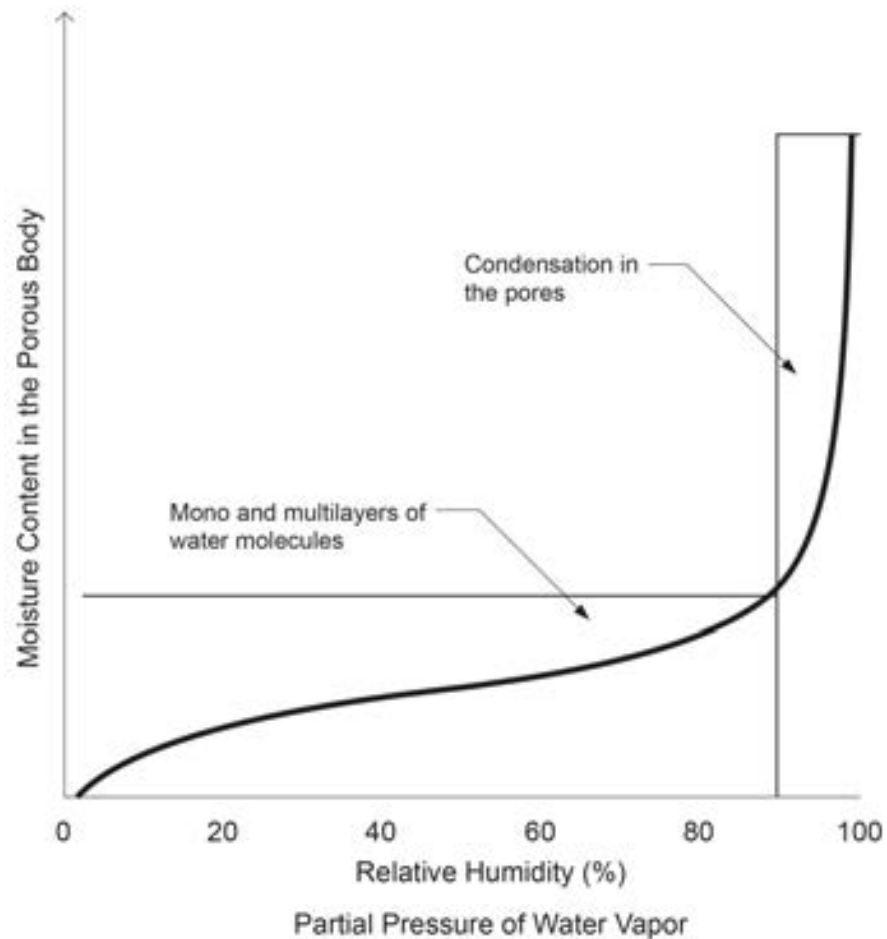
Stephen Brunauer

Paul Emmett

Edward Teller



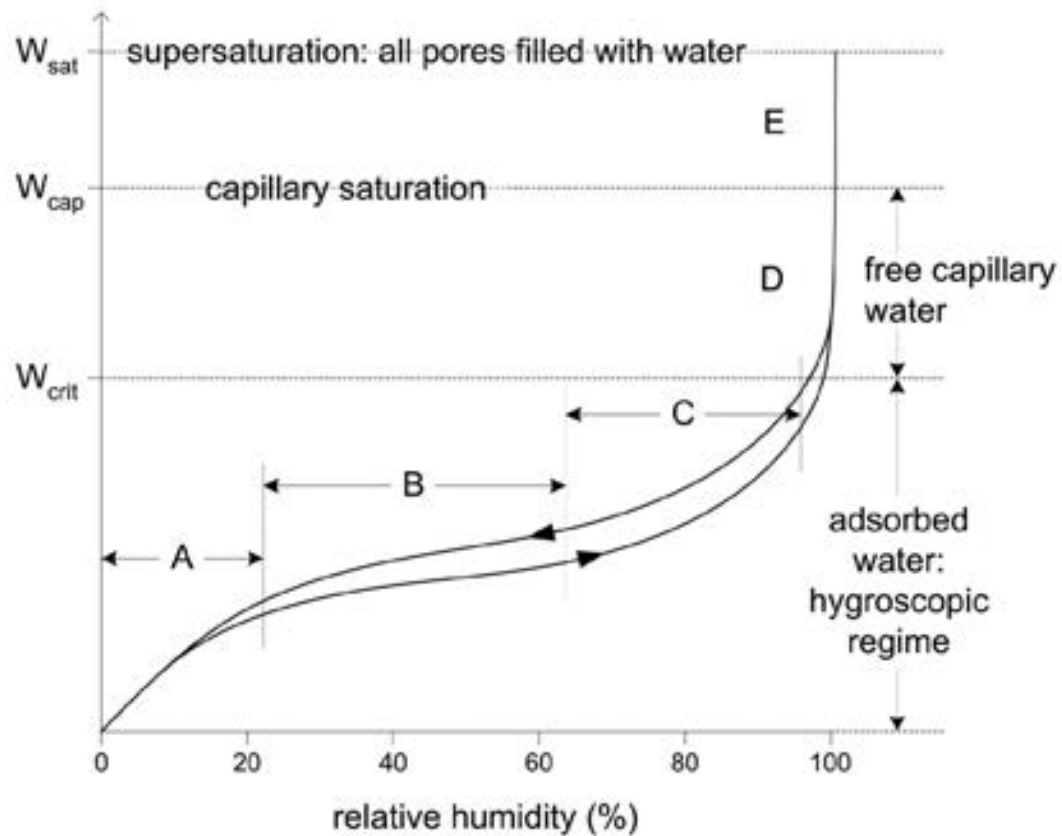
**Typical predicted sorption isotherm according to Kelvin equation
and modified BET theory**
From Straube & Burnett, 2005



Change in the storage of moisture in a porous building material as the partial pressure of water vapor in the ambient air increases from zero to full saturation value at a given temperature.

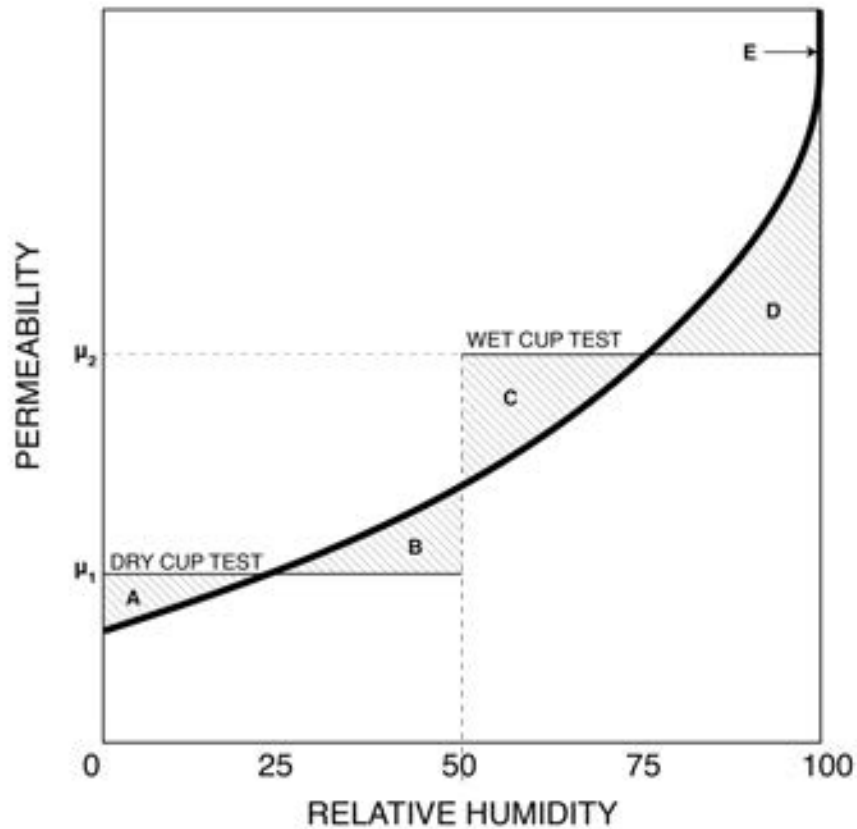
Sorption Curve

From M.K. Kumaran, ASTM MNL 18-2nd Edition,
Moisture Control in Buildings, 2009



- A: Single-layer of adsorbed molecules
- B: Multiple layers of adsorbed molecules
- C: Interconnected layers (internal capillary condensation)
- D: Free water in Pores, capillary suction
- E: Supersaturated Regime

Regimes of moisture storage in a hygroscopic porous material
 From Straube & Burnett, 2005

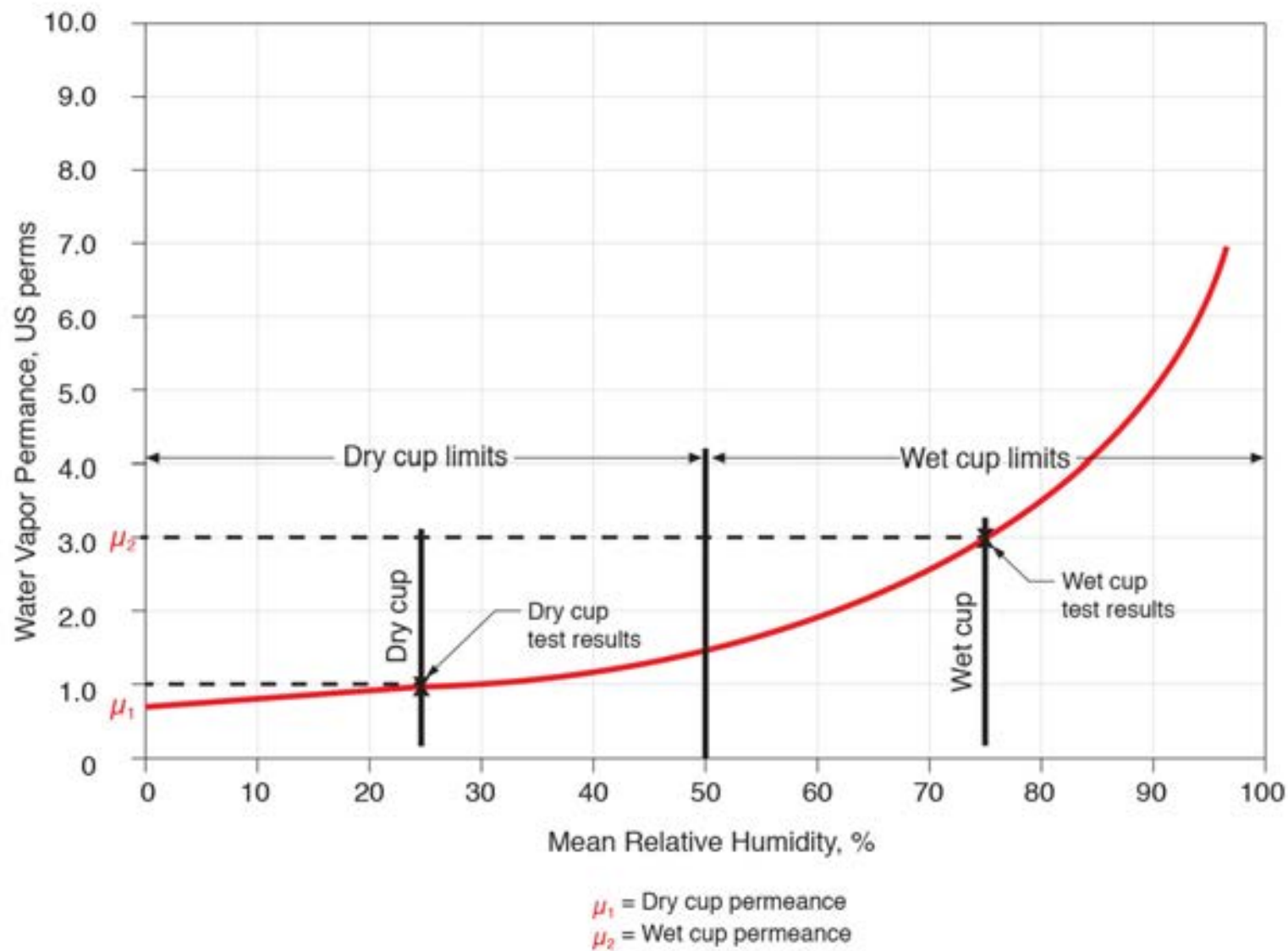


- A - Single-layer of absorbed molecules
- B - Multiple layers of absorbed molecules
- C - Interconnected layers (internal capillary condensation)
- D - Free water in pores, capillary suction
- E - Supersaturated regime

Relationship between Dry Cup and Wet Cup
Adapted from Joy & Wilson, 1963



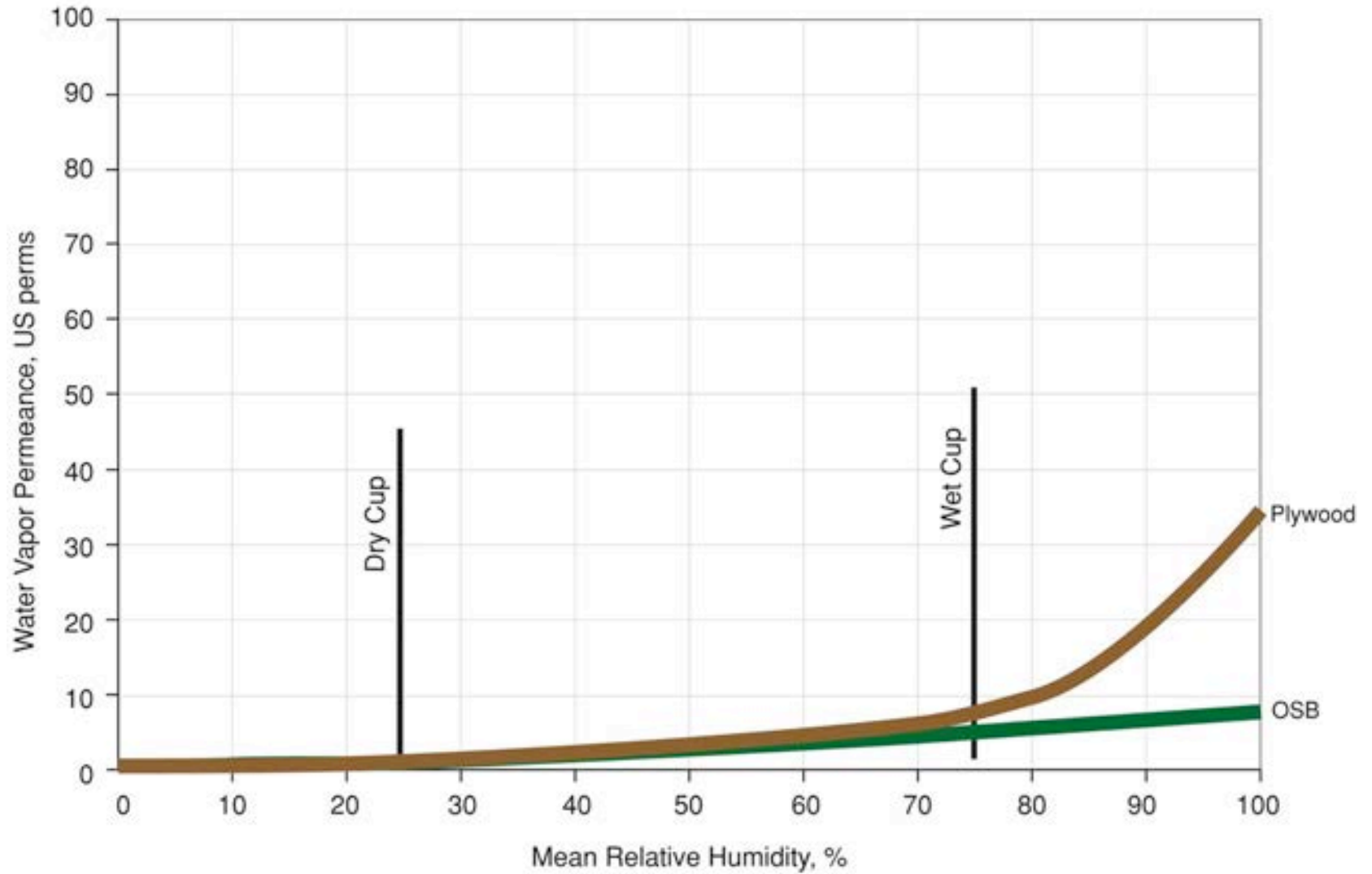
Water Vapor Permeance vs. Relative Humidity



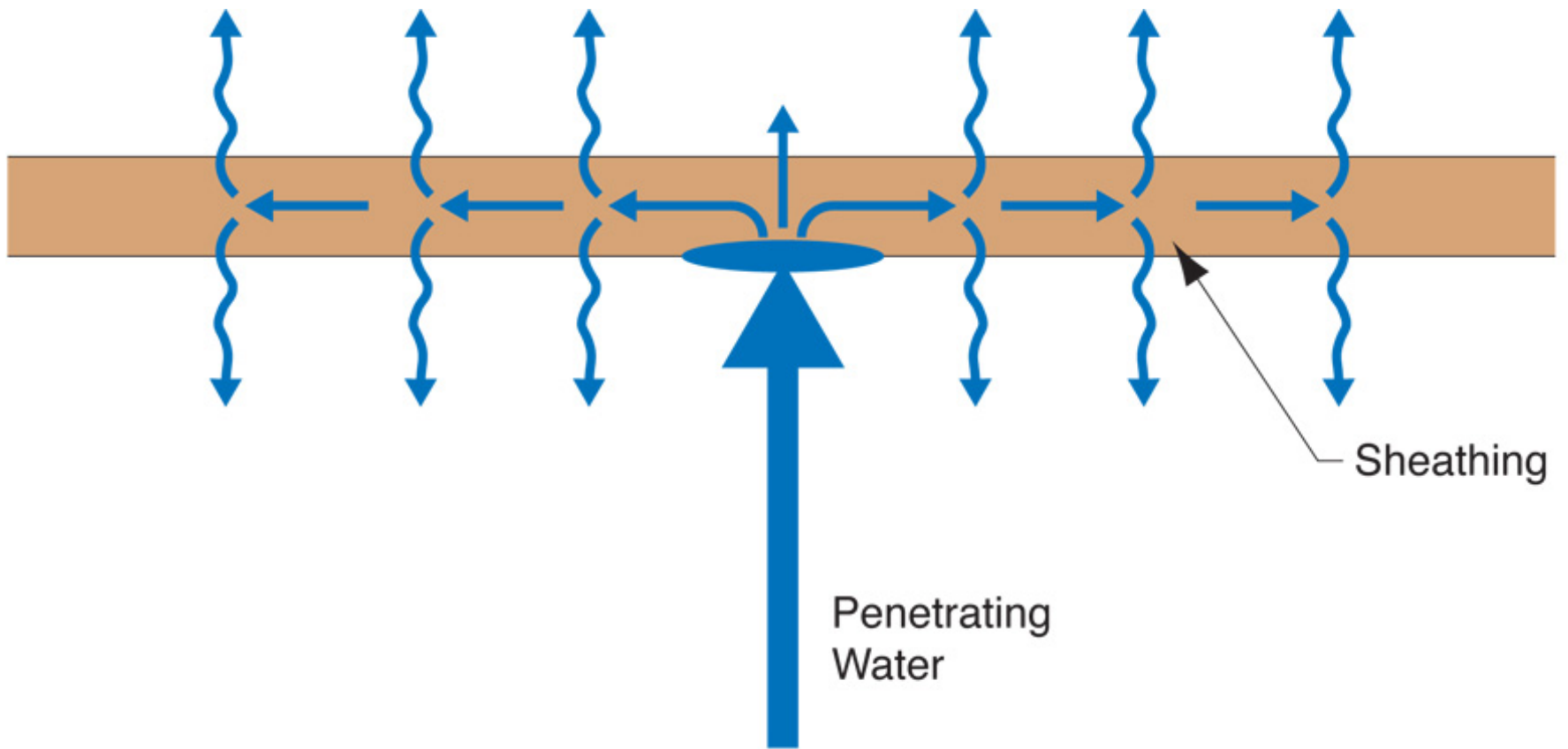


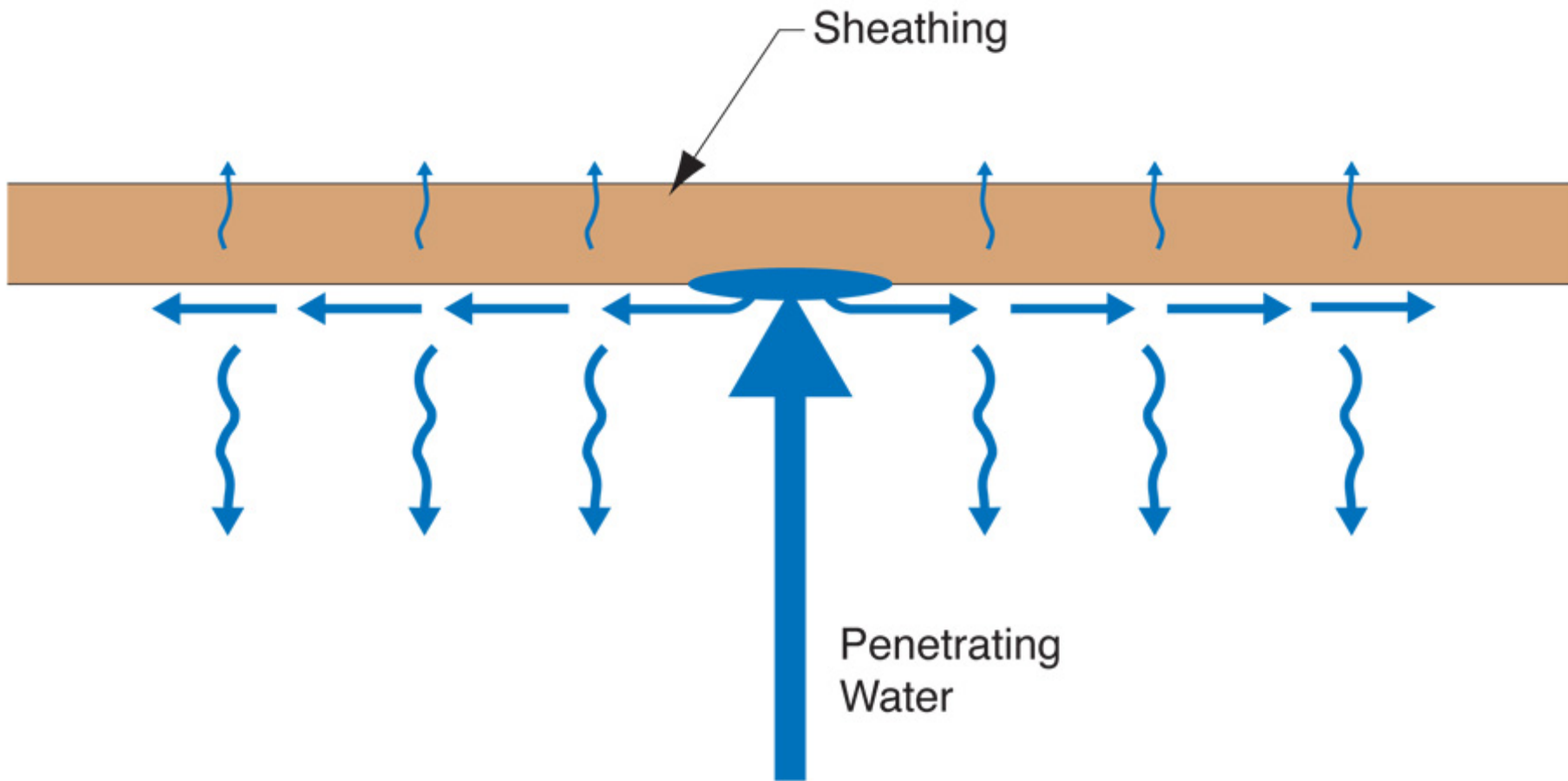


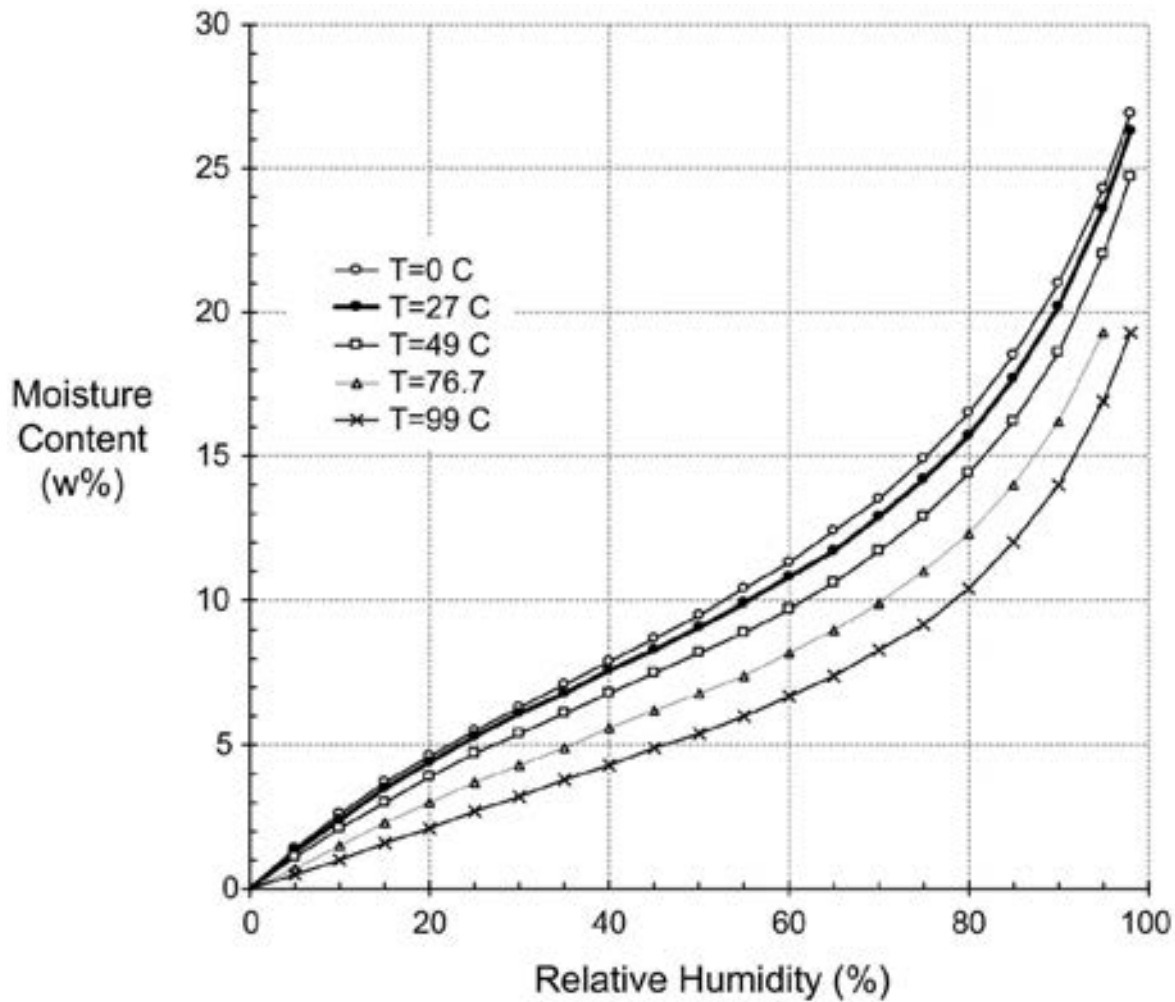
Water Vapor Permeance of Sheathing Materials





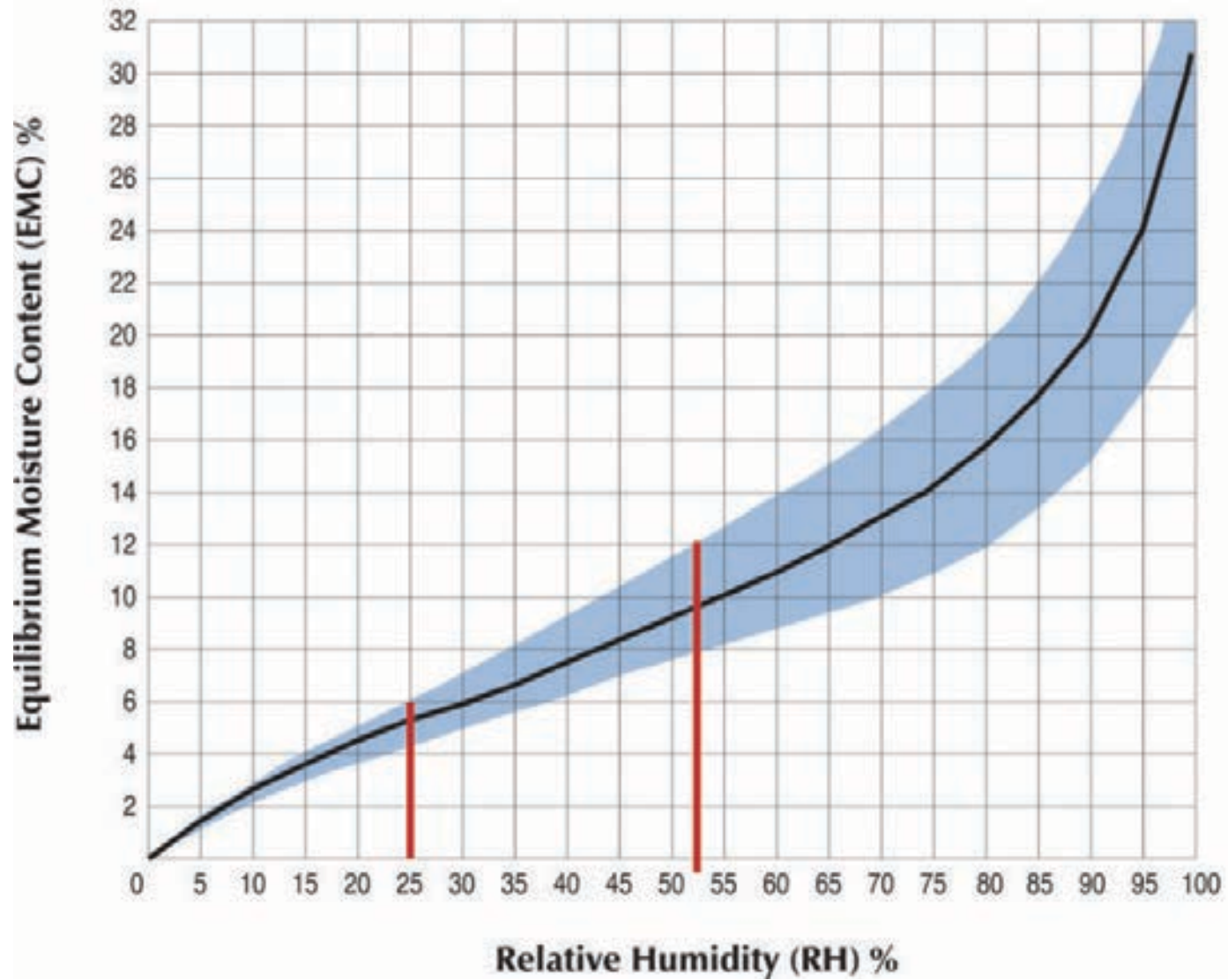






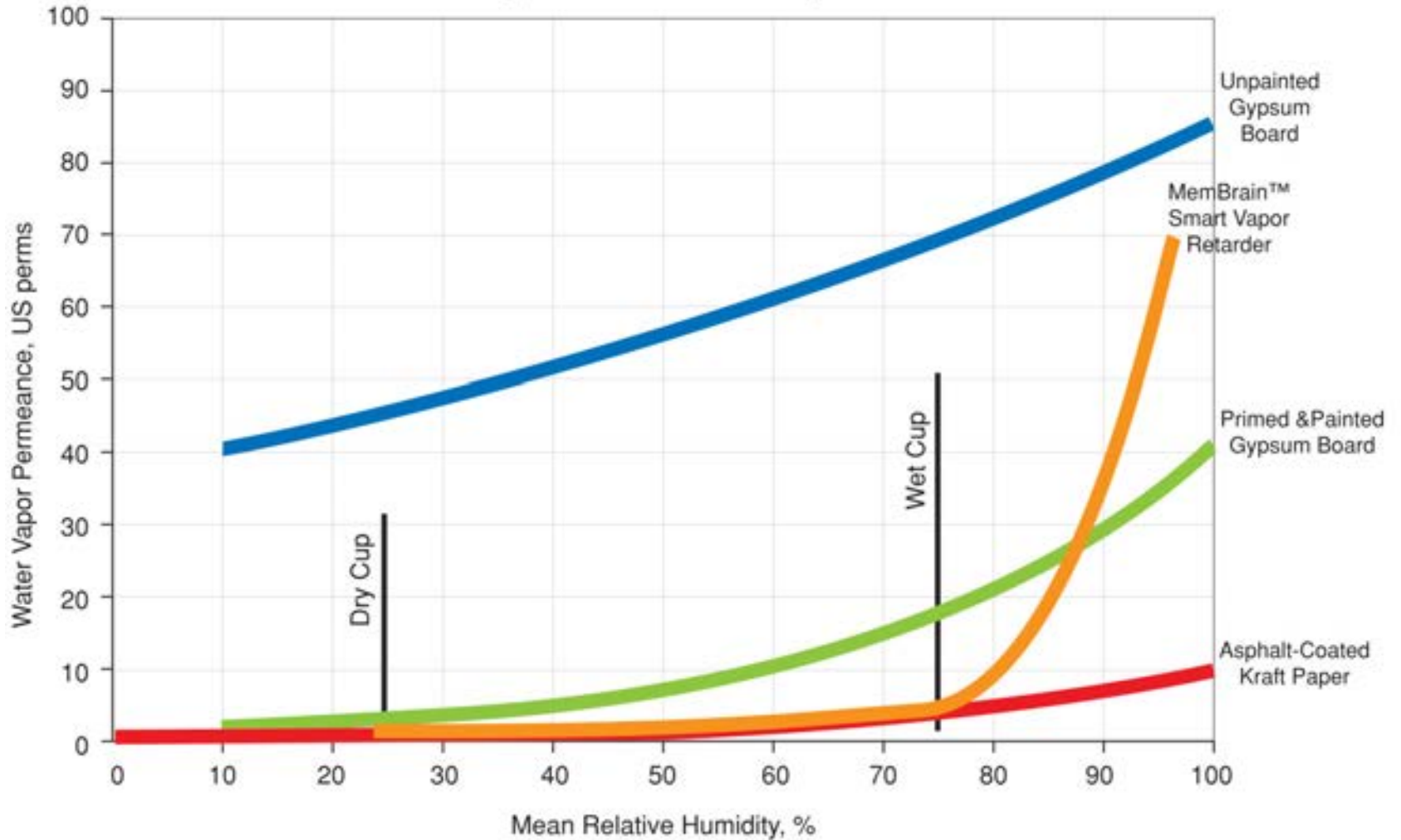
Average sorption isotherm for wood as a function of temperature
 From Straube & Burnett, 2005

Moisture Content vs. Relative Humidity

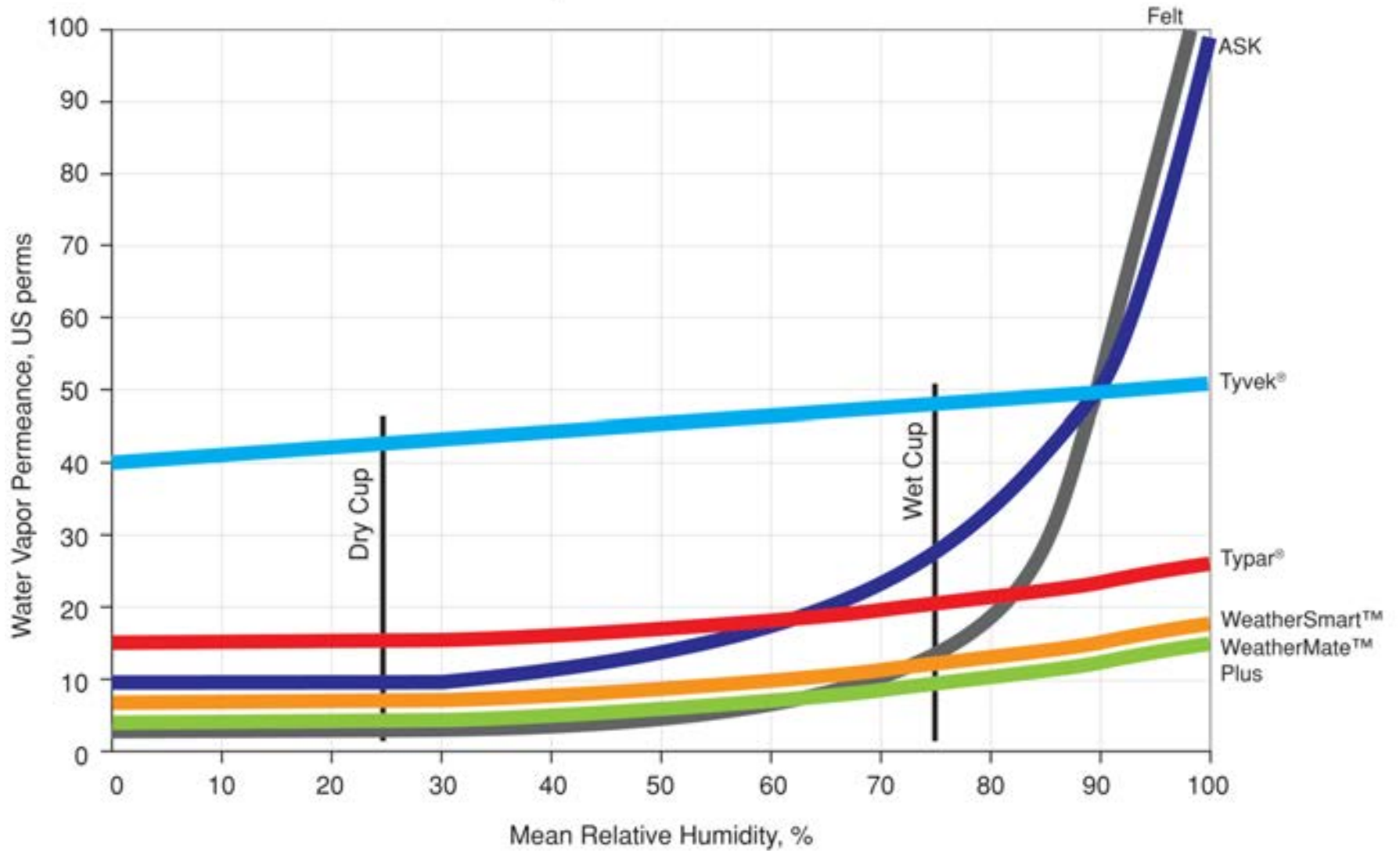




Water Vapor Permeance of MemBrain™ Smart Vapor Retarder, Primed and Painted Gypsum Board, Unpainted Gypsum Board and Asphalt-Coated Kraft Paper



Water Vapor Permeance of WRB's



Laws of Thermodynamics

Zeroth Law – Equal Systems

First Law - Conservation of Energy

Second Law - Entropy

Third Law – Absolute Zero

2nd Law of Thermodynamics

In an isolated system, a process can occur only if it increases the total entropy of the system

Rudolf Clausius

Moisture Flow Is From Warm To Cold
Moisture Flow Is From More To Less

Moisture Flow Is From Warm To Cold
Moisture Flow Is From More To Less

Thermal Gradient – Thermal Diffusion
Concentration Gradient – Molecular Diffusion

Moisture Flow Is From Warm To Cold
Moisture Flow Is From More To Less

Thermal Gradient – Thermal Diffusion
Concentration Gradient – Molecular Diffusion

Vapor Diffusion

Thermodynamic Potential



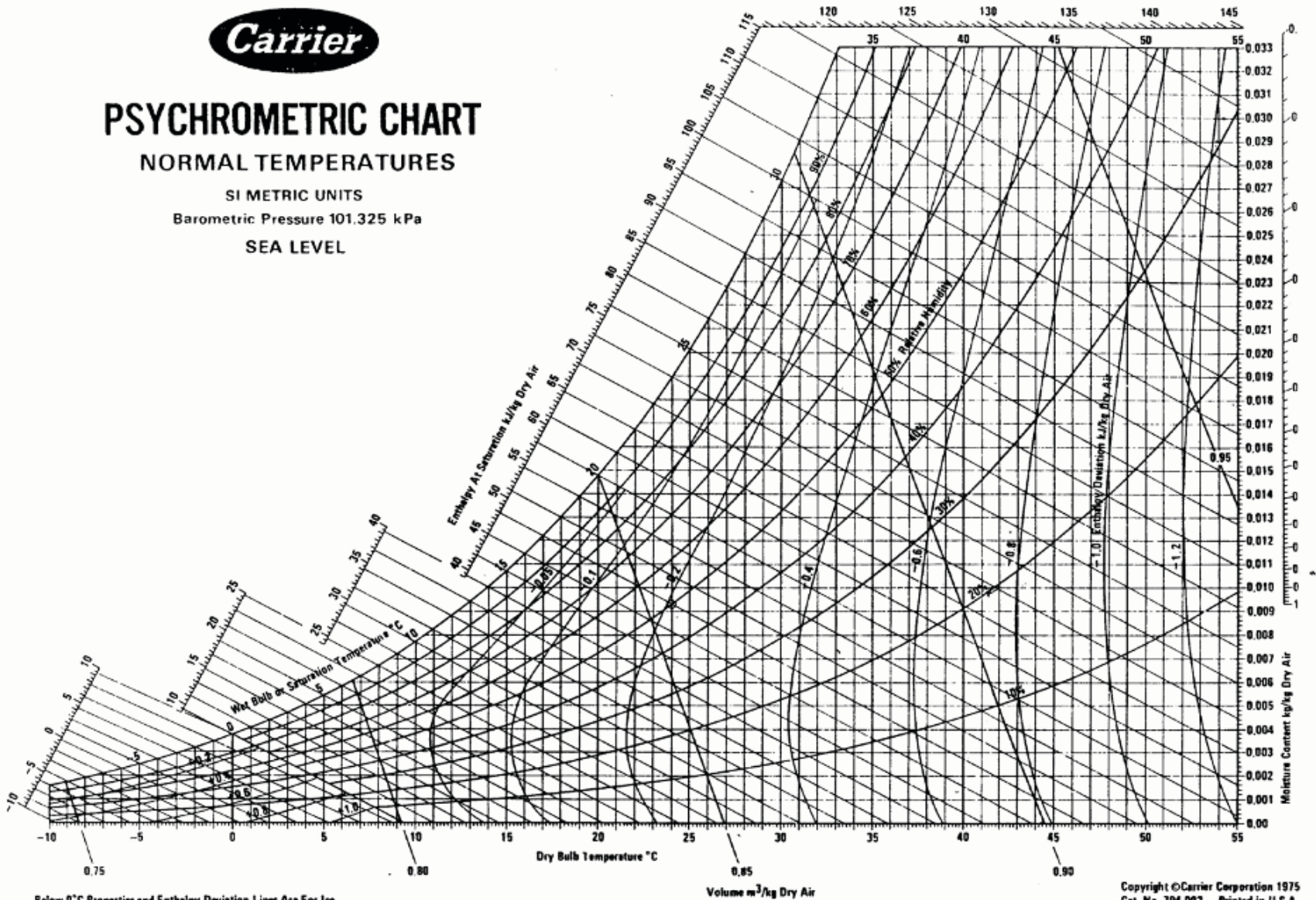
PSYCHROMETRIC CHART

NORMAL TEMPERATURES

SI METRIC UNITS

Barometric Pressure 101.325 kPa

SEA LEVEL



Below 0°C Properties and Enthalpy Deviation Lines Are For Ice

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Vapor

Diffusion

Convective Flow

Vapor Concentration

Air Pressure



**Higher Dewpoint Temperature
Higher Water Vapor Density
or Concentration
(Higher Vapor Pressure)
on Warm Side of Assembly**

**Low Dewpoint Temperature
Lower Water Vapor Density
or Concentration
(Lower Vapor Pressure)
on Cold Side of Assembly**



**Higher Air
Pressure**

**Lower Air
Pressure**

